

UNITED STATES

NUCLEAR WASTE TECHNICAL REVIEW BOARD

Meeting of the Panel on the Repository

Disposal of aluminum-clad, highly enriched, DOE-owned spent fuel; vitrified high-level waste; and immobilized weapons-grade plutonium in a repository

December 17, 1997

Radisson Riverfront Hotel Augusta  
Augusta, Georgia

NWTRB BOARD MEMBERS PRESENT

Dr. Daniel Bullen, Meeting Chair  
Dr. John Arendt  
Dr. Paul Craig  
Dr. Alberto Sagnas  
Dr. Jeffrey Wong

NWTRB STAFF

Dr. Carl Di Bella, Senior Professional Staff  
Ms. Linda Hiatt, Management Assistant

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**Introductory Remarks**

Daniel Bullen, Meeting Chair. . . . . 3



1           We have a very full agenda today, so I think we  
2 need to get started on time. My name is Dan Bullen, and I'm  
3 a member of the U. S. Nuclear Waste Technical Review Board,  
4 and I will be chairing today's meeting.

5           It's actually been several years since the Board  
6 paid a visit to Savannah River, so I'd like to take a few  
7 minutes outlining what the Board is, and after that, I'll  
8 make a few introductions and give a little background and the  
9 ground rules for today's meeting.

10           In 1982, Congress enacted the Nuclear Waste Policy  
11 Act. The law created the Office of Civilian Radioactive  
12 Waste Management, or OCRWM, within the Department of Energy,  
13 and charged OCRWM to develop repositories for final disposal  
14 of the nation's spent nuclear fuel and high-level radioactive  
15 waste.

16           Five years later, Congress amended the 1982 Act to  
17 limit OCRWM to characterizing a single site for final  
18 disposal, that being the Yucca Mountain site, which is a site  
19 300 meters below Yucca Mountain on the western edge of the  
20 Nevada Test Site.

21           In the same 1987 amendment, Congress created the U.  
22 S. Nuclear Waste Technical Review Board as an independent  
23 agency to review the technical validity of OCRWM's program,  
24 and to periodically furnish the Board's findings, conclusions  
25 and recommendations to the Secretary of Energy, to Congress,

1 and to the public.

2           The U. S. President appoints our board members from  
3 a list of nominees submitted by the National Academy of  
4 Sciences. Each nominee shall be eminent in his or her field  
5 of science or engineering and shall be selected solely on his  
6 established record of distinguished service. A full board  
7 consists of eleven members.

8           As you see by the overhead here, our members come  
9 from a wide range of organizations and institutions, and I'll  
10 go through their backgrounds briefly. As I do so, I will ask  
11 the ones present today to either stand up or raise their  
12 hand.

13           Our Chair is Jared Cohon. He's the President of  
14 Carnegie Mellon University, and holds expertise in  
15 environmental systems analysis and hydrology. He is a  
16 Registered Civil Engineer.

17           John Arendt--John, would you recognize yourself--  
18 began his career in the Manhattan project in 1945, and works  
19 in nuclear materials transportation and nuclear materials  
20 facilities, their quality assurance, quality control, and  
21 inspection. He is a chemical engineer, and as I noted, he is  
22 here with us today.

23           My name is Dan Bullen, and I coordinate the Nuclear  
24 Engineering Program within the Department of Mechanical  
25 Engineering at Iowa State University. My expertise is in

1 nuclear waste management, performance assessment modelling,  
2 and material science. I'm a nuclear engineer by training.

3           Norm Christensen, who is not here today, is the  
4 Dean of the Nicholas School of the Environment at Duke  
5 University, and is an expert in biology and the ecology.

6           Paul Craig is Professor of Engineering Emeritus at  
7 U.C. Davis, with expertise in energy policy issues associated  
8 with global environmental change.

9           Debra Knopman, who is not here today, directs the  
10 Center for Innovation in the Environment at the Progressive  
11 Foundation in Washington, D.C., and has expertise in  
12 hydrology, environmental and natural resource policy, and  
13 systems analysis.

14           Priscilla Nelson, who actually chairs the  
15 Repository Panel which is hosting this panel meeting, is also  
16 not with us today. I'm acting on her behalf as chair of this  
17 meeting. She is the Program Director of the Directorate for  
18 Engineering at the National Science Foundation. Her  
19 expertise is actually in rock engineering and underground  
20 construction.

21           Dick Parizek is Professor of Geology at Penn State  
22 and specializes in hydrogeology and environmental geology.

23           Alberto Sagηgs is Professor of Materials  
24 Engineering within the Civil Engineering Department at the  
25 University of South Florida. His expertise lies in corrosion

1 and materials engineering, physical metallurgy and scientific  
2 instrumentation.

3           Jeff Wong is a toxicologist at Cal. EPA,  
4 specializing in risk assessment and scientific team  
5 management.

6           The Board is supported by ten professional staff  
7 and six clerical staff. Most of the staff are full-time  
8 federal employees, but some are part-time. Except for  
9 Priscilla Nelson, who is a full-time federal employee, Board  
10 members are part-time federal employees.

11           Also sitting at the front table today is Dr. Carl  
12 Di Bella. Carl, would you acknowledge yourself here? Carl  
13 is a chemical engineer. He is one of the members of the  
14 Board's professional staff, and has done a lot of the behind  
15 the scenes work necessary for today's technical program.

16           Dr. Dan Metlay also did a great deal of work, and  
17 unfortunately will not be able to join us today. He's  
18 another staff member from our Washington office.

19           I would like to acknowledge and thank another staff  
20 member who does a Herculean effort, Linda Hiatt in the back  
21 of the room. Would you raise your hand, please, Linda?  
22 She's responsible for putting together the logistics of  
23 today's meeting, and more importantly, she would like  
24 everyone to sign in so she can make you a name tag. Is that  
25 not correct?

1           One of the very important ways our Board goes about  
2 obtaining information needed to accomplish our mission is by  
3 holding public information gathering meetings. We hold three  
4 full Board meetings and some five to ten panel meetings  
5 yearly, and this is one of those panel meetings.

6           The full Board meetings are generally two days  
7 long, covering a variety of issues, and are attended by the  
8 entire board. Panel meetings are shorter, more focused, and  
9 are attended by Board members with specific interests in one  
10 or more of the topics to be covered. As a result of these  
11 meetings and our access to other relevant information, the  
12 Board makes assessments and recommendations about how to  
13 improve the technical and scientific aspects of the Waste  
14 Management Program.

15           A bit more information about the Board may actually  
16 be found on our web site, and this is a commercial for that  
17 site, the HTML, or the address for our web site is  
18 [www.nwtrb.gov](http://www.nwtrb.gov)., and all of our reports are summarized and all  
19 of our calendar is there, so you can actually look at our  
20 calendar to see upcoming Board meetings.

21           Today's meeting is a meeting of the Board's panel  
22 on the repository. The purview of the panel includes  
23 everything that would be engineered at Yucca Mountain,  
24 including the surface and underground aspects of the  
25 repository, the waste packages that would go into the

1 repository, and the waste form that the packages would  
2 contain.

3           There are a number of wastes at Savannah River that  
4 are destined for disposal at Yucca Mountain should the site  
5 be found suitable for the development as a repository. These  
6 wastes includes spent foreign research reactor fuel and  
7 vitrified high-level waste, among other things. We  
8 understand that the process, development and manufacture of  
9 vitrified surplus weapons plutonium will also be done at  
10 Savannah River. This waste again will be destined for  
11 disposal at the repository.

12           Now, if there's one single triggering reason for  
13 the meeting, it's the aluminum-clad, highly enriched uranium  
14 spent fuel, most if not all of which comes from research  
15 reactors. I'm the facility director for the Iowa State  
16 University research reactor, which has aluminum-clad spent  
17 fuel, and when I first saw the designs that were proposed by  
18 the OCRWM program for the direct disposal of spent fuel, I  
19 had three reactions, which were all caused by a combination  
20 of the material's high enrichment, coupled with its limited  
21 engineering stability of the cladding for long-term disposal.

22           The first is that the material clearly presents a  
23 heightened long-term criticality control issue within the  
24 repository. The second is that the material probably has a  
25 relatively high monetary value. It wasn't cheap to make

1 highly enriched uranium. And, finally, both the monetary  
2 value and the potential utility for nuclear explosives of  
3 this material would make it attractive for future generations  
4 to retrieve, with motivations that are not necessarily  
5 coincident with those of the United States. These  
6 essentially could compromise, or these issues could  
7 compromise the performance of the entire repository long-  
8 term.

9           Now, this morning, we will focus on DOE-owned spent  
10 fuel, with a particular focus on the aluminum-clad, highly  
11 enriched spent fuel. This afternoon, we will move on to some  
12 of the other wastes that will be disposed of in the  
13 repository, specifically vitrified high-level waste from  
14 reprocessing and vitrified surplus weapons plutonium.

15           Now, just a brief background on the ground rules  
16 for today. I want to alert the speakers that brief questions  
17 from the front table, the panel members and staff, will be  
18 allowed during the course of the presentation. This is a  
19 little bit different than we have in our formal board  
20 meetings.

21           We keep it a little bit more informal to have  
22 interaction between the board or panel and the speakers.  
23 However, there are either 10, 15 or 20 minute time periods at  
24 the end that will be saved for essentially longer questions,  
25 and if time permits and we've exhausted questions from the

1 front table, we would be happy to take questions from the  
2 audience during that time period. So please identify  
3 yourself and we'll recognize you for questions.

4           Now, I also want to point out that there is a  
5 public comment period at the end of the day, and Linda Hiatt,  
6 who we introduced in the back, has a special sign-up sheet  
7 for public commenters. Please sign up with her if you wish  
8 to make a statement or ask a question at the end of the day.  
9 We ask that the statements be limited to five minutes or  
10 less, and we will try to take the statements in order of  
11 sign-up on the sheet.

12           In all cases throughout the day, people asking the  
13 questions or making statements should use a microphone and  
14 should identify themselves, and that includes Board members.

15           I also want to say a few words about the Board's  
16 positions and the Board's pronouncement and member  
17 statements. The NWTRB, this Board, has a very important  
18 role. What it says is taken seriously by policy makers and  
19 the members of the public. The Board generally conveys its  
20 findings, conclusions and recommendations in written form in  
21 the form of formal reports, letters to Congress and/or the  
22 Secretary of Energy, or to the Director of OCRWM, and in  
23 written Congressional testimony.

24           Of course, the Board consists of several  
25 individuals, each of which has his or her own style. Each is

1 free to say whatever they choose. But comments by individual  
2 Board members, including me, are just that, Board member  
3 comments.

4           On occasion, one of us, especially Chairman Cohon,  
5 will make statements on behalf of the Board that that person  
6 explicitly states to be Board positions. Otherwise, when we  
7 make individual comments, they are no more than that.  
8 Whether comments by Board members eventually become Board  
9 position, only time will tell. But, of course, a Board  
10 member's thinking is relevant. In effect, at these meetings  
11 when we make statements and ask questions, we are thinking  
12 out loud as a Board. Our thoughts and comments do not  
13 represent Board positions unless we indicate so, but they may  
14 be on their way to becoming positions. And if they do, we  
15 will convey them in writing.

16           Now it's time to get on with the meeting. Our  
17 first speaker is Howard Eckert, who is going to introduce the  
18 topic of DOE spent fuel. Before he starts, however, I want  
19 to particularly thank him for his efforts in putting together  
20 the technical program on the meeting today. It's not been a  
21 simple undertaking because the speakers in today's meeting  
22 come from many parts of DOE, and the organization is very  
23 diverse and it's a real difficult task to pull those  
24 together.

25           So without further ado, I will give you Howard

1 Eckert, and he will give us an overview.

2           ECKERT: Good morning. As Dan said, I'm going to  
3 briefly give you an overview of DOE-owned spent fuel, and  
4 this includes actually commercial fuel, not just the fuel  
5 produced by DOE's production and research reactors.

6           What I'll cover today is our current inventory, how  
7 much we have, where it is, what fuel will be coming into our  
8 inventory from operating reactors--we're still generating  
9 fuel today--some characteristics of the fuel, how we can  
10 distinguish or describe different types of fuel, the  
11 Department's strategy for dealing with the fuel, storage,  
12 transportation, eventual disposal, and then some ongoing  
13 efforts between the Office of Environmental Management and  
14 the Office of Civilian Radioactive Waste Management,  
15 cooperative efforts to move the fuel along the path toward  
16 disposal.

17           We use three metrics to quantify our fuel. Metric  
18 tons heavy metal is the traditional metric incorporated in  
19 some of our Congressional legislation. Of more relevance to  
20 storage would be the volume of fuel we have to deal with,  
21 cubic meters. For handling, the number of assemblies, or  
22 piece count terminology, is more relevant.

23           Most of our fuel, and we have roughly 2,500 metric  
24 tons heavy metal, or 1,200 cubic meters, most of this is  
25 stored at three sites, Hanford, Idaho and Savannah River.

1 Depending upon which metric one uses, one could say that  
2 Hanford has most of the fuel if the use the metric tons heavy  
3 metal, at 86 per cent. If you talk about volume, then Idaho  
4 has almost half of it, depending upon which argument you're  
5 trying to make.

6           This is the current inventory as of this month. It  
7 does not include fuel that we've made a decision to process.  
8 We are processing or reprocessing fuel at Savannah River.

9           There's some additional fuel at West Valley,  
10 commercial fuel, that's in storage, a small amount at Oak  
11 Ridge in the high flux isotope reactor spent fuel storage  
12 pool, and we've taken ownership of the fuel at the Fort St.  
13 Vrain facility in Colorado.

14           For planning purposes, we've blocked out a period  
15 of time of roughly 40 years, to the year 2035, which is about  
16 the time, for planning purposes, the repository is going to  
17 stop accepting fuel. During this period of time, we expect  
18 to receive fuel from up to 41 research reactors. Actually,  
19 they'll be fewer, since not all of the original participants  
20 are going to ship us fuel. We have operating reactors, as I  
21 said, at Oak Ridge, the high flux isotope reactor. We have a  
22 reactor, which I believe is not operating at the moment, but  
23 may again, at Brookhaven National Laboratory, the high flux  
24 beam reactor, and we have the advanced test reactor at Idaho.

25           There are about 30 research reactors at various

1 universities around the United States, and the Navy of course  
2 is generating a good deal of our spent fuel, preponderance of  
3 the HEU.

4           I'm not going to go into all of the numbers, but  
5 for Savannah River, if we just look at the amount of fuel  
6 that we plan to receive at this site, if you look at volume,  
7 it turns out to be approximately a third of the fuel that  
8 will be shipped to one of the two sites. If you look at I  
9 guess the number of assemblies, it's probably more like two-  
10 thirds. These are the only two sites, Savannah River and  
11 Idaho, that we'll be receiving fuel in the future.

12           Foreign research reactor fuel is and has been  
13 coming to Savannah River. We expect next year, it will be  
14 shipped also to Idaho. Hanford will basically retain or keep  
15 the fuel it has in its possession, possibly shipping some of  
16 the fuel to Idaho for treatment. But all the other sites  
17 where we have fuel, we'll eventually consolidate that at  
18 Hanford, Idaho and Savannah River.

19           These are the characteristics. The first one is  
20 geometry. Geometry, the fuel meats, or the actual fuel  
21 contained within the cladding, the enrichment, the condition  
22 of the fuel, and what we call special case fuel, something  
23 particularly different about it; these are the general  
24 characteristics that we can use to describe fuel and to group  
25 it or put it into different categories.

1           For example, we have 250 different types of fuel  
2 within our inventory. To work with that fuel, that is, to  
3 analyze it and decide what to do with the fuel, how to treat  
4 it, how to store it, how it's going to behave in the  
5 repository, we group it into approximately a dozen different  
6 groups or categories so we can handle it more easily.

7           And what I've done is just given different  
8 representative fuel types that exhibit one or more of the  
9 characteristics I've shown here. For example, the N-Reactor  
10 fuel is the first one on the list. It consists of two  
11 concentric cylinders. It's a zircaloy-clad fuel element. I  
12 think it's roughly two and a half feet long. It has a  
13 uranium metal fuel meat.

14           This particular fuel, as most all of our production  
15 fuel, was designed and handled to be readily processed to  
16 obtain materials for weapons production. It was not meant to  
17 be stored long-term in our water basins. So this has  
18 experienced significant corrosion. It represents what we  
19 have under here condition, the degraded fuel.

20           The TRIGA fuel, the nomenclature, something about  
21 test reactor, I forget what the "I" is for--but it's made  
22 by--isotopes--General Atomic. This fuel is used in a number  
23 of our foreign and domestic research reactors. It's a  
24 cylindrical fuel. The cladding can be either zirconium,  
25 aluminum or stainless steel. One of the fuel meats is the

1 zirconium hydride, and this is roughly two and a half feet  
2 long, generally in good condition, and the fuel we're taking  
3 back from foreign research reactors.

4           The Advanced Test Reactor is an aluminum-based  
5 fuel, aluminum clad with an aluminum alloy meat. It has  
6 curved plate elements. Again, I believe this is roughly two  
7 and a half, three feet long. This is from our reactor in  
8 Idaho.

9           Some of the fuel I'm not going to discuss. It will  
10 be discussed by Mark Barlow later in the morning. But this  
11 is the high flux isotope reactor, an aluminum fuel.

12           The Fort St. Vrain, I guess has a unique shape,  
13 it's a hexagonal block of graphite with holes drilled to  
14 accept the fuel rods, also to allow the coolant to flow  
15 through it. The fuel is called a fuel compact. It's  
16 actually a solid rod composed of uranium carbide pellets  
17 centered with carbon, and these are inserted into the holes  
18 within the block of graphite. These are in storage both at  
19 Idaho and at a dry storage facility in Colorado that I said  
20 the Department has taken ownership of. We are also now the  
21 licensee of that facility with the Nuclear Regulatory  
22 Commission.

23           The materials test reactor fuel is another element  
24 that Mark will discuss in greater detail, since these are  
25 coming into the Savannah River plant, and this is the bulk of

1 the fuel assemblies we have to deal with at the Savannah  
2 River site.

3           The experimental breeder reactor is shut down. The  
4 fuel for the most part, or I guess all of the fuel is  
5 actually at Idaho. These are fairly long rods, about five  
6 feet long, about a half inch in diameter, stainless steel  
7 clad. And what's unique about this fuel is it has a sodium  
8 bond. There's sodium contained between the cladding and the  
9 fuel. We have about 60 metric tons, I believe, in total of  
10 sodium bonded fuel, not all EBR-II fuel. The blanket fuel  
11 contains depleted uranium.

12           Also what I guess I've neglected to mention earlier  
13 was the enrichment of most of this fuel. If we go back, the  
14 N-Reactor fuel has--

15           BULLEN: Excuse me. Could I interrupt you for just a  
16 second?

17           ECKERT: Sure.

18           BULLEN: We have a distinguished visitor in the back,  
19 Secretary Pena.

20           PENA: I heard you were having a meeting, so I thought  
21 I'd just come on down.

22           Please sit down. Please relax, Ladies and  
23 Gentlemen. I normally don't barge in on people's meetings.  
24 Let me just say to the Board members how much I appreciate  
25 your very hard work.

1           I spent yesterday touring Savannah River and  
2 looking at the tritium facility and waste processing  
3 facility, and drove by the canyons and other things. So as  
4 all of you know, this is a very important group of citizens  
5 who have volunteered their time to give us some guidance in  
6 the Congress on how we can think of complex-wide issues, and  
7 we very much appreciate their recommendations, your ideas on  
8 how we can deal with Yucca Mountain, which we have responded  
9 to, and suggestions of doing an east-west cut, and other  
10 things.

11           So I just wanted to drop by and say thank you very  
12 much. I happened to be talking to some business people next  
13 door and heard you were here, so I wanted to come by and  
14 thank you for taking the time to be here and to do this kind  
15 of work. So I appreciate it.

16           Thank you. Happy Holidays to all of you.

17       BULLEN: Thank you, sir.

18       ECKERT: What I was going to do is recap the enrichment,  
19 which I had neglected to talk about. The N-Reactor fuel is  
20 very low enriched, on the order of 1 per cent. TRIGA varies  
21 from--we still call low enriched uranium anything up to 20  
22 per cent, and TRIGA starts at about 20 and goes up to 93 per  
23 cent enrichment. The advanced test reactor is 93 per cent.  
24 Fort St. Vrain is 93 per cent, and EBR-II is, I'm not sure, I  
25 believe that's also HEU.

1           Well, let me finish up with this one. The EBR-II  
2 fuel, as I said, was sodium bonded, and we're looking at  
3 means for removing the sodium. The one treatment technology  
4 under development at Argonne National Laboratory, both East  
5 and West, is the electrometallurgical processing technology,  
6 and that is actually undergoing a demonstration today and it  
7 shows great promise for processing our more difficult fuel,  
8 such as the EBR-II fuel.

9           The EBR-II fuel is low enriched. Thank you.

10           Okay, what else might I have missed. The Three  
11 Mile Island Unit II core debris is in canisters at Idaho.  
12 And I think I've covered the other examples here.

13           Now, we have roughly 2,500 metric tons of fuel on  
14 site, more coming in. Dealing with this fuel, both the  
15 storage, transportation and ultimate disposal, is the primary  
16 activity of the group that I work for, Office of Spent Fuel  
17 Management, in cooperation with other offices within the  
18 Department.

19           Several years ago, in '93, the Office of  
20 Environmental Health and Safety at DOE conducted a  
21 vulnerability assessment and published their working group  
22 report, which listed 105 vulnerabilities with the storage of  
23 spent fuel throughout the complex. As part of our effort, we  
24 put together a plan of action to resolve these  
25 vulnerabilities, and particularly those that were of higher

1 priority, such as the fuel in storage, the N-Reactor fuel in  
2 storage at the K-Basins. Approximately half of these  
3 vulnerabilities have been completely resolved. The others  
4 are taking much longer, due to the cost and time required to  
5 deal with the fuel.

6           One of the means for resolving the vulnerabilities,  
7 besides putting it into safer, longer term storage, dry  
8 storage, is to process it, and the production reactor targets  
9 and fuel at Savannah River are being reprocessed. Actually,  
10 the targets have been completed. The step ongoing now with  
11 RW is to understand what will be required to prepare that  
12 fuel for ultimate disposition.

13           To understand better what's required, we have  
14 decided we were going to work more closely with RW, and we  
15 have. Over the past two years, we've been able to--actually,  
16 the memorandum of agreement, the standard contract that RW  
17 has with the commercial utilities is an agreement we have  
18 about ready to be signed. It's completed. It's just waiting  
19 for some final concurrences. And this describes the  
20 conditions under which RW will accept DOE-owned fuel.

21           RW and the contractors have actually done some of  
22 the analyses, criticality analyses on DOE-owned fuel. We  
23 felt although we could have done it, it was expedient  
24 initially to have the RW/M&O contractors do it, since they  
25 know exactly what has to be done, and we work closely with

1 them.

2           The key documents coming up in this year, the  
3 Viability Assessment and the Total System Life Cycle Cost  
4 Evaluation, will explicitly contain information on DOE-owned  
5 fuel. There are a number of requirements documents that go  
6 into greater detail on what DOE has to do with the fuel to  
7 have it accepted by RW.

8           There is an interface control document now in  
9 review that gives the sort of bounds perhaps on geometry, for  
10 example, that we must meet for the equipment and storage  
11 envisioned by RW in the repository. And to better understand  
12 all of these items, learn where RW is, where EM is, we've  
13 been holding semiannual strategy meetings of several days  
14 where we go into greater detail about what's in the  
15 documents, better understanding of how we're going to meet  
16 those requirements, and planning what we have to do as a  
17 group in the next six months in order to move firmly along  
18 that path toward disposal.

19           As I say, DOE-owned fuel is primarily owned within  
20 DOE by the Office of Environmental Management. I guess the  
21 Navy owns the next largest chunk. The Office of Nuclear  
22 Energy owns some, particularly that which is being stored at  
23 the operating reactors, HFIR, HFBR. But as the reactors get  
24 shut down, the operating entities within DOE want to put it  
25 into a caretaker mode, and that's where EM is taking more and

1 more of the fuel.

2           That's all I have. Are there questions?

3           BULLEN: Thank you, Howard. Questions from the Board?

4 Carl Di Bella?

5           DI BELLA: You showed on your overhead for the N-Reactor  
6 fuel, that it has zircaloy cladding.

7           ECKERT: Yes.

8           DI BELLA: And yet a substantial amount of that fuel is  
9 degraded, I understand.

10          ECKERT: Yes.

11          DI BELLA: Zircaloy really shouldn't degrade in the kind  
12 of chemical environment that that fuel has been exposed to.  
13 Do you know what has caused the degradation? Is it a  
14 corrosion phenomena of the zircaloy?

15          ECKERT: It is, but not primarily the zircaloy, but of  
16 the uranium meat within. When they unloaded the N-Reactor  
17 fuel, it was, as I understand it, just basically tumbled down  
18 a chute into bins to be collected, so it wasn't handled  
19 carefully. The cladding was damaged, breached, and where the  
20 uranium metal is exposed to water, it naturally corrodes  
21 rapidly.

22           You might be able to see some corrosion. This  
23 isn't one of the worst ones, but the cladding would tend to  
24 peel away from the meat as the corrosion products expand. I  
25 guess I didn't bring another one that had a more graphic

1 description, but these are the concentric cylinders within a  
2 canister in which it was stored in the N-Reactor spent fuel  
3 storage pool.

4 BULLEN: Jeff Wong, Board?

5 WONG: I just have kind of a general question. We've  
6 been waiting to hear about the DOE's waste isolation  
7 strategy. What role does the waste isolation strategy have  
8 in your planning of your management of your fuel forms, or  
9 what input do you have into the waste isolation strategy in  
10 dealing with your fuel forms?

11 ECKERT: Waste isolation means you're referring to the  
12 ultimate disposal. I guess our input is to work with RW to  
13 make it reasonably cost effective to dispose of our fuel. We  
14 certainly are not going to dictate a disposal criteria.  
15 That's their job. We are almost in the role of a commercial  
16 utility. We have fuel that we need to dispose of, so we have  
17 to live within the RW requirements. But as they evolve, the  
18 documents that are written, we get to review and comment on,  
19 so that we can suggest changes perhaps that would make it  
20 easier and perhaps more economical, safer for us to dispose  
21 of the fuel.

22 BULLEN: Any other questions from the Board?

23 (No response.)

24 BULLEN: Thank you very much, Howard. We're right on  
25 schedule.

1           Our next speaker is David Curtis. He is the  
2 Director of the Reactor Materials Division in Naval Reactors.  
3 In that capacity, he is actually responsible for the  
4 program's expended core facility at the INEEL, where Naval  
5 reactor cores are examined and where Naval spent fuel will be  
6 prepared for ultimate disposal. And David will speak to us  
7 about Navy spent fuel.

8           CURTIS: Thank you. I'm here today to talk about Naval  
9 spent fuel and how it fits into the DOE program. Some of the  
10 introductory material I'm going to talk about is going to be  
11 very similar to the material that Rich Guida from the Naval  
12 Reactors Program presented to a panel of the Board about a  
13 month ago. He focused mostly on transportation. I'm going  
14 to focus mostly on the repository aspects of it.

15           Details of the Naval spent nuclear fuel are  
16 classified, but the results of the analyses and tests that we  
17 have done on this are unclassified and can be discussed in a  
18 public forum such as this. More detail than I can present in  
19 the 20 minutes that I have today is provided in an  
20 unclassified basis to the appropriate parts of the various  
21 organizations involved, and detailed technical backup  
22 information is presented to people that have clearances, such  
23 as John Arendt from the Board, shortly hopefully Dan Bullen  
24 from the Board, Carl Di Bella of the Board staff, and various  
25 regulatory agencies, such as the Nuclear Regulatory

1 Commission.

2           Let me take a moment to introduce the Naval Nuclear  
3 Propulsion Program. The Naval Nuclear Propulsion Program is  
4 a joint program of the Department of Energy and the Navy. We  
5 are a part of both agencies, and the ships that we provide  
6 reactors for are a key part of the Navy's defense mission.

7           CNN regularly tells the story of these ships. As  
8 we're sitting here today, the Nimitz and the George  
9 Washington, two nuclear powered aircraft carriers, are in the  
10 Persian Gulf. Both of them sprinted to the Persian Gulf  
11 recently to provide a military presence. The Nimitz came  
12 from Hong Kong. The George Washington was in Haifa. And  
13 being able to make the high speed transits and arrive at  
14 station ready for the mission is a key part of the Navy's  
15 story.

16           The submarines that we have don't get the CNN  
17 coverage, but submarines are in most of the world's trouble  
18 spots also.

19           The Naval Nuclear Propulsion Program is about 40  
20 per cent of the Navy's principal combatants. We have about  
21 the same number of total reactors as the civilian commercial  
22 nuclear power industry. We have between 110 and 120 now. I  
23 think the commercial nuclear industry has a little fewer than  
24 110.

25           We have 4,800 reactor years of safe operation

1 without a reactor accident. That's about twice the--actually  
2 more than twice the accumulated experience of the civilian  
3 nuclear industry. I think the civilian nuclear industry now  
4 is between 2,000 and 2,500. The fact that ours is without a  
5 reactor accident is really one of the keys as to why we can  
6 take our ships with these nuclear powered reactors into over  
7 150 ports in over 50 countries worldwide.

8           This slide, which is in the pass-out, gives a few  
9 more statistics. I'm not going to spend time dwelling on it.  
10 You can review the statistics later; just a few more  
11 statistics about the program.

12           I would like to get on to talking about what Naval  
13 spent nuclear fuel is. Naval spent nuclear fuel is solid.  
14 It's metallic. It's not flammable. It's not hazardous. We  
15 have done a TCLP test and gotten EPA agreement that it is not  
16 hazardous. We have that certification.

17           Naval spent nuclear fuel is--or Naval fuel is built  
18 and designed to be operated on war ships, and as such, it  
19 needs to be very rugged. The design requirements for the  
20 fuel are well in excess of 50 g's, which is considerably more  
21 than the other types of spent nuclear fuel that you're  
22 talking about here. The Naval fuel fully contains all of the  
23 fission products and the long-lived radioactivity, and it  
24 operates for very long periods of time.

25           Most of the ships that we have now have cores that

1 are operating for about 20 years between refuelings, some  
2 longer. I'll get to that in a minute. But they operate at  
3 power for long periods of time and they operate in very close  
4 proximity to the crew. Sailors live, eat, sleep literally  
5 within feet of these operating reactors. The reactors have  
6 to be designed to be able to take rapid power transients. The  
7 commanding officer of the ship may decide he wants to stop,  
8 sprint, et cetera, so the power transients that they're  
9 designed for are much, much faster than commercial power  
10 transient. So the whole design philosophy of the Naval fuel  
11 drives us to come up with very, very rugged reactor cores.

12           We periodically take ships and do shock tests of  
13 the ships. Let me move this up on the screen a little bit.  
14 This is a picture of a shock test of the Theodore Roosevelt,  
15 one of our carriers. What this represents is tens of  
16 thousands of pounds of high explosive being detonated about a  
17 ship length away from the ship. That rocks the ship pretty  
18 well. The reactor cores come through this kind of an  
19 experience very straightforwardly, very easily.

20           Key to what we say about the behavior of fuel in a  
21 repository is our knowledge of the fuel. We have very  
22 detailed knowledge of the fuel through the manufacturing  
23 process. After the cores are made, they go through a design  
24 and a manufacturing certification. Then they go through  
25 detailed acceptance testing. The testing is followed in

1 detail, and the operation through core lifetime is followed  
2 in a great deal of detail, so that we know very well what  
3 each and every reactor core is, how it was made, how it was  
4 designed, how it's operating, how it's behaving.

5           After we're done with the cores, after they get  
6 removed from the plants, from the ships or the prototype  
7 reactors, we then take them all to the expended core facility  
8 that Dr. Bullen mentioned at the Idaho National Engineering  
9 and Environmental Lab, INEEL, where we examine each of the  
10 cores to basically confirm that the performance of the core  
11 was in fact what we expected. On some limited number of  
12 cores, we do destructive, more detailed examinations to  
13 understand more.

14           In addition to that, we have an extensive radiation  
15 test program, and have had for many years. We've used MTR,  
16 ETR. We're currently using ATR, where we thoroughly explore  
17 the failure modes of our fuel. We think we know pretty well  
18 what causes them to fail, where the limits of mother nature  
19 are, what doesn't cause them to fail, et cetera.

20           The key to much of what we know and think about the  
21 behavior of our cores is based on the examinations that we do  
22 in Idaho of the spent cores.

23           There's a very powerful economic and a very  
24 powerful operational incentive to have long lifetime cores,  
25 but we couldn't do it unless we were confident technically

1 that the cores could last. Our first core, which was put  
2 into the Nautilus in the early Fifties operated for two  
3 years. Our newest ship, the Seawolf, which was commissioned  
4 in July of this year, has a core in it that we expect to last  
5 the 30 year life of the ship, will never be refueled, will  
6 have only one core, will be the core load for that ship.  
7 That's a big improvement, both in terms of just the  
8 availability on line, the cost and economics in terms of  
9 buying the cores and maintaining the cores, and also in  
10 having the reduced number of cores to worry about for  
11 reprocessing.

12           Before 1992, after we had examined the cores at the  
13 expended core facility, they were sent to the Idaho Chemical  
14 Processing plant, where the unused uranium 235 was recovered.  
15 In 1992, reprocessing of our cores at ICPP was stopped, or  
16 the decision was made. Since then, we're in kind of a  
17 transition phase. We've been preparing our fuel for  
18 repository disposal. We've got some construction projects  
19 underway to facilitate that in Idaho.

20           What we intend to be doing is we intend to  
21 basically canisterize the fuel. In '96, we had an EIS that  
22 was published where we looked at various kinds of container  
23 systems. We had a Record of Decision in December of 1996.  
24 We selected a Dual Purpose Canister. What we intend to do is  
25 to put the fuel in canisters in Idaho. We will be storing it

1 temporarily in Idaho until either an interim storage site or  
2 a repository is open for it. Then we will be responsible for  
3 transporting it to either the repository or an interim  
4 storage site when one is open.

5           The DPC system is currently in a design stage, and  
6 we're working on that. It's our goal, it's our intention  
7 that once these canisters arrive at a repository, only the  
8 canister needs to be handled. The fuel won't need to be  
9 handled. We can't really make that claim on the record until  
10 the acceptance criteria for the repository get established,  
11 but we expect and it's our intent to try to make the DPC  
12 sufficient for emplacement in the repository.

13           Let me talk some about the amount of Naval spent  
14 fuel that we're talking about here. As Howard mentioned,  
15 there are a variety of different metrics. There are a  
16 variety of different coins of the realm. The one most  
17 frequently used is the metric tons of heavy metal.

18           We currently have 14 metric tons of heavy metal of  
19 Naval spent fuel. By 2035, we are projecting 65 metric tons  
20 of heavy metal. I have on this slide also the DOE non-naval  
21 fuel now and in 2035, and an estimate of the commercial spent  
22 nuclear fuel now and in 2035. The small amount of Naval fuel  
23 is really largely because we have fairly small reactors, and  
24 very infrequent refuelings.

25           So ultimately, by the coin of the realm, we'll have

1 about a tenth of a per cent of the spent nuclear fuel in the  
2 country, about a tenth of a per cent of the amount of spent  
3 nuclear fuel that is intended for the first repository.

4           Other possible metrics are total weight or volume  
5 or piece count, as Howard mentioned. The 65 metric tons  
6 represents volume of canisters, or weight of canisters.  
7 These number over here are volume and weight of the waste  
8 form within the canisters. We'll have about 300 canisters of  
9 Naval spent fuel, which is I think about 3 per cent of the  
10 number of canisters, the piece count of canisters for the  
11 repository. Each canister will be about 66 inches in  
12 diameter, some slight variation in lengths, but the longest  
13 ones will be 212 inches.

14           We've at this point calculated several aspects of  
15 the performance of Naval fuel in a repository. The next  
16 several slides summarize some of the key results that we've  
17 come up with.

18           Because the design and manufacturing of the fuel  
19 drives us to build very, very rugged fuel, we calculate on a  
20 best estimate basis that the fuel will remain substantially  
21 intact in a repository environment for in excess of a million  
22 years. In that million year period of time, the  
23 radioactivity in the fuel has decayed by more than four  
24 orders of magnitude.

25           Specifically, the best estimate prediction is that

1 the cladding will not be penetrated by corrosion on any of  
2 the elements; that the best estimate prediction is there will  
3 be zero corrosion through the cladding in a million years in  
4 any of the elements. Now, this doesn't assume that the  
5 canister provides a whole lot of protection. We're assuming  
6 that 10 per cent of the canisters are degraded and  
7 ineffective at about 3,000 years, and the rest of them are  
8 degraded and ineffective at 10,000 years.

9           We further conclude that the basic fuel assembly  
10 geometry is maintained intact for in excess of a million  
11 years.

12           As a result of this, the only radioisotopes that we  
13 expect to be released to the drift are from the crud layer  
14 that's on the fuel elements, and small amounts of impurities  
15 and activated elements in the cladding that corrode, and as  
16 soon as the cladding does corrode, to the extent that it does  
17 corrode, we assume that all of that material is released to  
18 the drift.

19           The peak release rate is much less than a curie per  
20 year. It is carbon-14 and it occurs relatively early in the  
21 repository life.

22           Now, although we don't anticipate it, we have  
23 calculated so far two cases that are hypothetical cases where  
24 if there is something that happens to expose the fuel  
25 material itself, what would the consequences be. In one

1 case, what we calculated was a mechanical damage, such as  
2 from a rock fall, that physically breaks the canister, or  
3 what's left of the canister, breaks any of the internal  
4 support structure, and shears through our fuel.

5           The other case, what we assumed is if we had enough  
6 corrosion so that it corrodes through the fuel by whatever  
7 reason, and that's, you know, beyond the 99.98 percentile on  
8 this, in this case, we would be exposing 1,000 elements, in  
9 this case, we'd be exposing a couple hundred elements. In  
10 neither case are we contributing significant uranium, or for  
11 that matter, significant fission products to the drift.

12           We've also done nuclear criticality calculations  
13 for a number of cases. We do use highly enriched uranium in  
14 our Naval reactor cores. The fact that we use highly  
15 enriched uranium is not a show stopper, we think. It turns  
16 out the amount of U-235 per container is about the same  
17 amount as the amount of U-235 per container of commercial  
18 spent nuclear fuel. So we're not putting a lot more U-235 in  
19 there.

20           As a matter of fact, the fact that we don't have  
21 all the 238 at the beginning of life of our reactor cores  
22 means that to a substantial extent, we don't need to worry  
23 about plutonium. So we do have what we think is a tractable  
24 amount of uranium-235.

25           In addition, though, we are fixing with each of our

1 fuel assemblies hafnium control rods that are permanently  
2 attached to the assembly that will provide a nuclear poison  
3 for a long time. Hafnium corrodes even less than zircaloy.  
4 We think the hafnium control rods are going to be a nuclear  
5 shut-down presence for far in excess of the million years.

6           We've analyzed a number of cases with intact  
7 containers, degraded containers, degraded structural supports  
8 within the containers, damaged containers, a variety of  
9 assumptions on flooding, partially flooded, fully flooded,  
10 preferentially flooded, et cetera, et cetera, and in all of  
11 these cases, we think we can show that as long as we maintain  
12 the basic geometry of our fuel assemblies, that we have  
13 adequate nuclear shut-down from a criticality point of view.  
14 And we think that the basic geometry of the assemblies will  
15 be maintained for in excess of the one million year time that  
16 we're calculating here.

17       DI BELLA: A quick question, please, Carl Di Bella here,  
18 on the hafnium control rods. Are they the control rods that  
19 are actually part of the core, or are these new hafnium  
20 control rods that you put in?

21       CURTIS: The answer is some of both, probably, a  
22 mixture. We will use hafnium control rods where we have  
23 hafnium control rods. We will put some new hafnium in where  
24 we don't.

25       DI BELLA: How much does hafnium cost?

1           CURTIS: I'd hesitate to answer that off the top of my  
2 head. I can get you an answer. I guess I would say that  
3 this whole adventure is not cheap, but it's part of the price  
4 of doing business.

5           Finally, let me talk about decay heat. Decay heat  
6 is just not a problem. The decay heat per container for us  
7 is about half of the decay heat of the spent fuel of a  
8 typical commercial spent fuel container.

9           The peak repository heat load in any year from all  
10 of the Naval containers is about 700 kilowatts. That's about  
11 500 to 600 of those hair dryers that are in the hotel rooms  
12 up here. So it's not a tremendous decay heat load.

13           For about three years now, we've been interacting  
14 very closely with RW. It kicked off in November of '94.  
15 Some of us had a meeting with Dan Dreyfus and Lake Barrett,  
16 and since then, we've had a very close relationship with RW  
17 and with EM in terms of making sure that the things that the  
18 DOE is doing for the repository are in fact compatible with  
19 incorporating Naval spent nuclear fuel.

20           I'd like to highlight just a couple of things on  
21 this slide. The previous four slides, I gave some summary  
22 results of corrosion and degraded conditions and nuclear  
23 criticality and decay heat. All of that, and more, was  
24 provided in a document about an inch thick in July to the  
25 Yucca Mountain Site Characterization Office with detailed

1 calculations for use in their EIS. So that information has  
2 been presented. It's unclassified. Copies of that were made  
3 available to the Board and the Board staff.

4           The second thing, as Howard mentioned for EM, we  
5 are also working on a memorandum of understanding with RW.  
6 What that will do is that will nail down the details of the  
7 interfaces between us and RW to make sure that all aspects of  
8 controlling Naval spent nuclear fuel are adequately  
9 controlled as we go to the repository.

10           The third thing I would mention is that our  
11 principal prime contractor that supports us on this, the  
12 Bettis Atomic Power Laboratory, has opened a field office in  
13 the Yucca Mountain Site Characterization Office/M&O facility,  
14 so that we have an on-site presence in Las Vegas working  
15 closely with the M&O to make sure that we have coordination,  
16 cooperation, communication, et cetera, et cetera.

17           We also, it was on that slide and I glossed over  
18 it, but we will provide the detailed analyses of Naval spent  
19 fuel to the DOE so that DOE can use those in the license  
20 applications to the Nuclear Regulatory Commission for the  
21 repository licensing process.

22           In summary, I guess my points are that the same  
23 attributes that drive us to design and build reactors that  
24 are rugged for operation in extreme conditions, maneuvering,  
25 battle shock, et cetera, et cetera, that are suitable for

1 long-term operation in an enclosed environment like a  
2 submarine, with the crew in very close proximity, living in  
3 close proximity for months at a time, those same  
4 characteristics of the fuel, although it wasn't our  
5 motivation in designing it this way, turns out make the fuel  
6 almost uniquely--maybe I shouldn't say uniquely, but at least  
7 very suitable for repository emplacement.

8           Furthermore, our knowledge of the design of the  
9 reactor, the manufacturing of the reactor, the operation of  
10 the reactor, the post-irradiation examination of the reactor,  
11 and our knowledge of the behavior of the fuel system from our  
12 radiation test data lead us to believe that we have a solid  
13 basis for predicting the performance of our fuel in a long-  
14 term environment.

15           As I mentioned, we've got the detailed design of a  
16 number of things going in progress. We've got tests and  
17 analyses that are in progress right now. But we're confident  
18 at this point that the defense in depth that we have in our  
19 fuel and in our understanding of the fuel will basically  
20 confirm the situation that we think we have now where the  
21 strength and the integrity of the fuel will be maintained in  
22 excess of a million years. We believe we will show that.

23           The hafnium will remain in excess of a million years and  
24 will in fact provide adequate nuclear shut-down, and that the  
25 total releases of fission products and U-235 from Naval fuel

1 will present a very insignificant contribution to the total  
2 dose rate coming from the repository.

3           That's all I have. Are there any questions?

4           BULLEN: Thank you, David. Questions from the Board?  
5 John Arendt?

6           ARENDR: David, I think you showed the corrosion--or you  
7 didn't show the corrosion rate of zircaloy, but I believe you  
8 have corrosion data for some 30 or 40 years that you're  
9 basing your conclusions on, aren't you?

10          CURTIS: Yes. And as a matter of fact, yesterday and  
11 the day before, one of our materials scientists from the  
12 Bettis Atomic Power Laboratory was presenting a paper on  
13 zircaloy corrosion--I'm not sure which of those days, he  
14 didn't do it both days--but on one of those days, there's an  
15 expert elicitation panel this week somewhere on the West  
16 Coast, and we're presenting a paper on zircaloy corrosion at  
17 that meeting.

18                 Basically, what we have is a lot of experience out  
19 of in pile irradiated corrosion. We have tens of thousands  
20 of samples that we have irradiated, most of them in the  
21 advanced test reactor, for periods of time up to--well, ATR  
22 is now almost 30 years old, it's about 30 years old. There  
23 are some of those specimens that have been in almost from the  
24 inception. We have specimens that have been in for about 30  
25 years in autoclaves at the Bettis Atomic Power Laboratory,

1 and we have some--I mentioned the expended core facility--  
2 some of our earliest cores we've been keeping around as  
3 library samples just because there are unique sorts of  
4 things. A few years ago, we took out a core that had been  
5 sitting in the water pits at ECF for 28 years with no  
6 appreciable degradation of that.

7           But we have what I would characterize as good  
8 reliable engineering data at about the 30 year point, both in  
9 pile and out of pile. What we believe is that for the  
10 repository situation, the out of pile is probably more  
11 pertinent.

12         SAGYIS: Sagns, Board. On the corrosion matter again,  
13 the kind of environment that you would expect in the  
14 repository would be significantly different perhaps from some  
15 of the environments for which you have long-term experience.  
16 The accumulation or potential accumulation of chloride ions,  
17 for example, in the water, the concentrations that may exceed  
18 by several orders of magnitude where you may have experience  
19 would be an example of that.

20           Indeed, in a number of highly corrosion resistant  
21 materials, including super alloys and so on, have been looked  
22 at for the outer waste package, and for those materials,  
23 estimates of penetration after a million years would be  
24 considered very optimistic.

25           I just wanted to know a little bit more about how

1 they arrive at those estimates of, for example, one million  
2 years of penetration time. And you mentioned some  
3 percentages when you showed the transparency, the one on  
4 calculation results. Maybe you want to put it up again.  
5 That was the million years, like 1 per cent or 10 per cent.

6 CURTIS: This one?

7 SAGYIS: Yes.

8 CURTIS: I think perhaps what I mentioned was on this  
9 one, I mentioned the 99.98 percentile there, in the case of  
10 the accelerated corrosion. Let me address first your point  
11 on the fact that the environment in the repository is not  
12 likely to be the environment that we have our long-term data  
13 on. You're absolutely correct on that. We've looked very  
14 carefully at the environment that may be there.

15 We just recently started some tests at Livermore in  
16 the J-13 water. We're doing some tests in concentrated J-13  
17 water, but those have literally just started. So we don't  
18 have any real data on those.

19 We've looked at the types of chemical species that  
20 can be there. We've looked at the way that that could  
21 impinge on our waste form, and we think, and this gets into a  
22 little bit of the classified aspects of it, so I can't  
23 discuss this in great detail here, but we think that the  
24 bottom line is still going to be true. Even if there is some  
25 local penetration because of a persistent drip or something

1 like that, we think that is going to be very well bounded by  
2 the types of analyses that we've done that assume much more  
3 than very localized penetration in one spot.

4 SAGYIS: And that is estimated at 99.98 percentile for  
5 how many years?

6 CURTIS: This is for a million years. This is projected  
7 out to a million years.

8 SAGYIS: That's for a million years.

9 CURTIS: And the first penetration of that at that rate  
10 comes through at about 300,000 years, and so there is  
11 corrosion from the 300,000 year point to the million year  
12 point in this. There is exposure of the fuel material to the  
13 drift environment for 700,000 years there.

14 SAGYIS: And that is ascribed primarily to the corrosion  
15 resistance of the zircaloy that would be surrounding the  
16 elements?

17 CURTIS: Substantially, yes.

18 SAGYIS: Substantially. And what kind of thickness is  
19 that?

20 CURTIS: That's a classified number I'm afraid I can't  
21 discuss. I would say that we have provided that information  
22 to John Arendt and Carl Di Bella.

23 BULLEN: Paul Craig?

24 CRAIG: Craig, Board. Yes, I'm struck by the difference

1 between your million year time frame and the 10,000 year type  
2 of time frame from the DOE designs for the external  
3 canisters. This has to do with the materials that you're  
4 using. Can you tell us something about the nature of the  
5 materials that gives this two order of magnitude longer time  
6 frame for corrosion?

7       CURTIS: Well, zircaloy, which is the cladding that we  
8 use and is heavily used in the structural material of our  
9 reactors, was designed specifically as a non-neutron  
10 absorbing, corrosion resistant material. As a matter of  
11 fact, zircaloy was invented at Bettis Atomic Power Laboratory  
12 back in the late Forties or early Fifties at the start of the  
13 Naval Nuclear Propulsion Program, and it was designed and  
14 it's been tweaked through the years to make it more and more  
15 corrosion resistant.

16               Zircaloy 4 is what's used today in industry  
17 substantially, and zircaloy 4 is better than the earlier  
18 versions of zircaloy.

19               One of the characteristics of zircaloy corrosion is  
20 that the corrosion film tends to be very tightly adherent to  
21 the base metal, and tends to form a protective oxide film on  
22 the base metal. So that actually for a substantial period of  
23 time, the zircaloy corrosion protects the underlying base  
24 metal.

25               After a while, zircaloy goes through transition,

1 and then you get a steady state linear corrosion rate with  
2 time that's very well characterized. We've done a thorough  
3 analyses, electron microscopy, ta-da, ta-da, ta-da, to study  
4 just how that layer builds up, how it's structured, why it  
5 behaves as it does. The details of that really are  
6 substantially in this paper that I mentioned as being  
7 presented yesterday or the day before out on the West Coast.

8           That paper, I haven't seen a final published copy  
9 of that paper. Once there is a final published copy of that  
10 paper, I have already told Carl Di Bella I would make copies  
11 available to the Board.

12         BULLEN: Di Bella, Board?

13         DI BELLA: Switching subjects a little bit, you  
14 mentioned you used to, until 1992, reprocess spent Naval  
15 fuel. Where did the uranium go that was recovered from that  
16 reprocessing operation? Do you know? And were any of the  
17 other actinides recovered for any purpose, as far as you  
18 know?

19         CURTIS: I don't know the answer to that directly  
20 because Naval Reactors wasn't responsible for operation of  
21 ICPP, so I don't know the answer about the actinides  
22 specifically. I could try to find out the answer.

23           Before 1992, I was in a different job than the one  
24 I'm in now, and so I didn't even have the contact with ICPP  
25 that I have now.

1           In terms of the ultimate use of the U-235 that was  
2 recovered, my understanding was that was provided to the DOE  
3 for the weapons production reactors, I believe principally  
4 Savannah River, but some may have also been provided to other  
5 DOE reactors. But the ICPP is run by a different part of DOE  
6 than Naval Reactors.

7           Before 1992 when we gave them our fuel, we thought  
8 we were done with it. Now, it turns out there is some  
9 inventory of Naval fuel at ICPP, and on one of the slides I  
10 showed, it talks about a second Record of Decision in April  
11 of 1997. We have decided and have DOE concurrence that we  
12 will bring that Naval fuel back to the expended core facility  
13 and we will canisterize it and prepare it for repository  
14 disposal. So we are retaking possession of that fuel that  
15 was in the back log when the decision to stop reprocessing  
16 was made.

17         BULLEN: Bullen, Board. One last quick question. You  
18 talk about the 300 canisters and their size and their  
19 dimensions. Do you have any data or calculations on the  
20 surface radiation dose rate that you expect those canisters  
21 to have?

22         CURTIS: For transportation, there is a transportation  
23 overpack, and I don't have that number off the top of my  
24 head. If I can ask for a voice from the audience, is there a  
25 number?

1           Less than 10 milligrams per hour at 2 meters with a  
2 transportation overpack.

3           BULLEN: That's the transportation. Now, what you're  
4 assuming is that these containers would then be overpacked  
5 again at the repository?

6           CURTIS: We're assuming that there will be some sort of  
7 repository overpack or something for handling at the surface  
8 facility and to transport them into the drift. And in the  
9 October Board meeting, there was a discussion of how they  
10 would then get unpacked from the overpacks once they're down  
11 into the drift, so that as they sit in the drift, they  
12 wouldn't have that repository overpack there, is my  
13 understanding.

14          BULLEN: Well, that's the follow-on question, is do you  
15 know the radiation dose in repository? Has that been  
16 calculated?

17          CURTIS: It has been calculated. It will be high around  
18 the sides of the container. It won't be high on the ends of  
19 the container. The ends of the container will be shielded.  
20 They'll be welded. And so there's enough shielding built  
21 into the conceptual container design so that we can go do the  
22 welding. So it won't be high at the ends, but it will be  
23 high at the side, and I can't off the top of my head give you  
24 a number. I can get that number and give it to you later,  
25 but I don't have it off the top of my head.

1           BULLEN: I guess the key to the calculation would be the  
2 spectra that you'd expect to see. Primarily the cesium and  
3 strontium are the dominant gamuts, but after a few half  
4 lives, they're gone and you end up with the neutron spectra,  
5 which is going to be significantly different than the neutron  
6 spectra you'll get from commercial fuel. And so I'm  
7 interested in the concept of self-shielded packages, which  
8 means that you could have worker access in drift, and so  
9 that's why I'm asking the question.

10           A follow-on question essentially would be in dry  
11 storage, have you also done the calculation on the radiation  
12 associated with the near-field and any potential radiolysis  
13 effects and degradation products that may end up accelerating  
14 the degradation of the clad if you're making nitric acid at  
15 the surface?

16           CURTIS: In dry storage, we expect them to be dry. So  
17 we don't expect to be making nitric acid at the surface.

18           BULLEN: Zero per cent relative humidity dry storage; is  
19 that--

20           CURTIS: We are going to back fill them with inert gas,  
21 yes.

22           BULLEN: Okay.

23           CURTIS: We will dry them and back fill them with inert  
24 gas.

25           BULLEN: I guess in the long-term analysis then, if your

1 package does breach and you have moist air in there, do you  
2 still expect the million year performance of the clad, is  
3 kind of the bottom line question?

4       CURTIS: Yes, the bottom line question is yes, we assume  
5 that the package will breach. We assume that the package  
6 will breach in the repository. We assume that the package  
7 will breach. We assume that when the package breaches, we  
8 are putting stuff, which is other than nice pure water, into  
9 the proximity with the fuel assemblies, and we expect that we  
10 will have the million year performance under those  
11 conditions.

12       BULLEN: I look forward to seeing those analyses if I  
13 ever get cleared.

14       CURTIS: Yes, sir. We look forward to presenting it to  
15 you.

16       BULLEN: In an effort to keep us on schedule today, I  
17 think what we'll do is move right on to our next speaker.

18               Our next speaker is Mark Barlow from the  
19 Westinghouse Savannah River Corporation, and Mark will speak  
20 to us about the aluminum clad, highly-enriched spent fuel.

21       BARLOW: Good morning. I'm the manager of the Alternate  
22 Technology Program for Westinghouse and DOE Savannah River.  
23 My job involves the ongoing activities in performing tests,  
24 analyses and engineering studies for preparing the spent  
25 fuel, specifically the aluminum clad spent fuel, for interim

1 dry storage at Savannah River and ultimate disposal in a  
2 geologic repository.

3           There are two other speakers who will be speaking  
4 after me this morning on topics related to the engineering  
5 and the science behind alternate technology. My objective  
6 with my presentation is to introduce in a little bit more  
7 detail than Howard did a description of the aluminum clad  
8 fuel, its corrosion characteristics. I'll also describe to  
9 you the quantities and locations of the spent fuel today, and  
10 the projections for the future, and I'll also address the  
11 management strategy that we have developed for, again, the  
12 interim storage and ultimate disposal of this fuel.

13           The diagram to your left are two examples of  
14 typical configurations of the majority of the spent fuel that  
15 we refer to as research reactor spent nuclear fuel. 80 per  
16 cent of the fuel assemblies that we will be dispositioning  
17 are constructed of flat or curved or involute plates, as  
18 shown in that diagram. The box type shown on the left side  
19 of that diagram is the most common configuration. We refer  
20 to it as the material and test reactor, or MTR assembly.

21           Typically, the majority of them are a little over a  
22 meter in length, about 8 centimeters in width, and consist of  
23 somewhere in the neighborhood of 25 flat or curved plates.  
24 The typical loading of uranium in assembly as shown there is  
25 about kilograms. The enrichment can vary anywhere from low

1 enrich, below 20 per cent, but more commonly for the majority  
2 of that fuel type, they are highly enriched, in many cases,  
3 above 90 per cent enrichment U-235.

4           The involute type that's shown to the right, the  
5 majority of that type is what Howard referred to earlier as  
6 the high flux isotope reactor, or HFIR cores that have come  
7 from the Oak Ridge HFIR reactor.

8           The length of that element is a little bit less  
9 than a meter in length, a diameter of about 43 centimeters,  
10 and it is a much heavier element with about 10 kilograms of  
11 uranium in the uranium oxide core.

12           The spent fuel core material for the aluminum fuel  
13 is a two-phased alloy comprised of a dispersement of uranium  
14 aluminide, oxide or silicide and uranium pure aluminum  
15 matrix.

16           Typically the core plate thickness is about a half  
17 of a millimeter and is enclosed in cladding that's about  
18 three-eighths of a millimeter thick. Generally the fuel that  
19 we receive from the various research reactors is in very good  
20 condition, and in Dr. Iyer's talk, he will show some pictures  
21 of the condition the fuel is in.

22           There are instances of aluminum fuel having  
23 corrosion difficulty. Howard referred to those earlier.  
24 Those in our basins generally are the production fuel related  
25 targets or fuel assemblies that came out of our production

1 reactors at Savannah River site, and those are being  
2 dispositioned, as Howard said, through the canyon  
3 reprocessing process.

4           I've already mentioned that the enrichments for  
5 most of those fuel types are highly enriched, and in many  
6 cases, more than 90 per cent.

7           The time in the reactor varies greatly. Some of  
8 the fuels, essentially fresh fuel, there are some instances  
9 of fuel failure and removal of the fuel and put into  
10 temporary storage, but typical burn-up is in the range of  
11 about 30 per cent.

12           The thermal out put for the fuel assemblies also  
13 covers a fairly broad range, but in general, it's very low,  
14 and a typical number you can put it on a hook is about 10  
15 watts per assembly.

16           Because it is a two-phased aluminum alloy, the  
17 corrosion behavior is very similar to standard aluminum. The  
18 onset of corrosion is very much a function of the environment  
19 of which it is stored, and aluminum by nature is very  
20 susceptible to humidity and the temperature of the storage  
21 environment. However, we have found by experience that after  
22 the onset of corrosion, there's a formation of a thin oxide  
23 layer of the order of 10 to 50 microns in thickness, and that  
24 essentially provides a protective coating, that as long as it  
25 is maintained, it significantly reduces ongoing corrosion

1 rate.

2           When the cladding is breached and the fuel material  
3 is exposed, the release rate of the fission products is, as  
4 compared to commercial fuel, a much lower rate, and that has  
5 to do with the fact that the fission products are bound in  
6 solution with the aluminum matrix as opposed to commercial  
7 fuel where it may be more in a gas state.

8           In my talk in just a few moments, and then later in  
9 Dr. Iyer's talk, I will describe for you the program that we  
10 have going on at the Savannah River site that includes  
11 material testing and analysis, and is built upon the  
12 operational experience that we have here at Savannah River.  
13 I'll next describe our current program and the future program  
14 for managing this fuel.

15           With this diagram, what I'm intending to describe  
16 is how we will be managing the fuel at the Savannah River  
17 site and its final end state disposition. On the left-hand  
18 side, we have grouped the fuel by its either location or its  
19 relative stability with regards to environmental, safety,  
20 health implications.

21           The disposition path is shown from left to right.  
22 The boxes that are--it may be difficult to see, the shading  
23 is a little bit light--but the shaded boxes pertain to the  
24 disposition path of the research reactor spent fuel, and I'll  
25 walk through that in a moment. The solid boxes here

1 represent either existing facilities or activities within the  
2 Savannah River site. Those that are in a dashed box are  
3 planned future facilities. And over here, of course is the  
4 ultimate end state for disposal of the material.

5           Down here in the legend, and it's hopefully more  
6 legible in your hand-out, although it's a small copy, are a  
7 list of the various NEPA actions that pertain to  
8 dispositioning the materials here that I'll be describing.  
9 And I will not, for the sake of time, describe each of those  
10 in detail.

11           Across the top line, these are the materials that,  
12 as I mentioned, have come from our production reactors. They  
13 do include some research reactor material. For example, the  
14 Taiwanese Research Reactor was deemed to be an at risk  
15 material that should be dispositioned through the F&H canyons  
16 at the Savannah River site, so they are being reprocessed and  
17 they will be disposed of through the vitrified waste form  
18 from DWPF.

19           Generally, the topic for this morning pertains to  
20 the materials here in the category referred to as the stable  
21 aluminum fuels and targets, and there are some stainless and  
22 zirconium clad fuel which are in our receiving basis for off-  
23 site facility, off-site fuels facility. The stainless and  
24 zirconium fuels are destined to be shipped at sometime,  
25 probably in the 2010 time frame, to INEEL, and I do not plan

1 in my presentation this morning to be discussing the  
2 disposition of those fuels any further.

3           Future receipts of domestic and foreign research  
4 reactor fuel will be received into our receiving basin for  
5 off-site fuels, and I believe that on the tour tomorrow,  
6 you'll be visiting that facility. We also, because of space  
7 limitations, we are also storing fuel in the basin, the wet  
8 storage basin that's in proximity to our L-reactor.

9           Our plan, as Howard alluded to a few moments ago,  
10 is for those fuels that are not deemed to be at risk and  
11 would, therefore, be stabilized in the canyons. Otherwise,  
12 they will be transitioned out of wet storage into a new yet  
13 to be constructed dry storage facility at the Savannah River  
14 site, and then prepared for shipping to a geologic  
15 repository.

16         BULLEN: This is Bullen, Board, just a quick question.  
17 That's predicated on the fact that the repository will accept  
18 the fuel in that form? If it doesn't accept it in that form,  
19 then it goes back for some other stabilization?

20         BARLOW: Well, the objective of this facility is to have  
21 a road ready waste form, that we can demonstrate that it will  
22 meet the repository waste acceptance criteria.

23         BULLEN: So you'd really like to have Number 7 come  
24 before that box where you've got the repository, EIS that  
25 says yes, we'll take this waste form before you put it in

1 treatment and interim storage? I'm just trying to figure out  
2 your sequence.

3 BARLOW: Yeah, relative to the timing. Well, the reason  
4 it's reflected that way is based on the current schedule of  
5 activities.

6 BULLEN: Right, I understand that.

7 BARLOW: It's hard to predict. We would like to get out  
8 of wet storage into dry storage. If as a result of the EIS  
9 and the licensing process for the repository additional  
10 treatment is required, then yeah, we'd have to cycle back  
11 through this treatment.

12 BULLEN: Okay.

13 BARLOW: On the left-hand side, I've just presented a  
14 graphic of our plan for transitioning from wet storage into  
15 dry storage. Our current anticipated start-up for this new  
16 facility is in the year 2005. And as you'll see, these  
17 inventories here to the far left represent materials that are  
18 being stabilized through the canyons, and those basins were  
19 projected to be the inventory around the year 2000.

20 L-Basin and RBOF would be de-inventoried in about  
21 2009 or 2011. So, again, to answer your question, Dan, our  
22 intention is to get out of these basins, reduce the operating  
23 costs associated with maintaining those basins.

24 For this graphic, I'm just attempting to describe  
25 by three different measures what the current and future or

1 projected inventory of spent fuel will be. This does not  
2 necessarily represent the total inventory that will be in dry  
3 storage, depending upon the availability of the repository.  
4 But at least it's the throughput for this new facility that's  
5 destined for the repository.

6           The quantities that are shown here either in terms  
7 of the number of assemblies or cubic meters, volume or mass,  
8 in terms of metric tons of heavy metal, these quantities do  
9 include the 7,000 assemblies of INEEL fuel just for  
10 completeness. It also includes some material which are  
11 deemed to be potentially, but has not been decided through  
12 NEPA action, potential candidates for reprocessing. And they  
13 constitute a big percentage of the total mass, such that if  
14 they are reprocessed, ultimately what would be sent to the  
15 repository would be about 20 metric tons below the quantity  
16 that's shown here.

17           These materials, the reason they're considered  
18 candidates for reprocessing is that they include the EBR-II  
19 material, which at Savannah River has been de-clad, and  
20 because of it's uranium metal condition is deemed potentially  
21 not a good candidate for direct disposal at the repository.

22           Likewise, there are some particulate target  
23 residues from medical isotope production which are now in  
24 about 900 cans and don't constitute a very large mass.

25           As I mentioned, assuming that these candidates are

1 reprocessed, then in the end, there will be about 300 cubic  
2 meters of material that would be prepared for disposal in the  
3 repository. That is about 24 metric tons, as I said, lower  
4 than what's shown here, about 70 per cent of which, unless  
5 it's treated and diluted, would be highly enriched uranium.

6           A question had come up in an earlier talk about the  
7 number of canisters. We have not made the final selection of  
8 the technology for preparing for disposition of this  
9 material, and I'll discuss that in a moment, but depending  
10 upon the technology, the number of canisters that would be in  
11 this facility containing fuel in a dry storage environment  
12 range--our best estimates at this point range from somewhere  
13 between 400 to over 1,000, maybe 1,400 canisters. Those  
14 canisters would be then shipped to the repository and loaded  
15 in a waste package at the repository. By volume, this  
16 material represents less than 1 per cent of the total  
17 inventory that's projected to be placed in the repository.

18       SAGYIS: Excuse me. Saggys, Board. Again, the ordinate  
19 in that graph--

20       BARLOW: Oh, I'm sorry, it's the number of assemblies.

21       SAGYIS: Number of assemblies?

22       BARLOW: Number of assemblies.

23       SAGYIS; Thank you.

24       BARLOW: As I mentioned with do have a program for  
25 making determination of the best technology for

1 dispositioning and preparing this material, and that's what  
2 I'll be describing next.

3           In late 1995, DOE assembled a team of experts from  
4 within the DOE complex for people who have had working  
5 experience with these materials, and gave them the mandate to  
6 come up with a strategy for safe interim storage and disposal  
7 in a geological repository. This task team conducted its  
8 review over a several month time frame, and included a number  
9 of interactions with what's referred to as our technology  
10 champions, folks who thought that they had the best  
11 technology to disposition this material and prepare it for  
12 storage and disposal.

13           This charter came as a result of decisions and  
14 announcements that had been made by the Secretary of Energy  
15 to discontinue operation of the reprocessing alternative.  
16 And so this task team's job, by its terminology, was to come  
17 up with alternatives to the reprocessing technology.

18           In the task team report, they do show, for  
19 comparison purposes, costs and schedules associated with the  
20 reprocessing option, and as I indicated earlier, it is  
21 currently the Department's policy that those materials that  
22 are a potential health and safety risk would be reprocessed  
23 if there is an imminent concern.

24           In your hand-out, the following two slides contain  
25 a list of the alternate technologies that the task team spent

1 a number of hours and days and weeks evaluating and  
2 deliberating between themselves. Again, in the interest of  
3 time, my intention is not to present all of those, but rather  
4 describe their conclusions and the recommendations, and then  
5 in a little bit more specificity, the two technologies that  
6 we're pursuing today.

7           In evaluating the alternative technologies that are  
8 listed in your package, the task team used a multi-attribute  
9 analysis, and came up with a consensus among the team as to  
10 what were the conclusions of evaluating against the criteria  
11 that are listed here.

12           The first criteria was a judgment on their part of  
13 what they thought the confidence or the likelihood of success  
14 of being able to implement the technology, and within the  
15 established cost and schedule and the technical performance  
16 parameters as best they were understood at the time. This  
17 particular criteria was given a weight of 30 per cent.

18           Second criteria was cost, a comparison both of ten  
19 year cost and life cycle cost, and that cost was to include  
20 competing technology development, design, construction,  
21 start-up, operations of the new facility that I just  
22 mentioned a moment ago, as well as projection of the waste  
23 disposal costs which obviously are related to the number of  
24 canisters and the volume of material to be disposed of. This  
25 was also given a weight of 30 per cent.

1           Third criteria were the technical merits. Each  
2 technology was judged with respect to its conformance with  
3 environmental safety and health standards, waste form  
4 compatibility with repository requirements, again, as well as  
5 they were defined almost two years ago, and some  
6 consideration of potential proliferation issues. And that  
7 was given a weight of 20 per cent.

8           And last, but certainly not least, in DOE's  
9 priority was timeliness of implementation of the operation  
10 and the beginning of transitioning out of wet storage into  
11 dry storage. That was also given a weight of 20 per cent.

12           The next slide in your hand-out is a table that  
13 shows the results of their evaluations and the scoring that  
14 they came up with, and that's been about two years. Since  
15 that time, some of the input and some of the information they  
16 had available has evolved. We have a better understanding of  
17 the costs than we then had, a better understanding of some of  
18 the performance requirements. And so today, I don't imagine  
19 the scoring would come out the same exactly, but they did  
20 perform and we have looked back at their criteria, and in the  
21 end, we think that their conclusions still stand relative  
22 among the technologies.

23           BULLEN: Bullen, Board, with a quick question. Are you  
24 going to put up the table?

25           BARLOW: I can if you'd like to talk about it.

1           BULLEN:  Actually, just a quick question to ask about  
2 that.  I notice that in picking your weighing, and everyone  
3 is always going to criticize how you pick your numbers, and  
4 this is not a criticism, this is just a comment, that if  
5 costs weren't weighted so highly, if you switched cost and  
6 technical suitability as 30 per cent, 20 per cent, the other  
7 way around, would you expect a significant change?  I mean,  
8 the cost heavily weights toward direct co-disposal, according  
9 to your inventory there.  If it were only 20 per cent and the  
10 technical suitability of melt/dilute or press/dilute or  
11 whatever, which has a higher technical--and obviously we're  
12 the technical review board, which is why we look at those  
13 things--do you think it would change?  And particularly in  
14 your second choice, because if you look at the overall score,  
15 the difference between 66, 63, 62 and 60 isn't very big.

16           BARLOW:  That's right.

17           BULLEN:  And so were costs not so heavily weighted and  
18 technical suitability given a little bit more weight, would  
19 that skew the results a little bit differently?

20           BARLOW:  Well, as I said, they did do a sensitivity  
21 analysis, and I didn't prepare that graph.  The way they did  
22 the sensitivity analysis was to take out, or zero out one  
23 category, and then determine the ranking.  I don't recall  
24 exactly how the sensitivity result came out when the cost was  
25 zeroed out.  But I think I would agree with your perception

1 that most likely, at least between these two highest scores,  
2 they would come much closer, if not switch. And, you're  
3 right, because of cost and timeliness, the direct co-disposal  
4 and direct disposal came out a little bit higher than the  
5 others.

6       BULLEN: I think from my personal perspective, I'd like  
7 to see technical suitability be more heavily emphasized as  
8 opposed to cost, because I'm not sure we have as firm a  
9 handle on cost right now as we're going to have in the time  
10 frame for disposal. And so cost usually is one of those  
11 things that drives things early on, and maybe falsely so, and  
12 so you really worry about giving a lot of emphasis to cost.  
13 I mean, fully 30 per cent to cost is a strong emphasis there,  
14 whereas I think from a technical perspective, we'd like to  
15 see more weighing in that area.

16       BARLOW: An evaluation similar to this, but our  
17 intention is to conduct it a little bit differently, an  
18 evaluation similar to this will need to be done to narrow  
19 down between those that we are now implementing or  
20 developing, which are the direct co-disposal and melt and  
21 dilute. That evaluation may or may not have the same kind of  
22 weighing.

23               I would also mention that most recent cost  
24 estimates, in fact there was one just published last Friday  
25 and, among other groups who were interested, the National

1 Academy of Science was provided a copy, show that the  
2 difference in cost between direct co-disposal and melt and  
3 dilute is essentially zero, at least within the range of  
4 accuracy. So even with that weight, the scoring would  
5 change.

6 DI BELLA: While you've got the table up there, this is  
7 Carl Di Bella, I'd like to make--I'm not sure if this is  
8 going to come out as a comment or a question--the ultimate  
9 objection here is to get rid of this stuff, to dispose of it  
10 permanently, and it would seem to me that your criteria ought  
11 to include some sort of performance metric so that decision  
12 makers could look at the trade-offs between the various  
13 factors and performance. It appears right now all you have  
14 is like meets a minimum performance standard. I'm talking  
15 about performance in a repository meaning something like the  
16 dose or something of that sort. Am I correct in interpreting  
17 your table that you really don't have a performance goodness  
18 metric there?

19 BARLOW: Again, what I'm referring to here is an  
20 evaluation that was done a couple of years ago. And in the  
21 category called technical suitability, there was an attempt  
22 to evaluate the performance against waste acceptance criteria  
23 as it was understood. And since I was not directly involved  
24 in it, there may be someone who would correct me in the  
25 audience, but my understanding of it is that there wasn't

1 much of an attempt to give too much weight to going beyond  
2 acceptable. In other words, there wasn't a good, better,  
3 best grading that played very heavily into that evaluation.  
4 That, of course, is very subjective and it's somewhat  
5 difficult to quantify, but our intention in the evaluation  
6 that I'm going to describe in a few moments that will be  
7 happening here in about six months would be to give a little  
8 bit broader look to that consideration. And your input and  
9 the National Academy's input would be certainly valuable to  
10 that, among other stakeholders.

11           As the table showing the scores indicated, the  
12 simpler or less complex options generally came out ahead of  
13 the more complex technologies. And those that came out near  
14 the top, as we just mentioned, were direct co-disposal and  
15 melt or press and dilute. We talked about because of cost  
16 and timeliness, that direct co-disposal tended to be a little  
17 bit higher, but taking that out of consideration and looking  
18 at how you would rate melt and dilute, its advantages and  
19 technical suitability, it came out higher in that particular  
20 category.

21           Electrometallurgical treatment was acknowledged by  
22 the task team as an ongoing development activity that had its  
23 own funding, and ought to be kept in the wings, as it were,  
24 as a diverse, more advanced technical backup. And, in fact,  
25 part of the assessment or presentation and deliberations that

1 we had with the National Academy of Sciences two weeks ago  
2 included the description of ongoing effort related to  
3 aluminum fuel and electrometallurgical treatment.

4           However, within the Savannah River site and the  
5 program that I'm managing, we are not actively involved in  
6 that particular technology, but we are with direct co-  
7 disposal and melt and dilute. We have not continued the  
8 development of press and dilute primarily because, again as  
9 was pointed out on that graphic, press and dilute  
10 accomplishes some but not all of the, if you will, technical  
11 advantages of melt and dilute, and in our evaluation, the  
12 melt and dilute does not pose that much additional cost or  
13 technical complexity as compared to melt and dilute. So we  
14 are not pursuing that option at this time.

15           Subsequent to the Research Reactor Task Team  
16 completing its work and publishing its report, which is  
17 documented in two volumes, the first of which is a volume  
18 that describes just what I did in much more detail, and a  
19 second volume of appendices that provides primarily cost, but  
20 also some critical analysis and some other data that backs up  
21 this report. And if the Board does not have it, certainly we  
22 can provide a copy.

23           Shortly after this report was published, the  
24 Department of Energy directed Westinghouse Savannah River  
25 Company to implement the recommendations that are contained

1 within that program, and that's why I'm here. For about a  
2 year and a half now, we've been conducting the evaluations  
3 both on the technology side, and we've begun the engineering  
4 and project work for the facility. And though describing  
5 that project activity is not part of my discussion this  
6 morning, I'm available to answer questions about where we are  
7 with that.

8           But what we will spend some time on, and I'll  
9 describe here briefly, is the technology program, and later I  
10 guess the next speaker, Dr. Iyer, will go into more depth  
11 than I will here. Generally, I would describe the goals of  
12 our program to be three-fold; first, to conduct and  
13 ultimately complete waste form qualification studies on both  
14 the direct co-disposal form as well as a melt and dilute  
15 product form. Those studies are being done for the  
16 environment, both for dry storage in an interim basis at  
17 Savannah River, and for repository disposal, such that,  
18 again, we're looking at those environments and the variations  
19 associated with those.

20           These studies and the technical basis, which I'll  
21 mention in a moment, will become the input to a technology  
22 decision which we are now scheduled to make in the fall of  
23 next year. In the summer of next year, Dr. Iyer and our team  
24 will be preparing a report that will be an assessment of  
25 these technologies, these two alternatives, and provide a

1 recommendation to the Department as to which one, or perhaps  
2 even a combination of the two, we would recommend for  
3 implementation.

4           The technical data package of course is the  
5 necessary documentation that needs to be provided, both as a  
6 basis for design and construction of a facility at Savannah  
7 River, but likewise for the design, the environment impact  
8 study and the licensing of the repository.

9           Discussions have been going on and have increased  
10 in frequency, and sometimes intensity, between ourselves and  
11 DOE RW in sharing documentation and in face to face meetings  
12 on what this data package needs to consist of. And in fact,  
13 yesterday all day, and I think all day today, we have  
14 engineering and technical folks meeting with DOE RW  
15 discussing what the data needs are and how that data is going  
16 to be used so that we can be sure that the information that  
17 we're generating is compatible with their needs.

18           And thirdly, as I mentioned a moment ago, we are  
19 defining those functional requirements necessary to build the  
20 facility that may or may not include treatment, but certainly  
21 would include certain conditioning steps to prepare the fuel  
22 not only for dry storage, but ultimately for disposal.

23           In this past little over twelve months, we have  
24 made excellent progress, we think, and as I mentioned, not to  
25 steal Dr. Iyer's thunder too much, let me just hit some of

1 the highlights of what we have accomplished in the technology  
2 side of our program.

3           Within the direct co-disposal arena, we have  
4 documented a set of specifications for drying of the aluminum  
5 fuel, and this is based upon vacuum drying tests that we've  
6 conducted at the site. We have also, both within the site  
7 organization and in conjunction with other DOE laboratories,  
8 and DOE RW's M&O, have conducted a number of performance  
9 studies that include thermal analysis, air and vapor  
10 corrosion testing and criticality analysis, which will be  
11 part of the topic following Dr. Iyer.

12           We've also installed, or are in the process of  
13 installing a test canister to be used out in the facility  
14 with real irradiated spent fuel for the purposes of  
15 monitoring and validating its performance against the models  
16 that we have been creating and are refining as we go through  
17 our qualification studies. That test canister should be in  
18 place by March of this coming year and be ready to have fuel  
19 installed into it.

20           Within the melt and dilute area, most of the focus  
21 has been in the process development side of things. That  
22 includes looking at the various crucible materials and the  
23 performance of those materials during the melting process.  
24 We have been looking at furnace design, and one of the  
25 primary concerns or considerations for moving forward with

1 this technology is how to deal with the by products that come  
2 as a result of melting these spent fuel elements in a  
3 furnace. And so the off gas system is an important component  
4 of that process that we are spending a lot of effort  
5 addressing.

6 All of this work has been documented in reports,  
7 many of which have been provided to the National Academy of  
8 Science and to the U. S. Nuclear Regulatory Commission for  
9 their information, as well as the Department of Energy, and  
10 they're available to the Board at your request.

11 Looking ahead at least for the next six to twelve  
12 months as we anticipate this technology decision, our focus  
13 on direct disposal is really to do more expanded and detailed  
14 analyses than those that we've been able to accomplish in the  
15 past several months. And likewise, within the melt and  
16 dilute process, we envision, or we have committed to and have  
17 schedules for starting up prototype facilities and apparatus  
18 within the laboratory to demonstrate the process works, and  
19 we will do that to form various compositions of alloys. And,  
20 again, Dr. Iyer will discuss that for you in a little bit  
21 more detail in a few minutes.

22 I'd like to switch gears now at the Board's request  
23 and take just a few minutes and describe the activities and  
24 the dialogue that are taking place now with the U. S. Nuclear  
25 Regulatory Commission.

1           A few months ago, the Department of Energy and the  
2 USNRC agreed to an exchange of technical information  
3 regarding this program, and specifically what DOE has  
4 requested is that the NRC staff, technical staff, provide  
5 comments regarding the suitability of the various  
6 technologies as it pertains to repository acceptance, and  
7 some of the issues that may relate or pertain to that.

8           Since this memorandum was signed, we have provided,  
9 as I mentioned a moment ago, to the NRC the documents, some  
10 of the primary documents that describe the work that's been  
11 completed and the work that's planned. Tomorrow, we will be  
12 having really the first working meeting that we've had with  
13 the NRC to make plans for activities between now and at least  
14 six, twelve months from now. So we do not yet have feedback  
15 from the NRC that I can relate to you today.

16           In fact, if they haven't introduced themselves, I  
17 think that at least two of the folks that we'll be meeting  
18 with from the NRC are here this morning.

19           Finally, as I mentioned, and that you're aware, the  
20 National Academy of Sciences has been asked to assist DOE by  
21 conducting a review of this program, and that has begun.  
22 Specifically within the National Research Council's Board on  
23 Radioactive Waste Management, a principal investigator has  
24 been assigned, and a panel of experts has been assembled to  
25 conduct a review in anticipation of this upcoming decision

1 that needs to be made.

2           The National Academy of Sciences has met with us  
3 twice. First, just a small group with the principal  
4 investigator, Milt Levenson, and their study director, Kevin  
5 Crowley, for the purpose really of kind of scoping out the  
6 effort, understanding what information is available.

7           Then the second meeting took place two weeks ago in  
8 this building down the hall, and it was a day and a half of--  
9 really a full day of presentations made by myself and my  
10 staff, as well as individuals from DOE RW's M&O conducted a  
11 number of analyses and studies that we are using and will be  
12 described for you a little bit later. Their schedule is to  
13 provide DOE with a report in March. It's a very compressed  
14 time frame by their standards, and they've expressed that a  
15 number of times, that concern, but they have asked their  
16 panel of experts to provide their written input by Christmas,  
17 and presuming that they're all going to do that, they are  
18 committed to providing the Department with a report. That  
19 report and its recommendations, its comments and observations  
20 will certainly be factored into our assessment and ultimately  
21 the Department's decision.

22           That concludes my presentation, and I thank you for  
23 the opportunity to present it.

24           BULLEN: Thank you, Mark. This is Bullen, Board.

25           As a followup to your last viewgraph, I've been in

1 contact with Dr. Levenson, and have agreed to provide him  
2 everything that we learn in this meeting to further augment  
3 his studies, and he has also agreed to provide us with his  
4 input, actually his drafts as they come out. So we are very  
5 interested and are very aware of this study and look forward  
6 to seeing it.

7           Any questions for Mark from the panel?

8           (No response.)

9           BULLEN: Okay, I'm going to defer questions from the  
10 audience in light of keeping us on schedule. We will take a  
11 break for how about 13 minutes, and reconvene at 10:45.

12           (Whereupon, a brief recess was taken.)

13           BULLEN: Could we reconvene, please, have at least the  
14 Board members come back and have their seat at the table?  
15 And could everyone grab their coffee and pull up a chair?

16           Our next speaker is Dr. Natraj Iyer. He manages  
17 the technical activities at the Savannah River Site Spent  
18 Nuclear Facility Technology Development Program, which  
19 includes aluminum spent nuclear fuel, alternative treatment  
20 technology. He also manages the Materials Application and  
21 Corrosion Technology group at the Savannah River Technology  
22 Center. And Dr. Iyer will speak to us about the treatment  
23 options for aluminum clad, highly enriched uranium spent  
24 nuclear fuel disposal.

25           IYER: Good morning. I want to thank the Board for this

1 opportunity to talk about the alternate treatment technology  
2 program that's currently underway at SRS. As Dr. Bullen  
3 mentioned, my name is Natraj Iyer and I'm from the Technology  
4 Center at SRS, and I'm going to be talking to the activities  
5 that are currently underway in the context of the alternate  
6 technology program.

7           What I'm going to do is give kind of a background.  
8 I know Mark has covered some of the aspects of aluminum fuel  
9 and its characteristics, and what I'm going to try to do is  
10 try to go into some more detail and give some background on  
11 aluminum fuel, show you a little bit about its corrosion  
12 performance in basin storage or wet storage, which could be  
13 one of the conditions the fuel is in before it's put into a  
14 road ready package either for direct disposal or melt/dilute,  
15 also show what the corrosion performance is in dry storage.  
16 Again, those are kind of the starting conditions. And then  
17 really share with you the road map we have for the technical  
18 activities related to the options that we are currently  
19 pursuing.

20           I should point out that the charter that was given  
21 to us as part of this program was to look at both these  
22 options. As the task team looked at all the different  
23 options, we were not necessarily champions or defectors for  
24 any of the options. So what we hope to do and attempt to do  
25 is look at both these options objectively in the context of

1 the decision drivers that our leaders have laid out, and see  
2 how things play out.

3           As most of you know, there are really three  
4 different kinds of aluminum fuel that are being considered as  
5 part of the MTR aluminum, the foreign research and domestic  
6 research reactor. Primarily, the UAlx type and what that  
7 consists of, the UAlx phase, the aluminite phase and aluminum  
8 matrix, and it's primarily made--some are made by casting  
9 extrusion technology, but most of it's made by primarily the  
10 rolling or casting technology.

11           Another kind of fuel is the U3O8 element of fuel,  
12 where again you have a U3O8 phase and an element of matrix.  
13 And then the low end rich fuel, a lot of it is the silicite  
14 fuel, which is kind of the more recent fuel that came out of  
15 the RRTR program, the Reduced Research and Test Reactor  
16 Program, and what this consists of is primarily the silicite  
17 phase, again in an aluminum matrix. As I said, typically  
18 most of these fuels are made by power metallurgy followed by  
19 some kind of metal working, be it rolling or extrusion. Some  
20 of it is made by casting, but that's a fairly small fraction.

21           Obviously what we are more interested in is what do  
22 these fuel microstructures look like after irradiation. And  
23 what you see for all the three types of fuel, obviously  
24 you're going to see the fission cavities or the fission  
25 porosities in these fuels, and both in the UAlx fuel and the

1 U3O8 fuel, a lot of it is converted to the UAlx phase. So  
2 what you see in the oxide fuel, for example, is, depending on  
3 the burn-up, a lot of UAlx phase, a little bit of oxide, and  
4 then primarily a aluminum matrix. And, again, with the  
5 silicite fuel, it's primarily silicite and aluminum.

6           So if you're talking either dry storage, direct  
7 disposal, this is kind of the condition of the fuel that  
8 we're looking at as we get it ready to be packaged.

9           I just wanted to share with you a little bit about  
10 corrosion performance. There has been a lot of discussion,  
11 both in the technical community and in the operations area in  
12 terms of performance of aluminum fuel. The fact is aluminum  
13 is reactive, but what's unique about aluminum is it's  
14 reactive, so it forms the initial oxide layer very quickly.  
15 And once it forms that oxide layer, it passivates very  
16 readily. So in a sense, as long as the integrity of the  
17 oxide layer is maintained and the environment is maintained,  
18 aluminum fuel is very corrosion resistant, and that's been  
19 borne out again and again in a lot of our field experiences,  
20 both in wet storage and dry storage.

21           Typically, what happens with most of these fuels is  
22 in reactor service, you very rapidly for a typical  
23 irradiation, anywhere from 50 to 70 per cent burn-up, you  
24 build up a bromide layers of about 50 microns, and as long as  
25 the handling is typical, that is, you don't initiate

1 scratches on the fuel, the bromide layer is very resistant to  
2 any kind of corrosion. And so when we take this into the  
3 basins, and we looked at both basins, and by basins of  
4 origin, I mean basins across the world in a lot of foreign  
5 countries which are not necessarily either technically or  
6 economically built up to the extent the U. S. is, so we  
7 looked at the fuel in the basins of origin and then of course  
8 the basins at SRS.

9           What you find is typically unless you breach the  
10 oxide surface somehow, you see very good corrosion  
11 performance for long periods of time. Over the last three  
12 years, we have had an extensive foreign research reactor off-  
13 site inspection program, where our folks have gone and  
14 actually visually looked at fuel and fuel plates in a lot of  
15 these foreign countries, Brazil, Venezuela, Uruguay, and a  
16 lot of those other countries, and what we have found is what  
17 you see on the right is kind of typical of over 95 per cent  
18 of the fuel that we see as what is categorized as FRR fuel,  
19 and that we expect back at SRS. That is, it's in near  
20 pristine condition. Between 2 to 5 per cent of the fuel has  
21 some degree of corrosion, and what I've shown on the left-  
22 hand side there is the absolute worst fuel that we've seen so  
23 far in our inspection of a lot of different foreign basins.  
24 And even that particular fuel where you see these oxide  
25 build-up, or the corrosion product build-up, that particular

1 fuel, if you stick it in the basin water by itself and do a  
2 leach test, or what we call a SIP test, which is one of the  
3 accepted tests, the leaching of cesium 137 is almost non-  
4 detectable. It's less than about 8 or 9 nanocuries per hour,  
5 but it kind of gets into the range of sensitive--and things  
6 like that. But it's almost non-detectable. And as I said,  
7 that is the absolute worst case fuel that we've seen.

8           A big difference between the leaching  
9 characteristics in aqueous environment, and I'm talking about  
10 a typical basin aqueous environment as opposed to the  
11 repository environment. What happens is even though you  
12 initiate a pit, which is a primary form of degradation, it  
13 passivates very quickly and you form an oxide layer over the  
14 pit. So typically unlike commercial fuel, where if you have  
15 a breach, it pretty much releases all the fission products  
16 because of the interconnected porosity, in this case, the  
17 release is really driven by diffusional processes, which are  
18 time, temperature, diffusion coefficient driven, because all  
19 the fission products are really attached in the core.

20           So that's the big difference between alloy fuels  
21 and what we see in commercial fuels, and because of that, in  
22 typical basin aqueous environments, we don't really see--or I  
23 would say see insignificant leaching, where we get into the  
24 so-called non-detectable or near non-detectable uncertainty  
25 range. But that again only goes to the 5 per cent of the

1 total inventory or typical FRR fuel that we'd be receiving  
2 back is more what you see on the right side.

3           That particular fuel, for example, was stored in  
4 wet storage for 25 years in Brazil, and a wide range of  
5 chemistries over its life storage history, and that's typical  
6 of a lot of the foreign research reactor basins.

7           At SRS, most of our basins are maintained to 1 to 2  
8 micromole semens per centimeter, if that's the right unit.  
9 So most of our basins maintain very pure conditions, and  
10 again, unless you see a breach of the oxide layer itself, we  
11 haven't seen any corrosion.

12           We have an extensive corrosion surveillence coupon  
13 program where we periodically remove corrosion samples and  
14 actually also look at our fuels visually to see if corrosion  
15 has been initiated, and if it has proceeded.

16           I didn't bring the slide on dry storage, but then  
17 talked a little bit about dry storage. We've done extensive  
18 work, which could be another adverse condition for some of  
19 these fuels. What we've seen also in our foreign inspection  
20 is that dry storage under a wide range of conditions, from  
21 fairly humid conditions in Brazil where they actually had  
22 some water leak on some of these fuels, to controlled  
23 conditions, I would say more controlled conditions, again the  
24 condition of the fuel that we've seen in dry storage under  
25 wide ranging conditions has been fairly pristine, as long as

1 the oxide layer hasn't been breached.

2           Now, within SRS, we have done extensive work on dry  
3 storage to basically develop the storage criteria. But what  
4 I was trying to point out was really the range of experience  
5 that we observed in receiving these fuels from the different  
6 basins across the world.

7           I'm going to switch gears now, having described the  
8 condition of the fuel that we're going to be receiving for  
9 some kind of alternative treatment, I'm going to talk a  
10 little bit about the program that Mark alluded to. It's  
11 referred to as the alternate technology program, and it was  
12 initiated in FY97, so we are a little over a year underway.  
13 And the purpose of this program, our charter was really  
14 implement the recommendations of the Research Reactor Task  
15 Team that DOE headquarters had instituted, and the primary  
16 recommendation was to pursue direct co-disposal and as a  
17 backup, melt/dilute.

18           As the program got started, our approach has been  
19 we are pursuing direct co-disposal and melt/dilute on equal  
20 par in terms of the way we are approaching the issues, and as  
21 I said, we're looking at it in the context of the number of  
22 decision drivers, and just a year from now, we're going to  
23 see how things play out.

24           What direct or co-disposal is is primarily putting  
25 a spent fuel canister in a waste package which consists of

1 DWPF glass canisters. And the melt/dilute option is taking  
2 the fuel, melting it, adding depleted uranium to dilute the  
3 enrichment, solidifying the fuel directly in a canister, and  
4 thus flexibility in terms of what that canister material  
5 could be. For example, it could be in a canister, or we  
6 could remove it from the canister and put it in the spent  
7 fuel canister. But one option is, for example, if titanium  
8 alloy or some other hastelloy is a good engineered barrier,  
9 then the option does exist to cast it in a crucible of that  
10 material, seal it, and then put it in the spent fuel  
11 canister. So it does provide a lot of flexibility in terms  
12 of how we would process it.

13           The primary drivers for direct disposal, as you saw  
14 from the report, was both the cost and the timeliness, and of  
15 course the issues and challenges are to demonstrate  
16 criticality control and demonstrate that to the satisfaction  
17 of all the stakeholders. That was one of the primary  
18 challenges. The other big challenge is the characterization  
19 requirements.

20           I failed to mention that as these fuels come from  
21 across the world, they have different pedigrees in terms of  
22 the way they've been characterized or the way the  
23 characterization data has been assimilated in all these  
24 different countries. And so the question is how do you  
25 reconcile that versus what the RW requirements may be for

1 characterizations. So the characterization requirements  
2 could be quite significant for direct disposal. That's still  
3 something, you know, will evolve through discussions, but it  
4 could be significant given that they all have different  
5 pedigrees as they come from across the world.

6           The dilution option has the benefit that we  
7 basically erase the bulk of the history of the fuel, so the  
8 characterization requirements are fairly limited in context  
9 of direct disposal. You do have the benefit of the dilution,  
10 which helps with the criticality control.

11           And, finally, as we have done the work, and you'll  
12 see later, it does have a significant volume reduction  
13 potential, up to 70 per cent, and I'm going to get into that  
14 a little later.

15           So the way the program was laid out is our primary  
16 decision drivers were the spent fuel performance, and by that  
17 we mean both performance preclosure and postclosure in the  
18 repository, the characterization requirements, costs,  
19 schedule, licensability, stakeholder acceptance. And when  
20 you look at those decision drivers in terms of what needs to  
21 be done technically, we basically can categorize in three  
22 boxes. One is the spent fuel form development, and I'll get  
23 into what that means a little later. The other one is  
24 performance, and then another issue unique to alloy fuels is  
25 the fact that you need to be very careful how you interpret

1 the performance results. And so you need test protocols.  
2 For glass, there's an ASTM standard, which is commonly used  
3 for commercial fuels. Again, there are standard techniques  
4 that have been used. No such technique exists for aluminum  
5 alloy fuel. And so that particular activity is just as  
6 important if you're going to get good scientific data which  
7 can be input into the PA, and other data needs requirements.

8           So the way the program was organized for both these  
9 options was to look at these three major issues so that we  
10 could provide the information both to our management and all  
11 the stakeholders in terms of all the decision attributes.

12           Just getting into it in a little more detail, what  
13 we mean by spent fuel form development is for direct  
14 disposal, it's really coming up with the road ready storage  
15 criteria in the context, that is, what is the--we already had  
16 a dry storage criteria for aluminum fuel established here at  
17 SRS, and what we are doing there is building that and taking  
18 it to the next step in the context of what we know currently  
19 is the waste acceptance criteria, and to come up with what we  
20 call the road ready package criteria for aluminum fuel. So  
21 that's really what it means in the context of direct and co-  
22 disposal.

23           On the other hand, in the context of melt/dilute  
24 process, it means what's the right composition of your  
25 melt/dilute form. What is the dilution level, what's the

1 uranium aluminum ratio, and what is the result in  
2 microstructure and what are the characteristics. So it's  
3 really the bench scale process development leading to some  
4 kind of what we call the optimum microstructure to drive the  
5 performance.

6           The performance in both cases, in direct disposal,  
7 it's really performance of the fuel microstructure. I showed  
8 you earlier the irradiated microstructure. In the context of  
9 melt/dilute, it's really performance where you don't really  
10 have those fission pores. It's a fairly homogeneous  
11 microstructure with uranium-aluminum ratios. So it's really  
12 looking at performance of those kind of microstructures.

13           And the reason we are doing this is really to feed  
14 two critical data needs. One is to feed the process  
15 requirements so that the SRS site can start developing  
16 functional requirements and start planning towards the  
17 feasibility, and the other one is to address the data needs  
18 that RW has as they get into PAs. Right now, most of the  
19 information, for example, that's being used in the  
20 preliminary performance assessments are assumptions of  
21 aluminum dissolution rates and pure uranium dissolution  
22 rates. And as this program evolves and as we start  
23 generating dissolution rates for the exact microstructures,  
24 those will be fed into RW.

25           And then the last block at the bottom, the test

1 protocols, is the activity I mentioned earlier, is to make  
2 sure that the way we measure the performance characteristics  
3 and the way we interpret it is consistent and has consensus  
4 to ASTM or some other national consensus board.

5           What I'm going to do next is kind of show you a  
6 road map of the technical program for both of these options,  
7 and then just highlight some of the accomplishments. My  
8 intent here is to just give you an idea of the kind of  
9 activities, and then as I said earlier, tomorrow as we go  
10 through the tour, if you'd like more details on any one  
11 specific activity or all of them, we'd be glad to provide it  
12 as we walk through the tour, since I can't condense  
13 everything in a half an hour.

14           But the direct disposal area, as I said, there are  
15 several activities, and those are these major blocks leading  
16 to basically what we call the technical basis for direct  
17 disposal. That is how that body of information that would  
18 help the decision makers make the right technical decision.  
19 The first block is the spent fuel form definition, the first  
20 two blocks, and the form development and development of the  
21 storage criteria. What that is is basically building on the  
22 dry storage criteria, looking at the environmental conditions  
23 that are required in the road ready package canister so that  
24 you basically don't have degradation preclosure or very  
25 limited degradation within acceptable limits up to the

1 preclosure stage. And that has been defined and we have  
2 documented that, at least the preliminary road ready storage  
3 criteria.

4           The next block is basically a better definition of  
5 the road ready package. In this context, we are primarily  
6 working with the national program as it defines the various  
7 canister configurations.

8           The performance analysis, a big aspect of direct  
9 and co-disposal, is obviously the criticality analysis, since  
10 we are dealing with high enriched uranium, and the next  
11 speaker is going to talk to that, but we're working with RW  
12 M&Os as they perform the early criticality analysis for some  
13 of these fuel types to see where we stand, just so that again  
14 we get the body of information that we need to make a wise  
15 and right decision.

16           One major aspect of criticality performance  
17 obviously with high enriched uranium is the response and the  
18 reconfiguration of materials. That's just as important as  
19 just normal degradation. How do they reconfigure and  
20 reconstitute through the geological time. So that is a major  
21 activity.

22           Obviously, a lot of the criticality analysis is  
23 already underway, and what we hope to do with the information  
24 we generate is go back and either validate the assumptions or  
25 fine tune the assumptions, and that's the intent.

1           And at the end, what we expect to have is for each  
2 of the attributes in the waste acceptance criteria, basically  
3 have a technical basis, for example, for reactivity,  
4 pyrophoricity, compatibility, the dissolution rates, and have  
5 that package of information for both these options, in this  
6 case, I'm showing it for direct disposal, so that come next  
7 fall, we would have the same set of information for both  
8 these options in the context of repository performance, and  
9 at the same time, also have the functional requirements for  
10 both these options so that we can make the decision.

11           What I have in these blocks on the side are  
12 primarily some of the specific activities that have been  
13 going on in fiscal '97 and what's currently underway in  
14 fiscal '98. A lot of the activities in fiscal '97 focused on  
15 the definition of the road ready storage criteria, and what  
16 we have currently underway, as Mark said, is basically a lead  
17 surveillance program that we're trying to go validate the  
18 corrosion models that were developed for the road ready  
19 package.

20           The focus of the '98 activities is really to look  
21 at the materials response reconfiguration issue, and also  
22 start generating the information in the repository  
23 environment, and those tests are underway so that we can get  
24 some realistic dissolution rates. And I'm going to get to  
25 that a little later.

1           The next slide I'm just going to highlight. This  
2 by no means is a summary of all the activities. But I'm just  
3 going to highlight some of the activities that had been  
4 underway. The first block on the left is the road ready  
5 package. As I said, we have established the drying criteria  
6 and the backfill criteria, and we're actually conducting  
7 field vacuum drying tests for these MTR assemblies in the  
8 canister to make sure we have enough information to issue the  
9 drying specs so that we would have reasonable functional  
10 requirements for the storage facility.

11           At the same time, we also have what we call--where  
12 we initiate the lead surveillance canister, and the purpose  
13 of that is really to validate our storage criteria and our  
14 corrosion models, our degradation models. And what we are  
15 doing in there is we have a highly instrumented canister,  
16 which is shown in the photograph on the second block on the  
17 right. We have a highly instrumented canister which has all  
18 the bells and whistles in terms of temperature measurement,  
19 gas pressure measurement, everything else, so that we can  
20 validate both our thermal models and our degradation models  
21 as we take a canister to the extremes of the safety and  
22 operating envelope.

23           We have developed degradation models, and these  
24 have been done parametrically initially using typical cold  
25 samples, including gamma radiolysis effect, and then the

1 models are being validated actually using hot samples. So  
2 that's the approach we've taken, is we have done a very  
3 extensive parametric analysis using cold samples in gamma  
4 radiation with radiolysis effect, and then we go and pick key  
5 spots and then do a validation using hot samples. That's the  
6 approach we've been using, and found it to be very cost  
7 effective.

8           And the last block on criticality analysis, the  
9 next speaker is going to talk to that, so I'm not going to  
10 say anything at this time.

11           I'm going to switch gears and talk a little bit  
12 about the melt/dilution process. And as I said, you'll see a  
13 lot more detail tomorrow as you walk through the lab  
14 specifically on this process, because we will show you our  
15 apparatus and what we've been doing. But the key blocks  
16 again are development of the melt/dilute form, and by that,  
17 what I mean is to make sure, number one, we know we can cast  
18 these alloys because that's how we made the fuel at SRS for a  
19 number of years, but make sure that we can get the right kind  
20 of microstructures to make sure we understand what those  
21 microstructures mean in terms of performance, and also make  
22 sure some of the process issues in terms of crucible mold  
23 interactions and the various options we have, for example, do  
24 we pour it, do we cast in crucible. If you cast in crucible,  
25 does it make sense to cast in a carbon steel crucible, or is

1 there something to be gained casting it in an engineered  
2 barrier kind of material and sealing it. So those are the  
3 kinds of issues we are looking at.

4           The other major issue is the fission product  
5 release, and this is in the context of the functional  
6 requirements for a feasibility at SRS. So we have done  
7 extensive analysis of what the total fission inventory is for  
8 all the 20,000 elements coming back, what would be the  
9 inventory, how would we process it through a melt/dilute  
10 furnace, and what the primary actors are and what kind of off  
11 gas system do we need to treat those.

12           Then the next block is small scale validation. A  
13 lot of the up-front work was doing bench scale. And by bench  
14 scale, I mean hockey puck kind of samples. And what the  
15 small scale validation is aiming to do is primarily taking  
16 full-scale MTRs and taking it through a melt/dilute process  
17 to show and convince ourselves that we are making a  
18 homogeneous product. And where that stands is during FY98--  
19 well, during FY97, we actually did a full-scale MTR  
20 melt/dilute experiments, and during FY98, we actually are  
21 developing a facility which would be more prototypic, that  
22 is, switching from a resistance furnace to an induction  
23 furnace with induction stirring so that we have a more  
24 homogeneous microstructure. And the plan is then to take  
25 that to FY99, actually do a--

1           And then the last block is the form assessment,  
2 which is looking at the waste acceptance criteria attributes,  
3 such as reactivity, pyrophoricity, corrosion resistance, gas  
4 generation, et cetera, and make sure we have a technical  
5 basis for those, and develop that. And that obviously, a lot  
6 of it is common to direct disposal, except for the  
7 microstructure dependents.

8           BULLEN: Bullen, Board with a quick question before you  
9 leave this. As you've done the melt/dilute, then you don't  
10 have to worry about the criticality analysis or the  
11 criticality--

12          IYER: No, one of the activities, and I don't know if  
13 it's in that block, but one of the activities we have  
14 underway in FY98 is the criticality analysis.

15          BULLEN: Okay. It's in the far right block way at the  
16 bottom, yeah. I see it now.

17          IYER: Okay. And that's currently underway, at least  
18 the scoping analysis is currently underway in-house, and then  
19 taking it to the next step through the RW/M&O, we haven't  
20 resorted to that extent yet.

21          BULLEN: So it's sort of dependent upon the geometry of  
22 the package as you put it in, and then how it reconstitutes  
23 again from the waste form as it dissolves?

24          IYER: Right.

25          BULLEN: And I guess that's the follow-on question which

1 you're trying to address.

2 IYER: Yes. This viewgraph just highlights again some  
3 of the accomplishments in melt/dilute. I've kind of spoken  
4 to a lot of that. The extreme left block, again just shows  
5 you the kind of variables we looked at in our bench scale  
6 work, and most of them are process issues as opposed to  
7 waste--well, we have looked at waste form microstructure  
8 dependency on those process issue.

9 What the right block shows is basically the volume  
10 reduction potential. The typical nominal numbers that we've  
11 been using for all our analysis is about 1,400 canisters for  
12 direct co-disposal. And in that context, if you go to 20 per  
13 cent dilute, depending on the alloy composition, you  
14 typically get in the 400 ballpark range, 400 canisters  
15 ballpark range for the melt/dilute process.

16 The block at the left bottom kind of shows just a  
17 schematic of the off gas system. And I'm going to get to the  
18 radionuclide inventory in a bit, but just shows that you have  
19 to worry about the cesium, iodine and krypton, they are major  
20 players, and we currently have detailed work going on both  
21 bench scale and then taking it to full scale in terms of what  
22 that off gas system would look like.

23 And the extreme right block at the bottom is just a  
24 view of the furnace that we have, which you'll see tomorrow  
25 when we do the melt demonstration, melt/dilute demonstration

1 of a full scale MTR.

2           I'll talk a little bit about form testing because  
3 as I said earlier, this is just as critical for these alloy  
4 forms as the rest of the work. And the main reason is for  
5 the glass waste form, as I said, there's been a lot of work  
6 done in the aqueous environment, and also for commercial  
7 waste form.

8           SAGYIS: Excuse me. Sagηgs, Board.

9           Are we now back into the initial form, or--

10          IYER: This is common--

11          SAGYIS: Not the melt/dilute form any more?

12          IYER: Okay, in terms of this particular activity, what  
13 we are trying to do is make sure we understand as a function  
14 of the aluminum uranium ratio, for example, in the  
15 microstructure, how that affects performance. So this  
16 particular activity is common to both forms, and one  
17 particular point, for example, the typical nominal 19 per  
18 cent uranium alloy, which is a standard MTR, that would be  
19 the point which applies to direct co-disposal. Some of the  
20 other compositions would apply to the melt/dilute. So it's  
21 common.

22          SAGYIS: Thank you.

23          IYER: And this particular activity is more  
24 understanding dissolution performance of such alloy waste  
25 forms because after all, they are heterogeneous dissolution,

1 unlike the typical UO<sub>2</sub> or glass. What we have here, and I'm  
2 going to get to that at the bottom, but what we see is  
3 heterogeneous dissolution. So as we do dissolution tests and  
4 get data, it's very important that we understand its  
5 dependence to microstructure as we interpret the results. So  
6 that's really what we're trying to do here.

7           And the approach we've taken is we are developing  
8 what we said is a test protocol, and what that is is just an  
9 assembly of tests, the kinds of tests you need to do to  
10 define degradation performance, and we are working through  
11 the ASTM 26 committee to get a consensus on that  
12 methodology.

13           The tests we are looking at are the--and we are  
14 working with BNNL and INEEL in this activity.

15       DI BELLA: Excuse me. Di Bella. Are you testing the  
16 feed to the melt/dilute process, or the product, or both?

17       IYER: In this case, it will be the products.

18       DI BELLA: Thank you.

19       IYER: In this activity, we are primarily focused on the  
20 performance in the repository environment. So this is in the  
21 J-13 and the modified J-13 environments. And the kind of  
22 tests we are looking at are the flow test, drip test. Being  
23 an alloy form, it does offer--we have the flexibility of  
24 doing also chemical tests, especially for long-term  
25 predictability, and also vapor phase test.

1           But what I wanted to show here was the  
2 microstructure dependence of these on dissolution. That is,  
3 you take a typical MTR element, and what you see is aluminite  
4 particles,  $UAl_4$  typically, and what you see here is aluminum  
5 matrix with a eutectic microstructure. So what you are  
6 looking at here is kind of an off eutectic microstructure in  
7 the uranium aluminum phase diagram.

8           What happens in dissolution is people tend to  
9 relate dissolution of these fuels to aluminum dissolution  
10 rates, and that's debatable at this point because we know  
11 that a lot of the fission products we know are tied in with  
12 the core. In terms of the partitioning of the fission  
13 product between the aluminum matrix and the aluminite phases,  
14 we are trying to understand that. What is the partitioning  
15 of fission products? One could speculate a lot of it is tied  
16 up with the uranium, but we are trying to understand that at  
17 this point. And depending on that partitioning, what happens  
18 typically in the dissolution tests, or in J-13 environments,  
19 is the aluminum will start dissolving and the aluminite  
20 particles pretty much stay intact. And we've tried to kind  
21 of show that here where your matrix is kind of dissolving,  
22 and what you end up with is aluminite particles.

23           Now, for the aluminite particle, it's typically  
24 reasonably stable in the context of the aluminum, and so what  
25 is the release from those particles or the fission product

1 associated with the particles are at the particle boundary.  
2 And so what we are trying to do in this activity is make sure  
3 we can come up with a test and a methodology so that we can  
4 interpret the results correctly.

5 SAG<sub>γ</sub>IS: Okay, Sag<sub>η</sub>gs, Board again. Just to make sure  
6 of the terminology, we are talking about now dissolution in  
7 the aqueous phase of an alloy that may or may not be a dilute  
8 alloy.

9 IYER: Right.

10 SAG<sub>γ</sub>IS: I'm trying to understand the difference between  
11 dilution and dissolution here. Now, those microstructures  
12 that you're showing over there are for which waste form? Is  
13 that the initial waste form or the dilute?

14 IYER: No, this would be one of the diluted products.

15 SAG<sub>γ</sub>IS: Diluted like in which ratio again?

16 IYER: This is uranium, 19 per cent alloy, and it's an  
17 off eutectic microstructure. So what you end up with is a  
18 eutectic phase in that aluminum matrix.

19 SAG<sub>γ</sub>IS: Right. But the overall composition in there  
20 would be approximately what?

21 IYER: 19 per cent uranium.

22 SAG<sub>γ</sub>IS: 19 per cent uranium? And the rest?

23 IYER: Aluminum.

24 SAG<sub>γ</sub>IS: And the rest aluminum. And this is after

1 dilution?

2 IYER: Yes, this is after dilution, and typically if  
3 you're looking at MTR aluminum irradiated at 50 per cent  
4 burn-up, what you see is somewhat similar. That is, you  
5 primarily see the UA14 phase in aluminum matrix.

6 SAG<sub>γ</sub>IS: But the initial fuel, what composition would it  
7 have again? The initial fuel, what alloy composition would  
8 it have?

9 IYER: The typical MTR averages to about 19 per cent,  
10 but then it does go through irradiation, so there are  
11 changes.

12 SAG<sub>γ</sub>IS: Okay. Maybe we can talk afterwards. Somehow I  
13 get into the dilution process, may be confused here.

14 IYER: Tomorrow in the lab, we have posters with the  
15 phase diagram and all the different microstructures. We can  
16 show you, or I can show you. I'll be glad to pursue it. As  
17 I said, we have people--sometimes there's confusion, at least  
18 when I look at the phase diagram, in terms of weight per  
19 cent, and then the dilution levels. And so we have a matrix,  
20 and maybe that will help.

21 SAG<sub>γ</sub>IS: Okay.

22 IYER: I just wanted to kind of summarize the last slide  
23 again showing the aluminum form characteristics for both  
24 these options. I just wanted to summarize by showing some of  
25 the waste form characteristics for these options. The direct

1 disposal, as I said, the microstructure, you do see porosity,  
2 and typically what you see are those three phases, aluminum  
3 plus UAl<sub>3</sub>, plus UAl<sub>4</sub>, with melt/dilute, and what we've shown  
4 already with some early work in J-13 environments is that the  
5 corrosion resistance of aluminum plus UAl<sub>4</sub> is better than the  
6 mixed microstructure, which is aluminum plus UAl<sub>3</sub> plus UAl<sub>4</sub>.  
7 Now, how much incrementally better and what that does to  
8 performance assessment, that's a different question. But it  
9 is better.

10           And so with direct disposal you basically have the  
11 microstructure you get in hand. With melt/dilute, you do  
12 have the flexibility of tailoring the microstructure so that  
13 you can end up with an aluminum plus UAl<sub>4</sub> microstructure.  
14 When you get to criticality, we haven't done the criticality  
15 analysis, but the presumption is poisons, we know poisons are  
16 necessary for direct disposal, you'll hear that in your next  
17 talk. Poisons are probably going to be necessary for  
18 melt/dilute too. If necessary, though, we can make that part  
19 of the melt/dilute process so that the form will be integral  
20 to the microstructure again, so that you basically have  
21 uniform degradation as part of that. You basically form a  
22 UAl<sub>x</sub>, or whatever the poison is, compound.

23           In terms of radionuclide release for direct  
24 disposal, you have fission gases in the pores, fission and  
25 activation products, although we're not sure how it's

1 partitioned between those three phases. In melt/dilute, you  
2 pretty much, all the cesium is gone, the iodine is 90 per  
3 cent gone, krypton is gone, technetium is still there, and  
4 you do reduce the total inventory, so even if you do  
5 dissolution tests and you get the dissolution rate, you would  
6 expect that your actual release is going to be lower.

7           And then finally, proliferation resistance, which  
8 is depending on which party you talk to, which can be a big  
9 driver. You do have isotopic dilution. That was one of the  
10 attributes of this process when we initially looked at  
11 press/dilute, we did do a paper analysis on the press/dilute  
12 process and we found that we really can't get a homogeneous  
13 product and it actually created more problems in terms of  
14 drying and storage, because we had to insert DU plates and  
15 roll it, and it didn't make sense.

16           So in the context of melt/dilute, we have isotopic  
17 dilution and as long as we stay in the sub-liquidous or  
18 liquidous range in the phase diagram, which is what our plan  
19 is, and so there's no separation of U-235.

20           So this kind of gives you an idea of the technical  
21 activities that are underway, and our current plan is to, as  
22 I said, to have this body of information by next fall so that  
23 it will help in the decision making. In the meantime, the  
24 way this program is translating to do, primarily it will be  
25 in the validation stage next. That is, going from cold and

1 some hot work.

2 BULLEN: Thank you, Natraj. Questions from the panel?

3 SAGYIS: Yes, Sagyis, Board again.

4 Do I understand--so you're conducting right now  
5 corrosion rate measurements in the dilute form of the fuel.  
6 Do I understand then that there was very little corrosion  
7 rate information on this before now?

8 IYER: Yes. Primarily what's been used before now is  
9 the uranium metal corrosion rate, or the element of metal  
10 corrosion rate. There's been very little corrosion rate  
11 information of mixed microstructure, like UAlx in matrix.  
12 There's a lot of corrosion rate information in aqueous  
13 environment that is pure water or typical basin water. We  
14 have a lot of that, and there's a lot of information again in  
15 good quality aqueous vapor phase environment, because we have  
16 gathered a lot of that information in the context of dry  
17 storage. But in the J-13 repository environment and the  
18 modified J-13, there's I would say what we are generating is  
19 probably the first pieces of data.

20 SAGYIS: I see. And this data began to be developed you  
21 would say like in the last year or so?

22 IYER: In the last eight months. And so those tests are  
23 still underway and we don't really have the data coming in.  
24 I would say really in the last six months.

25 SAGYIS: So we're going to have corrosion rate

1 information over a very short time period, at least for  
2 coupons, and then of course--

3 IYER: You should have some preliminary data by spring--  
4 late spring to early summer, preliminary data. And a lot of  
5 this data, by the way, is as I said in conjunction with BNNL.

6 SAGYIS: All of this is of course very, very preliminary  
7 kind of information when we're talking about extremely long-  
8 term durability?

9 IYER: Correct. Let me qualify that. The corrosion  
10 rate information will be for the direct disposal form, that  
11 is, the fuel that we have. As far as the melt/dilute, until  
12 we go through the hot demo, we won't be testing that hot.  
13 But we will have qualitative information based on how these  
14 phases dissolve in these environments.

15 BULLEN: John Arendt?

16 ARENDT: Thank you. I want to make sure I understand  
17 these last three papers, or even the next one. Assuming that  
18 the aluminum clad fuel cannot be disposed of directly in a  
19 repository, the treatment options that you're looking at  
20 would be conducted in lieu of direct disposal; is that right?  
21 And if that is right, are you going to have to characterize  
22 each group of fuel, or are you going to have to inspect all  
23 the fuel? Or once you decide on a treatment technology, are  
24 you going to treat all the fuel in the same manner? Am I on  
25 the right track, or did I get lost?

1 IYER: Let me try to clarify that. What we are doing  
2 right now is we're not saying aluminum fuel cannot go to the  
3 repository. We are looking from the context of performance  
4 assessment; can we put aluminum fuel in a direct disposal  
5 form in the repository. And if for some reason that runs  
6 into problems, be it criticality or stakeholder acceptance or  
7 whatever that may be, we're looking at the backup option, the  
8 option of diluting the fuel, and then packaging that the same  
9 way you would package any other fuel. That is, once you get  
10 a diluted product, it still goes through the road ready  
11 package, and then disposing of that. So that's really what  
12 we're looking at. But at this point, there's a lot of work,  
13 and the next presentation is going to talk to this, which is  
14 looking at direct disposal of element of fuel in a  
15 repository, and there's some preliminary PA work done, which  
16 is I believe what the next speaker is going to talk about.

17 ARENDT: I guess what I wonder then if you, once you  
18 decide on the optimum treatment, will all the fuel be treated  
19 in the same way?

20 IYER: That's really--there's a lot of issues there.  
21 For example, the low enriched, less than 20 per cent salicite  
22 fuel, it will probably be a cost benefits kind of a driver,  
23 that is, can we dispose of that direct or in a co-disposal  
24 package, or does it make sense to take that to melt/dilute.  
25 So those are cost benefits issues.

1 ARENDT: I'm only talking about the highly enriched.

2 IYER: As far as highly enriched, it will either be the  
3 program we currently have underway, you're only looking at  
4 two options, which is direct or co-disposal as one, and  
5 dilution as the other one for all the high enriched aluminum  
6 fuel.

7 ARENDT: Right. But are you going to have to  
8 characterize or are you going to have to inspect? Will some  
9 of the fuel be--maybe it's too early yet, but will some of  
10 the fuel be able to be disposed of directly in a repository  
11 and some will have to be treated? And you're going to have  
12 to look at all the elements then to decide what the treatment  
13 is going to be, I assume.

14 IYER: I'm kind of giving you an answer off the top of  
15 my head. If we decide we're going to treat aluminum high  
16 enriched fuel, that is, the UAlx aluminum type fuels, my  
17 presumption is we'll probably treat all the fuel. That would  
18 make more sense, rather than trying to license two different  
19 forms. But as I said, there's a lot of work going on in  
20 trying to qualify direct co-disposal fuel, and at this point  
21 at least, we haven't heard anything in terms of the technical  
22 results that are coming back that would necessarily warrant  
23 that option. So we're looking at both.

24 BULLEN: Other questions from the panel? Carl Di Bella?

25 DI BELLA: I have a two-part question. What is the

1 temperature of your melter?

2 IYER: Oh, the melter temperature is very high, but the  
3 treatment process temperature we're looking at is below 1000  
4 C. It's around 850. But we're using an induction melter, so  
5 we could go very high if we had to.

6 ARENDT: At 1000 degrees C., you're going to volatilize  
7 a number of the fission products, that's correct, and  
8 apparently you're going to collect them. And where do they  
9 go?

10 IYER: Right. They're going to be--basically gets into  
11 different high level waste streams. The total inventory of  
12 the--when you look at the total 20,000 elements and look at  
13 the inventory, the total inventory of the cesium, iodine, et  
14 cetera, in the context of FRR is still very small in the  
15 context of all the high level waste we have. But they become  
16 part of our high level waste stream.

17 DI BELLA: Okay. They will come out in the HEPA  
18 filters?

19 IYER: Yes.

20 DI BELLA: And then how do you put HEPA filters into the  
21 high level waste stream?

22 IYER: Well, there's kind of a detailed washing process  
23 which Lee Hyder can talk to, and we'll show you that  
24 tomorrow.

25 DI BELLA: Okay.

1 IYER: But we have a scheme as to how we would basically  
2 wash out the cesium and other radionuclides.

3 BULLEN: Thank you very much, Natraj.

4 Moving on to our last speaker before lunch, not to  
5 put any pressure on Dave to be on time, we have Dave Haught.  
6 He is an engineer with the Yucca Mountain Site  
7 Characterization Project Office. He's responsible for the  
8 oversight of the development of waste package design,  
9 materials testing and modelling, waste form testing and  
10 modelling program, and work performed in support of the  
11 National Spent Fuel Program and the Office of Civilian--  
12 excuse me--Fissile Material Disposition. And Dave will speak  
13 to us today about the disposal of aluminum clad, highly  
14 enriched uranium spent nuclear fuels.

15 HAUGHT: As Dr. Bullen mentioned, I am David Haught, and  
16 I work at the Yucca Mountain Site Characterization Project.  
17 I'm not going to repeat all the stuff that he said.

18 One thing I would like to do is acknowledge some of  
19 the help that I've gotten in putting this together from Peter  
20 Gottlieb and Jim Doogood primarily, but others.

21 This is an overview of what I'm going to talk about  
22 today, the waste package design, performance assessment and  
23 the criticality analyses we've done to date on the aluminum  
24 clad HEU fuels. I'm also going to get into some of the  
25 process that we go through in performing these analyses.

1 That portion of it will not be repeated for the fissile  
2 materials disposition because the process is basically the  
3 same.

4           I put this up. Here's the waste package design  
5 that we have conceptualized to date. Based on the input from  
6 Savannah River, we have looked only at the MIT and the Oak  
7 Ridge Research Reactor Fuels, and so the design that you see  
8 here is for them. The concept we believe works across the  
9 board, but what you're going to see, because this is all  
10 we've looked at so far, I make no claim that MIT or Oak Ridge  
11 is bounding in any case. We haven't done that yet.

12           You've already heard a lot of this this morning.  
13 We are looking at a co-disposal concept with high level  
14 waste, and we are putting long term criticality control  
15 features in the canister.

16           I'm sure many of you have seen this picture before.  
17 In fact, I think you've even seen it earlier today. This is  
18 the co-disposal concept. You will occasionally hear it  
19 called the five pack. There may be an instance or two where  
20 we have a four pack, and the five refers to the number of the  
21 vitrified high level waste canisters that are in there.

22           Going right into performance assessment, we did a  
23 sensitivity analysis for DOE unspent fuel this past year, and  
24 besides just doing the sensitivity analysis for DOE unspent  
25 fuel, we took advantage of some of the things we have learned

1 since we did TSPA-95, and I'm going to go through some of  
2 these.

3           We have updated the percolation flux rate. These  
4 are the ranges. And if you looked across the repository  
5 footprint, that average worked out to about 6.2 millimeters  
6 per year. We looked at some various quantities of waste  
7 packages that would see drips, and the impact on that. We've  
8 updated the diffusion properties of the rock.

9           This is the same, but it bears mentioning because  
10 there have been a number of TSPAs done with different aerial  
11 mass loadings. The results I'm going to show you, or  
12 findings I'm going to show you today are based on 83 MTU per  
13 acre, centered-in-drift emplacement.

14           We've updated the near-field thermohydrologic  
15 calculations. And this last one, at least on this page, this  
16 last one is very significant. We have received some evidence  
17 and data that indicates that the solubility of neptune is  
18 about two orders of magnitude less than what we previously  
19 have been using.

20           Continuing on with some of the changes to the TSPA-  
21 95 bases, we have updated the waste package degradation  
22 studies. I had in here a slightly busier version of this  
23 slide that I deleted to try to avoid confusion, but I'm going  
24 to mention it anyway and take a chance.

25           Part of the updated waste package degradation

1 studies is we categorized waste packages into eight different  
2 failure groups. One of those groups was kind of--it was the  
3 most at risk group. We basically ignored any discussion or  
4 consideration of galvanic protection, and the results that  
5 you will see here for the aluminum based fuels are based on  
6 all the waste packages are in that group. So it's a fairly  
7 conservative curve.

8           We updated the saturation flux and the porosity.  
9 We have not included any climate cycles in this, which  
10 simplified things considerably. And at the bottom here is  
11 here are the amounts of materials that we have considered in  
12 this sensitivity analysis.

13           The findings: dose at the accessible environment.  
14 The peak dose is roughly equivalent to commercial spent fuel  
15 in both of the types of fuel that we looked at, the uranium  
16 aluminum alloys and uranium silicide fuels. We used those as  
17 basically two different groups.

18           Now, what you will see from the dose from the  
19 uranium aluminum alloy is that there is a peak earlier and  
20 around the 15,000 year time frame, due mostly to releases due  
21 to technetium and iodine. The curves I'm going to show you  
22 are going to be a combined release from all radionuclides.  
23 But typically what you see at that time frame is these are  
24 the major players.

25           It is less than an order of magnitude difference

1 for the aluminum--I'm going to call it the technetium/iodine  
2 peak and commercial spent fuel, and then it is also less than  
3 the peak dose, which occurs later in time.

4           Now, here's the graph for the comparison of the  
5 uranium aluminum alloy fuels versus an equivalent amount of  
6 commercial spent fuel, and here's that peak, the  
7 technetium/iodine peak that I was referring to before. But  
8 as you can see, the actual peak dose occurs out into this  
9 time frame, and they're roughly equivalent.

10           And in the case of the uranium silicide fuels, they  
11 are basically bounded, or roughly bounded by commercial spent  
12 fuel pretty much in all cases throughout time.

13       BULLEN: Bullen, Board. Just a quick question.

14       HAUGHT: Yes.

15       BULLEN: Kind of a point of order here, when you do the  
16 melt/dilute, which probably wasn't included in this because  
17 you're doing co-disposal, the iodine goes away, so you're  
18 going to drop that peak maybe a factor of two on order of  
19 magnitude, because there's no iodine?

20       HAUGHT: Yeah, you still have the technetium.

21       BULLEN: Right.

22       HAUGHT: But you're right. We have only looked at the  
23 direct disposal option.

24       BULLEN: Okay.

25       DI BELLA: While we're here, Di Bella, define

1 equivalency, please.

2 HAUGHT: An equivalent amount metric tons heavy metal.

3 DI BELLA: Okay.

4 HAUGHT: Okay, I'm going to move on to the criticality  
5 analyses, start off with kind of some ground rules that apply  
6 across the board. I've already mentioned it's based on the  
7 two fuels, and these are the enrichment characteristics of  
8 those fuels. We have analyzed the criticality potential with  
9 MCNP, and we have looked at some alternate neutron absorber  
10 materials and we have some conservative assumptions that we  
11 have applied to all the analyses. We are assuming fresh fuel  
12 and we have an optimum moderation in clay.

13 We're pursuing a phased analysis approach, and this  
14 is a description of Phases 1 and 2. There is a Phase 3.  
15 Phase 1 is the intact configuration, in which case we also  
16 identify a conceptual waste package, because the criticality  
17 analysis tends to be somewhat geometry dependent, so we have  
18 to kind of take these on somewhat of a case by case basis.

19 Phase 2, we look at degraded configurations within  
20 the waste package, and we have used EQ 3/6 to analyze the  
21 geochemistry. And we have varied some of the environmental  
22 parameters and corrosion and degradation rates.

23 Phase 3 is a cumulative analysis, and basically the  
24 Phase 3 is we look at the configurations outside of the waste  
25 package. So these are some of the things that we're going to

1 do.

2           One of the things that I had been asked to talk  
3 about today was the estimate of probability and consequences,  
4 since we are advocating a risk based approach, and Phase 3 is  
5 where that is done. The bad news is I'm not prepared to  
6 talk about Phase 3 today. That's planned for FY99.

7           Let me mention this, because I believe I need to  
8 talk about it later. When you look at the external  
9 configurations to the waste package, there are kind of three  
10 predominant deposition mechanisms that are of concern.  
11 That's absorption of the fissile material on clays or  
12 zeolites, the presence of a reducing zone for either organic  
13 or hydrothermal upwelling of hydrogen sulfide, which would  
14 tend to have the fissile material kind of collect there, and  
15 then there's a general chemical reaction with the host rock.  
16 And that is something that--a thought that needs to be  
17 carried on to the discussion about plutonium.

18       BULLEN: Bullen, Board. Just a quick question.

19           You said that the Phase 3 analyses won't be done  
20 until FY99. Does that mean that these data won't be  
21 available as sensitivity analyses for the TSPA/VA that's  
22 going to be done? Or will there be some VA analysis that  
23 will include these kinds of sensitivities?

24       HAUGHT: No, I don't believe that we will have any of  
25 this done as the sensitivity analysis for TSPA/VA for the

1 aluminum, or really any of the DOE spent fuels.

2 BULLEN: Okay. So the criticality analysis for VA will  
3 just be spent nuclear fuel and defense high level waste cans;  
4 right?

5 HAUGHT: Correct.

6 BULLEN: Okay.

7 HAUGHT: For external.

8 BULLEN: Right.

9 HAUGHT: Now, I apologize somewhat for this chart. It's  
10 very difficult to read, but there is a hand-out.

11 This is a configuration generator for how we get to  
12 the different--well, the degradation scenarios for how we  
13 reach the various configurations that we have evaluated. And  
14 there are four basic configurations that we have looked at.  
15 This is one of the places where the probabilities, we haven't  
16 assigned the probabilities yet for these boxes, and this is  
17 one of the places where the probabilities come into play.

18 The first configuration is basically a homogeneous  
19 mix of the clay and the degraded fuel within the waste  
20 package. I'll go through these quickly. We have a possible  
21 configuration where we have the fissile material stratified  
22 on the bottom.

23 SAGYIS: Excuse me. Sagηgs, Board.

24 HAUGHT: Yes, sir.

25 SAGYIS: Can you explain a little bit the meaning of

1 clay in your diagrams?

2       HAUGHT: It is the resultant degradation of the  
3 vitrified high level waste as it, you know, as it is altered  
4 and corroded, it tends to form a sort of clay-like mixture.

5       SAGYIS: Okay. In other words, the clay would be a  
6 corrosion product?

7       HAUGHT: Of the high level waste, yes.

8       SAGYIS: And the corrosion product of the composition of  
9 the ceramic--

10       HAUGHT: Right.

11                Another configuration that's possible is a  
12 stratified on the top, and finally, there is the possibility  
13 that we have some extreme stratification within the canister  
14 that contained the aluminum clad fuel. In other words, the  
15 canister is still roughly intact, so all of the fissile  
16 material is in here. It is not mixing with the clay, and  
17 it's still inside the waste package.

18                I'm going to go through some of the findings on  
19 criticality now. In the case of the Oak Ridge SNF, if that  
20 canister contains a carbon steel basket having a borated  
21 stainless steel between-layer separator, our evaluations are  
22 showing that that can remain subcritical in all  
23 configurations.

24                In the case of the MIT, it's a little more  
25 complicated. The SNF in an intact basket--now, in this case,

1 the fuel has degraded but the basket is still intact--we  
2 require approximately a kilogram of either boron or  
3 gadolinium distributed in an absorber plate.

4           Going further down, if you degrade both the basket  
5 and the MIT fuel, now you're looking at you have to have .25  
6 kilograms of gadolinium homogeneously distributed within the  
7 soup that's in there, if a stainless steel basket is used,  
8 but .12 kilogram if a carbon steel basket is used. The  
9 reason for that is we're considering the fact that as the  
10 carbon steel degrades, you have all the rust, and it takes up  
11 more space that otherwise water could fill.

12           More findings on MIT. Configurations external to  
13 the canister but internal to the waste package, stratified on  
14 top, we need .2 kilograms of gadolinium homogeneously  
15 distributed with the SNF. In this case, it's not homogeneous  
16 throughout the waste package; it's just in that top. And  
17 that is ignoring any contribution due to the iron.

18           Stratified on the bottom is .1 kilogram of  
19 gadolinium. And then the homogeneous mixed in clay, our  
20 evaluations are showing that that remains subcritical.

21           Current status is that Phase 1 is complete. Phase  
22 2 is in review. And Phase 3 is planned for FY99.

23           In summary, the co-disposal concept appears  
24 workable. We're showing a small impact to repository  
25 performance. I need to caveat that somewhat that in both

1 cases of the aluminum clad fuel and commercial spent fuel, we  
2 are assuming no credit for cladding. If we were able to  
3 develop a technical basis for taking credit for cladding in  
4 the commercial spent fuel, that may change somewhat. But  
5 given the situation we're in right now, they look equivalent.

6           The internal configurations can be maintained at  
7 subcritical levels. And then here's the work that we have  
8 yet to do.

9           That's all I have. It looks like you have a  
10 question.

11          BULLEN: Thank you, Dave. This is Bullen, Board. I  
12 applaud your ability to solve the problem that was posed, but  
13 the question I have is what criteria do you use, or how do  
14 you make a decision that maybe it's not such a good idea to  
15 dispose of or directly dispose of aluminum clad fuel, and you  
16 go back to the melt/dilute as an option? I see the analysis  
17 and I see that yes, indeed, if we so chose and we could get  
18 the homogeneous mixtures, we could design it so it wouldn't  
19 go critical. When or where does the program plan to make a  
20 decision as to do we co-dispose or direct disposed, or do we  
21 melt/dilute?

22          HAUGHT: I view that decision as not being RW's to make.  
23 That is Savannah River's to make. They have asked us to  
24 help them assess the feasibility of direct disposal, and  
25 that's the work that we're doing.

1           Now, I would hope that we're given an input into  
2 that decision, but I would still say that is Savannah River's  
3 decision.

4           BULLEN: Do you seriously think that you could just set  
5 a waste acceptance criteria that says I don't take this  
6 stuff? I mean, that's RW's decision; that's not EM's  
7 decision.

8           HAUGHT: The waste acceptance criteria, that's correct.

9           BULLEN: Right. And so if you set a waste acceptance  
10 criteria that said it's too hard to make this not critical,  
11 let's make you guys dilute and then melt it, don't you think  
12 that's something that RW could do?

13          HAUGHT: I think, yeah, I believe yeah, we have the  
14 purview to do that. We could set an impossibly high  
15 acceptance criteria for these fuels that would basically  
16 force Savannah River into a melt and dilute option.

17          BULLEN: I guess the follow-on question is it's going to  
18 cost money to add gadolinium, it's going to cost money to  
19 make sure the boron, stainless steel plates are there. It  
20 might be a whole lot cheaper for you to just say the waste  
21 form acceptance criteria is, boom, and you're done. And then  
22 the cost benefit essentially comes back to here, where you  
23 take a look at the waste form that's coming out of the  
24 processing facilities that you accept from.

25           I guess the question there would be how do you

1 communicate the cost differences? I mean, if you're going to  
2 do a cost benefit analysis, it's got to be the total system.

3       HAUGHT: I think with regard to our terms, and I'm  
4 really speaking off the cuff now, within some reason,  
5 whatever those acceptance criteria are is somewhat immaterial  
6 to us. You know, obviously, we don't want to throw things  
7 out to extremes, but I think within the acceptance criteria  
8 that I could imagine that would apply to a direct dispose  
9 versus a melt and dilute, the cost to RW to license that  
10 waste package is probably roughly the same, and given the  
11 quantity of waste packages that we're building, gets somewhat  
12 lost in the noise.

13               But, you know, again, I have kind of a range on  
14 what I envision those acceptance criteria would look like,  
15 and I'm assuming--I see that Mark just grabbed a microphone.  
16 I think he wants to weigh in.

17       BULLEN: Other questions from the panel first, and then  
18 we'll do that. Alberto?

19       SAGYIS: Sagyis, Board. So if I understand the  
20 diagrams, most of the reaction products are not going to be  
21 expansive. They're still being considered in these  
22 scenarios, the volume of the resulting product after  
23 interaction with the ingressed water?

24       HAUGHT: Yeah, there is some expansion of volume. I  
25 don't think there's enough to exceed the capacity of the

1 waste package itself. And I might add that there were three  
2 configurations that I didn't provide to you here in this  
3 package, primarily because they're not very interesting, one  
4 of which is where the materials have been flushed out of the  
5 waste package. And then there's the fully intact, and then,  
6 you know, the intact basket with the degraded, because in  
7 both cases, the criticality control features are in place.

8           But, Jim, do we have enough expansion to actually  
9 split the waste package open, or Peter?

10          GOTTLIEB: I'm Peter Gottlieb with the CRW M&O.

11           The waste package is one container. We could have  
12 the criticality because there's water in the waste package.  
13 There's obviously penetration of the waste package. Whether  
14 the waste package retains water or not once it's penetrated  
15 is a question that we analyzed both ways. We could have  
16 criticality even if the waste package doesn't retain water.

17           Was that what the question was?

18          SAGYIS: No, really what I meant was simply assume a  
19 number of reactions in the system must come in and creates  
20 products that have a certain molar volume, and then from  
21 there, you get the final volume of whatever was inside after  
22 interaction with water and oxygen and whatever else that may  
23 be in the package. So I was wondering if in these scenarios,  
24 the initial versus the final volume was considered, or if  
25 you're considering other things?

1           GOTTLIEB: Well, let me put it this way. The  
2 geochemistry code that was used, EQ 3/6, did a water balance  
3 for the water that's coming in and the water that's flowing  
4 out, whether the water is flowing out through holes in the  
5 bottom or overflowing through holes in the top. But there  
6 was a water balance which also included the water that would  
7 go into deposited minerals as a result of the reaction.

8           SAGYIS: So, anyway, do I understand then that the  
9 overall result of this was that the volume of the final  
10 product is not larger than the initial volume inside the  
11 container?

12          GOTTLIEB: Well, but the point is the initial volume  
13 inside the container is not necessarily the controlling  
14 parameter because you have water coming in.

15          SAGYIS: Right. Go I guess the question is after--

16          GOTTLIEB: So you could end up in a situation, although  
17 it's pretty extreme, you could end up in a situation where  
18 the whole waste package is filled with clay and other  
19 degradation products. More likely, much of that will have  
20 overflowed, and so the volume in the waste package won't be  
21 more than what was there originally. But it would never--you  
22 always have sufficient holes in the waste package for the  
23 water to get in, such that whatever was--any reaction  
24 products which increased the size would overflow the waste  
25 package. It wouldn't burst it.

1           SAGYIS: But of course you encounter a whole bunch of  
2 situations in which if you allowed something inside the  
3 material, there is enough process to allow for the water to  
4 come in, but the reaction product has a physical makeup such  
5 that it does not allow for it to come out through the same  
6 holes through which the water came in.

7           GOTTLIEB: Well, if you were to get very rapid  
8 reactions, that is possible. These are very slow reactions.  
9 Everything that we are modelling with the geochemistry code  
10 indicates these are reactions that take place over thousands  
11 or tens of thousands of years.

12          SAGYIS: Okay, thank you.

13          BULLEN: Other questions from the panel? Carl?

14          DI BELLA: Carl Di Bella. After hearing the last two  
15 presentations, I have a question for Mark Barlow, if he's  
16 still in the audience. And that is why wasn't the dissolve  
17 and dilute technology included in the alternatives studied?

18          BARLOW: The dissolve and dilute option was evaluated by  
19 the Research Reactor Task Team and in fact it was I think,  
20 and I'm going by memory now because I wasn't directly  
21 involved in that, but my recollection in reading the report  
22 was it was judged unfavorably compared to the other  
23 dissolution options. So it was not carried forward to the  
24 final grading, but it was considered.

25          BULLEN: Any other questions? I have one final one that

1 maybe I didn't get answered when I asked it originally. Is  
2 there a mechanism whereby either RW or EM can decide not to  
3 direct dispose? And if so, how does that happen?

4       HAUGHT: There is a mechanism, and in fact one of the  
5 ways--in fact there are mechanisms, when you get right down  
6 to it--one of the ways that that could be done is, as you  
7 said, we could give Savannah River an acceptance criteria  
8 that they couldn't meet. Before we did that, I think we  
9 would have to get together and make certain that such a  
10 decision would be in the interests of both of our programs in  
11 total, not that we would have to research a consensus on  
12 that, but that, you know, you would have a net benefit to it.

13           Other than that, it's really just a matter of  
14 working closely with them, telling them the kind of things  
15 that will have to occur in the design of their canister in  
16 order for an acceptance criteria to be met for direct  
17 disposal. I mean, one person from Idaho has described to me  
18 that, you know, right now what we've got--we have an  
19 acceptance criteria right now, and that's commercial spent  
20 fuel o borosilicate glass, and we could do that. But we are  
21 working with the National Spent Fuel Program to try to come  
22 up with some other different categories along with it, and we  
23 believe that that's probably of the greatest benefit to both  
24 of our programs.

25       BULLEN: Mark Barlow?

1 BARLOW: Mark Barlow from Westinghouse.

2 Westinghouse of course won't be making the  
3 decision, but we will be making recommendation that will  
4 reflect the waste acceptance criteria and the input and the  
5 feedback we've gotten both from RW and USNRC, National  
6 Academy of Science.

7 The decision that the Department will make right  
8 now regarding technology would be reflected in a Record of  
9 Decision to what was referred to as the Site Specific EIS.  
10 The current anticipated date for that Record of Decision is I  
11 think September or October of next year. And along with that  
12 decision will be a decision regarding this treatment and  
13 storage facility at the Savannah River site, and which fuels  
14 would be reprocessed and what kind of treatment would be  
15 involved.

16 BULLEN: Bullen, Board. So that would provide the input  
17 from your National Research Council review, from the NRC,  
18 from DOE, and then potentially any comments that we might  
19 have with respect to that?

20 BARLOW: And any stakeholder who reviews the draft EIS  
21 would have input that would be considered in that Record of  
22 Decision.

23 BULLEN: Okay, thank you.

24 At the risk of finishing four minutes early, and  
25 I'll thank Dave for that, we are now adjourned until 1:25,

1 when we will reconvene with the presentation about Savannah  
2 River Site.

3 Thank you.

4 (Whereupon, the lunch recess was taken.)

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A F T E R N O O N    S E S S I O N

1           BULLEN: Good afternoon and welcome back. Could I ask  
2 the Board members to come up and take their seats at the  
3 table, and everyone else to grab their cup of coffee and have  
4 a seat, please?

5           In our afternoon session, we're going to shift  
6 gears a little bit away from the highly mixed uranium and  
7 aluminum clad spent fuel and talk about, first, a little  
8 background on the Savannah River Site and the Defense Waste  
9 Processing Facility and then some of the characteristics of  
10 the vitrified high-level waste and the disposition of surplus  
11 weapons plutonium.

12           Our first speaker is Charlie Anderson. Charlie  
13 Anderson is going to give us an introduction to the Savannah  
14 River Site. He is currently the director of reactors and  
15 spent fuel division and responsible for the management and  
16 direction of the reactor programs and the spent nuclear fuel  
17 program.

18           Charlie?

19           ANDERSON: I'm going to begin with a map because I could  
20 tell from discussions during lunch with several people that  
21 the size of Savannah River is probably something that a lot  
22 of people have a hard time visualizing. Savannah River Site  
23 is 310 square miles, thereabouts, about 190,000 acres, and  
24 right here is the administration area, what's commonly  
25 referred to as the 700-Area and includes the materials

1 manufacturing facility. But, as a flow of material during  
2 the days when Savannah River was in production reactors, the  
3 fuel was fabricated here in M-Area. We had five production  
4 reactors which are basically--here's C, K, L--the colors  
5 don't come out here--R, here and here; so a kind of  
6 semicircle around the center of the site with the Canyon  
7 Separations Facilities here in F-Area and in H-Area where  
8 also the tank farms that are associated with each of the  
9 Canyon Separations Facilities. Then, DWPF right here, and  
10 Saltstone at this point here. There's some other facilities  
11 and I'll show a few slides here, some of those as we get  
12 through, but the 400-D Area and the pump house and this area  
13 right here. I was trying to think, I'm not sure. The  
14 dimension across here is right around 15 to 20 miles. It is  
15 20 miles to give you some idea then of the dimension. And,  
16 of course, the Savannah River runs right along in through  
17 here just to give a feel and a flavor for the size of the  
18 site.

19           It was established in 1950 by the Atomic Energy  
20 Commission; its purpose to produce nuclear materials for  
21 national defense. In 1972, it was designated as the nation's  
22 first environmental research park. The original facilities,  
23 the fuel and target fabrication facilities, I'm going to show  
24 some of these here. This is what is in M-Area where the fuel  
25 fabrication facilities are. This is a picture of one of the

1 reactors. This is K-Reactor. In this picture here, this is  
2 H Canyon, and then as you look in the background here, this  
3 is DWPF, the glass waste storage building. Over to the left  
4 right in this area just to the left would be the tritium  
5 facilities, and right here is the first part of the tank  
6 farms. In fact, in this area right here, were the first four  
7 tanks in H-Area and the tank farm would extend that in this  
8 particular area here. The term "Canyon Facility", if you  
9 were to look inside this building here, particularly during a  
10 construction area the way it's arranged, it looks like  
11 canyons where the process vessels fit within the shielded  
12 construction.

13           Another picture gives a little different view that  
14 shows--actually, if you put these two together, you would see  
15 quite a bit. This is the tritium facilities here with this  
16 being DWPF and Saltstone right here in the back. These are  
17 Saltstone vaults with the Saltstone Processing Facility right  
18 here. And, this is another part of the tank farm right here  
19 where the new style Type 3 tanks which are full secondary  
20 containment type tanks.

21           Also, in this area, this is RBOF, the Receiving  
22 Basin for Offsite Fuels. This is an interior shot of RBOF  
23 showing where we receive and have been receiving for some  
24 time the fuels from research reactors across the world and  
25 domestic research reactors in the United States, too. This

1 is the basin then where they are received and a good bit of  
2 them are now stored. The reactors, all five, pretty much,  
3 you know, are very similar as far as when you look at them.  
4 There would be different outbuildings and that type of thing.

5           It was originally operated for 40 years, 40 plus  
6 years, by the DuPont Corporation which is the difference at  
7 Savannah River Site as compared to the other sites in the DOE  
8 complex. There were a lot of contractor changes over the  
9 years at a lot of the sites and Savannah River for 40 years  
10 had the same contractor. Since that time, Westinghouse since  
11 1989 has been the operating contractor for Savannah River  
12 Site.

13           Of particular notice at Savannah River Site is that  
14 there's been a number of facilities that have been brought on  
15 line recently in the last few years. The Defense Waste  
16 Processing Facility which Neil Brosee is going to talk more  
17 about here shortly, the Saltstone Facility, In-Tank  
18 Precipitation which are feed preparation facilities for the  
19 Defense Waste Processing Facility. A lot of people refer to  
20 Defense Waste Processing Facility as the glass plant, the  
21 vitrification facility. The Consolidated Incinerator  
22 Facility has been brought on in the last, oh, over a year and  
23 a half, I guess, right now--about a year, right at a year  
24 ago. Replacement Tritium Facility and Low-Level Waste  
25 Vaults, a significant change in how we dispose of low-level

1 waste. In fact, that was something that I wasn't able to get  
2 a real good picture of, but looking at the site. I'll go  
3 back here; I meant to mention that. Right in this area right  
4 here is where a lot of the solid waste disposal facilities  
5 are, both the vaults and some of the true waste type interim  
6 disposal.

7           Major facilities restarted, F and H Canyons, heavy  
8 water purification, and then tank farms have been in  
9 operation for quite a few years. Even in those areas since  
10 they never really stopped operation, there's been some major  
11 milestones that have been accomplished there, too; the first  
12 tank closure in the DOE complex and, of course, a lot of  
13 infrastructure and construction in that area, as far as being  
14 able to support the retrieval of the waste and feed to the  
15 Defense Waste Processing Facility and cleaning out of some of  
16 the older style waste tanks.

17           Spent fuel basins, a lot of times people said my  
18 division's title is reactors and spent fuel division and none  
19 of the reactors are operating now. There's been a major  
20 change in the drive for the operating staff there that  
21 instead of operating a reactor, they are becoming much more  
22 as basin managers and looking, of course, at managing the  
23 special nuclear materials of the spent fuel in the basins and  
24 also how the enriched uranium and heavy water and other  
25 materials that we do have on-site that we're managing until

1 the ultimate disposition of them is determined or carried  
2 out, depending. Some of them where there is a path forward  
3 and others where we're still doing some studying to determine  
4 the final disposition.

5           With that in mind for the facilities and all, the  
6 primary missions at Savannah River Site, high-level waste  
7 processing to glass vitrification, environmental restoration,  
8 making a lot of progress in areas such as seepage basins and  
9 a lot of other areas across the site where we're going back  
10 and closing and returning things back to a stable condition.  
11 Stabilizing the nuclear materials left over from weapons  
12 production programs, a lot of emphasis there in order to put  
13 them in a stabilized form. Spent nuclear fuel receipts,  
14 obviously, the Foreign Research Reactor Program which is a 13  
15 year program. The main intent was to recover the highly  
16 enriched uranium that was provided by the United States to  
17 foreign countries under the Atoms for Peace Program.  
18 Research and development programs, there was a lot of  
19 research and development over the years at Savannah River  
20 Site for production missions. There's a lot of research and  
21 development now, particularly in the last 10 years and it's  
22 still heavy now as far as environmental restoration and  
23 material stabilization in order to achieve the other missions  
24 at Savannah River Site. Then, the economic development  
25 efforts. Trying to turn over defense capabilities and

1 defense attributes at the site and see if they can't be  
2 developed for peace time or turned into something that can  
3 help the area as far as economic development.

4           Operating philosophy always has been an increased  
5 emphasis for safety, disciplined operations, cost  
6 effectiveness for all of us that look at our budgets and how  
7 the cost of doing business has definitely risen. We've had  
8 to take a hard look and try to make that more effective, try  
9 to cut down on our costs, continuous improvement, and a  
10 teamwork that values people. When we look at the budget  
11 here, one of the biggest portions of the budget are the  
12 people at Savannah River Site.

13           Current staffing. Primary contractor, it says  
14 Westinghouse and partners. Westinghouse had a five year  
15 contract of its own before, and when that contract was  
16 renegotiated, they teamed up with several other contractors  
17 also. So, that's the reference here to Westinghouse and  
18 partners. It includes Babcock and Wilcox and BNFL and a  
19 number of other smaller contractors, about 13,000 people.  
20 Wackenhut which is the security contractor for Savannah River  
21 Site which is 750, our other contractors around 500 with a  
22 DOE staff at about 550 at this point, Department of Energy.  
23 So, that brings the total staffing a little less than 15,000  
24 people at the Savannah River Site. With the budget as shown  
25 here--and, even though these carry out to four significant

1 figures, I really intended to revise this chart to be a  
2 little--because it does vary up and down. There's lots of  
3 discussions about budgets around a \$1.4 billion range.

4           Future of Savannah River Site. Ongoing missions,  
5 obviously not only the accomplishment of environmental  
6 restoration, but demonstration of new technologies and  
7 methods of doing environmental restoration. High-level waste  
8 processing to glass, still in this country the only  
9 vitrification facility for high-level waste. Stabilization  
10 of legacy nuclear materials determining both the path forward  
11 and actually implementing the stabilization of those  
12 materials. Ongoing economic development efforts to leverage  
13 Cold War technologies and capabilities into the local  
14 economies.

15           Also, there's a lot of--as far as new missions for  
16 Savannah River Site. Most people have heard accelerator  
17 production of tritium; still one of the big goals.  
18 Additional nuclear material stabilization missions. As DOE  
19 as a complex looks at the various materials that they have  
20 across the complex, where they are, where they can be  
21 treated, there's a lot of discussion and looking at some of  
22 the facilities and the capabilities at Savannah River Site  
23 for those type of missions. Planned new facilities, actinide  
24 storage vault and spent fuel transfer and storage facility  
25 which is--the spent fuel and transfer and storage facility is

1 one that's looking at dry storage and preparation of nuclear  
2 material for the disposition into the repository. I  
3 apologize for having to leave a little early this morning,  
4 but I assume you all probably covered a lot of that  
5 information on the spent fuel this morning.

6 I meant for this to be fairly brief, but if there  
7 are any questions or anything, I can help. If not, I was  
8 going to let Neil--we'll get into the processes for DWPF.

9 BULLEN: Questions from the Panel? Actually, I have one  
10 and then Carl Di Bella.

11 You noted that Westinghouse had been the prime  
12 contractor for almost nine or 10 years now. Is there a  
13 change in the offing with acquisition of NBC? Does it look  
14 like there's going to be a change in the prime contractor?  
15 Or CBS, I'm sorry, get the right network here. Is there a  
16 change in the contracting that's going--

17 ANDERSON: In the name change, and as far as the  
18 contract is concerned, there is no changes that are planned  
19 as far as the contract part, no.

20 BULLEN: That would be foreseen, okay.

21 Carl?

22 DI BELLA: I'm curious. What were the materials that  
23 the materials that the materials production reactor made or  
24 perhaps concealed?

25 ANDERSON: Tritium and plutonium.

1 DI BELLA: Weapons grade plutonium?

2 ANDERSON: Yes. I'm not sure--it is weapons grade, but  
3 I mean, if you're going to get into a whole lot more detail  
4 than preliminary, I'm probably going to have to call in some  
5 help here.

6 BULLEN: Any other questions from the Panel?

7 (No response.)

8 BULLEN: Thank you very much. That was a very nice  
9 overview.

10 Our next speaker is Neil Brosee. After a  
11 distinguished career in the military and in civilian nuclear  
12 power, he joined the Savannah River Site or Westinghouse  
13 Savannah River Company in 1993 as a deputy manager of the  
14 Defense Waste Processing Facility. While deputy manager, he  
15 successfully completed both the code runs and the waste  
16 qualification runs for the DWPF, and just recently in August  
17 of '96 became the manager of the DWPF facility. He will be  
18 speaking to us about the DWPF.

19 BROSEE: Good afternoon. I'm going to cover the  
20 facility itself in an overview. The first thing to notice in  
21 high-level division is extremely unique in all of the areas  
22 on the Savannah River Site. The main reason is it's  
23 integrated. The Defense Waste Processing Facility cannot  
24 operate without every other facility in this division also  
25 operating. That's very unique in this area. It's not like a

1 canyon or a reactor. I have to have receipt of sludge,  
2 precipitant. I also have to have the tank farm to receipt  
3 for my recycled waste. So, the main point to gain is that  
4 the only way that this facility runs is with every other  
5 facility in the high-level waste division also operating.

6           Size-wise, going from Charlie's overview of the  
7 Savannah River Site, the waste right now is about 34 million  
8 gallons of waste stored in 51 underground tanks. I actually  
9 should say 49. We have closed the first two tanks in the  
10 complex. The first tank was actually closed in July of this  
11 year, and we just finished closing the second tank this  
12 month. So, it was the first two tanks; there are now 49  
13 underground tanks. The extended sludge processing section of  
14 the tank farm provides the sludge for my facility. Likewise,  
15 the other part of the in-tank precipitant and the late wash  
16 provides the precipitant to the facility, and I'll go through  
17 the process. Likewise, I send the recycle back to the tank  
18 farm. The canisters are moved and stored in the safe interim  
19 storage of the glass waste storage building.

20           Just to capture some of the time frame since you  
21 were last here, in 1993, we completed our cold chemical runs  
22 where we actually proved the process with the cold chemical  
23 feed. In 1994, we initiated melter heatup. We initiated  
24 melter feeding, and we poured our first non-radioactive  
25 canister. Later in 1995, we completed the waste

1 qualification runs which we're going to provide an overview  
2 of the details of what that included, but that provides a  
3 test of the extreme range of the glass that we can make and  
4 its acceptability. We completed the DOE ORR, operational  
5 readiness review, and on March 12, 1996, we commenced  
6 radioactive sludge operation.

7           The mission is very similar to, as Charlie pointed  
8 out, the one item of note that I'm very proud of is the first  
9 one. First of all, in October of this year, the site was  
10 awarded the ISO 14001 certification. We've also gone through  
11 the DOE's review of the VPP program for our safety program,  
12 as well as the DOE review of our integrated safety management  
13 system, both Phase 1 and Phase 2. So, we provided an  
14 integrated safety program that performs this evolution of  
15 vitrification.

16           This is a very simplified view of the entire  
17 process. I'll walk through the entire process and then I'll  
18 over each area in detail. As I mentioned, we receive the  
19 sludge from the tank farm into the chemical processing cell.  
20 Two main areas right now are the SRAT which is the Sludge  
21 Receipt and Adjustment Tank and the SME, Slurry Mix  
22 Evaporator. There are several other tanks involved, but  
23 those are the two of concern. The feed is then prepared. It  
24 is then transferred to the melter which has been continuously  
25 energized since its initial heatup. In the melter, we then

1 melt the glass or the frit slurry and then we pour into  
2 stainless steel canisters and then the canisters are then  
3 deconned, welded for final closure, and then moved one at a  
4 time for interim storage in the glass waste storage building.

5           Presently the heart of the melter feed preparation  
6 is the chemical processing cell. We receive the sludge into  
7 the SRAT which is the Sludge Receipt and Adjustment Tank. It  
8 is chemically adjusted, and for the present, we are using  
9 simulated PHA for the precipitant side. It is then processed  
10 and transferred to the Slurry Mix Evaporator. At this point  
11 is where the glass frit is added to the Slurry Mix Evaporator  
12 to start the blending process. It is then transferred to the  
13 melter feed tank which is an interim tank before it is fed to  
14 the melter. Likewise, this shows the connections back to  
15 other systems which I need to have functional in order to  
16 operate.

17           In the melter which is operating approximately 1150  
18 degrees Centigrade, the feed is fed into the top and it forms  
19 a melt pool right here. There are electrodes actually  
20 sending current through the glass. It's a jewel heated  
21 melter and that actually melts the frit slurry to make molten  
22 glass. The frit slurry that has not yet melted forms a crust  
23 or a co-path that floats on the top of that melted pool. The  
24 offgas is taken from the top of the vapor space and we have  
25 the vapor heaters or dome heaters, as we call them. The

1 glass is then moved up through the pour spout and poured into  
2 the stainless steel canisters. The stainless steel canisters  
3 are approximately 10 foot tall, two feet in diameter, and  
4 they're three-eighths of an inch thick stainless steel. At  
5 this point when the canister is full, there is a temporary  
6 tapered plug that's inserted into the top of that canister.  
7 We then perform a helium leak check in order to show that it  
8 is airtight. We need that before we send it on into the  
9 decontamination process. If it fails that leak test, we then  
10 reprocess that can back through in order to have it pass  
11 before we send it into the decontamination chamber.

12           In the chamber itself, one-by-one, the canister is  
13 grappled by its flange and with the top section it is rotated  
14 and moved up and down while it's being blasted with a frit  
15 slurry compound. Think of it as sandblasting. That removes  
16 any of the surface oxide or contamination on the outside of  
17 that can so the outside of the can is meeting all of the DOT  
18 transportation requirements. That frit slurry that we used  
19 at decon is now sent back and is used into the Slurry Mix  
20 Evaporator for the next batch. So, that's one way that we  
21 reduce that type of waste. We just reuse it into the next  
22 batch that we process.

23           At that point, it is moved out of the chamber. It  
24 is smeared to show that it has been cleaned or it is re-  
25 decontaminated and it is sent to the weld test cell. At that

1 point, there is a hydraulic ram that pushes the tapered plug  
2 into the throat of the canister and puts the actual final  
3 welded cap in place. The cap itself is actually larger than  
4 the flange and this is an upset welding; in fact, it's the  
5 largest upset welder in the world. We press that top into  
6 the flange while we're applying 240,000 amps for one and a  
7 half seconds with about 80,000 pounds of force. That then  
8 provides a weld that's as strong as the three-eighths thick  
9 canister itself. It is then done a final smear and then  
10 transported to the glass waste storage building.

11           The glass waste storage building is a seismically  
12 reinforced concrete block below grade. The SCT is a unique  
13 device. It alone weighs close to 120 tons. It handles each  
14 canister at a time. It has a shielded container that pulls  
15 the canister up inside. It moves; its top speed is about  
16 five miles per hour. We only use two miles per hour. We  
17 transport it from the vitrification building to the glass  
18 waste storage building. The machine is powered by two  
19 diesels for reliability. It lifts the plug from the floor  
20 and that's about a four foot thick plug. It then moves the  
21 trolley forward, puts the canister in its slot, then rotates  
22 the trolley back and puts the plug back in the hole, and then  
23 goes back and picks up the next one. There's over 2200 slots  
24 in this first glass waste storage building for storage. It  
25 does have forced ventilation, but the natural air flow is

1 designed to come in. The air arises around the canister and  
2 actually will cool itself.

3           For the three years since March 12 that we've been  
4 running radioactive sludge, the first year we had a set goal  
5 of 60 and we beat it. The next year, we had a set goal of  
6 150 more and we beat that. And, right now, you can add five  
7 more to that. We just finished the 39th can today. So, for  
8 a brand new one-of-a-kind facility, that's the first three  
9 years of production. We did pour the one millionth pound of  
10 vitrified glass on the 3rd of November with the 14th canister  
11 this year.

12           Now, the waste acceptance process is just as  
13 involved as the process itself. We have established the  
14 waste acceptance product specifications that we meet. These  
15 basically have five sections. There's three technical  
16 sections, there's a QA section, quality assurance, and then  
17 the requirement for documentation and administration. The  
18 way in which we meet that at the Defense Waste Processing  
19 Facility is through four steps. We have the waste form  
20 compliance plan which is a general description of the process  
21 and the methods by which we meet these specifications. We  
22 then have the waste form qualification report which is at  
23 least 13 volumes of all the tests that we have performed and  
24 the results that show the compliance with these  
25 specifications.

1           We now have the production records which we use for  
2 every can and every batch to show that we have complied  
3 during the production with this program that we have  
4 established. And then, we have the final storage and  
5 shipping records in order to show while it's in the glass  
6 waste storage building its storage conditions.

7           Two basic inputs form the production records.  
8 First of all, for every canister there's a canister wallet  
9 which shows the canister that we use, the supply, the fill  
10 height, and various data points in the process as we measure  
11 according to our compliance plan for every canister. So, for  
12 272 canisters, we have 272 canister wallets. Every procedure  
13 that was used showing all of the data. Likewise, we have a  
14 batch wallet for the chemistry side of the process so that  
15 every SME batch, Slurry Mix Evaporator, we can show the  
16 chemical composition, the batch acceptance, and the  
17 radionuclide inventory. So, there's two types of wallets  
18 that form information to the production records.

19           Now, I want to look at the physical process with  
20 these controls in mind. First of all, we have taken a sample  
21 of the sludge in the large tank in the tank farm. That's  
22 what we call the macro-batch. We have a complete analysis to  
23 show the chemical and radionuclides used for that macro-  
24 batch. The tank we're now using is about 500,000 gallons.  
25 We go through the various processes and down at the Slurry

1 Mix Evaporator is our hold point. We sample every Slurry Mix  
2 Evaporator batch, and if it does not meet acceptable  
3 standards, it is then remediated. So, we do not process it  
4 further until it passes this hold point. We then feed it  
5 into the melter feed tank and go on into the physical  
6 process. Likewise, we sample the melter feed tank for every  
7 batch and periodically we take a glass stream sample from the  
8 pour stream again to show compliance on every canister and  
9 every batch.

10           That's basically the overview of the process and  
11 the control process.

12           BULLEN: Thank you. Questions from the Board? Carl?

13           DI BELLA: Can you tell me how many canisters one melter  
14 feed batch tank is equivalent to?

15           BROSEE: Oh, let me answer that from the Slurry Mix  
16 Evaporator. Every batch we move forward is about six to  
17 eight cans. The melter feed tank is kept continuously  
18 varying level because the batch is the control point, is the  
19 Slurry Mix Evaporator. We process about six to eight cans in  
20 every chemical batch.

21           BULLEN: I have a quick one. You mentioned that you  
22 check it for a leak rate parameter on the cans after you've  
23 poured them when you put the conical slug in the stop. If it  
24 doesn't pass the leak rate, how do you remediate? Do you go  
25 back and repair the can or do you have to re-pour it into a

1 new container?

2           BROSEE: There's two things that we can check for first.  
3 The first thing we do is we check the gasket of the test  
4 device itself against a known flange because the gasket we  
5 use wears. If that's the failure, then we replace the gasket  
6 and retest the can for acceptability. If the can itself has  
7 failed, then we have what is known as a repair plug. We  
8 actually push the tapered plug and sleeve into the throat of  
9 the canister and we insert the repair plug which is a  
10 straight plug into the can and we re-leak test it. From  
11 there, it passes on.

12           BULLEN: So, most of your failures are in the neck area.  
13 They're not seam welds or side welds on the can?

14           BROSEE: Absolutely. That's correct.

15           BULLEN: If there were a failure of a side weld, you'd  
16 have to go back and chop the can up and start over or how  
17 would you remediate that one?

18           BROSEE: We have not had that condition occur. The  
19 process would be put aside and we would have to build a  
20 repair plan for that canister. Any canister that does not go  
21 through the waste form compliance plan as we have written it  
22 has to have a unique repair plan that is approved by DOE  
23 before we can use that repair. And, that would be one that  
24 we would have to do that with.

25           BULLEN: Okay. As a followup to that question with

1 respect to your storage in the glass waste storage building,  
2 do you monitor can degradation? Do you periodically inspect  
3 or do you have a plan that you will inspect prior to shipment  
4 of how the cans are going to degrade in the environment?  
5 It's a pretty aggressive environment to pour 1150 degree C  
6 glass into a stainless steel container, and after you've  
7 cleaned the surface, it's going to basically be re-oxidized,  
8 but you'll also have some potential for pitting corrosion and  
9 the like in the moist air environment that you have in that  
10 storage building. Do you inspect and how will you document  
11 those kinds of issues?

12       BROSEE: There are three answers to your question.  
13 First of all, we have done several studies just on the  
14 initial design of the building itself. The thickness of the  
15 304 L stainless steel has various corrosion rates and we  
16 already know all that information. That has gone into the  
17 design of the glass waste storage building. The second part  
18 of the answer is we have put several canisters of non-  
19 radioactive glass from the previous test periods in the glass  
20 waste storage building in various locations. So,  
21 periodically, we'll pull the non-radioactive cans out and do  
22 tests on them to see if our calculations, assumptions, and  
23 projections are, in fact, true. Then, before shipment, we  
24 will have to do some verification prior to the final closure  
25 of the storage record.

1           BULLEN: As a followup to that, the cans that you've  
2 poured, thus far, have not had a very high radionuclide in  
3 loading; so, they're not extremely radioactive?

4           BROSEE: That is correct. They are not.

5           BULLEN: So, surface radiation dose is actually an area  
6 that I'm interested in. What comes to mind is the Climax  
7 Mine test where they put fuel assemblies in 304 L in the  
8 ground and they had alternating heaters, fuel assemblies,  
9 heaters, fuel assemblies, and when they pulled them out, the  
10 radiolytic decomposition of the moist air nearby the  
11 stainless steel had enhanced corrosion greatly around those  
12 while the ones that had heaters in them were as pristine as  
13 when they went in. And so, the question I have is in the  
14 ventilated building that you have, as your glass becomes  
15 hotter--meaning more radioactive, not thermally hotter--as  
16 the glass becomes more radioactive, you have the potential  
17 for in a moist air environment in South Carolina, as we are,  
18 for enhanced radiolysis and enhanced corrosion of those cans.  
19 I just wondered if there were plans--and you mentioned there  
20 were--to monitor this, but to monitor actual pour canisters  
21 as opposed to your test canisters?

22          BROSEE: We would not monitor the pour canisters unless  
23 the non-radioactive canister shows some reason why we should.

24          BULLEN: Okay. But, they're in enough of the radiation  
25 field to see the same kinds of effects?

1 BROSEE: That's correct.

2 BULLEN: Okay. Carl, you had another question?

3 DI BELLA: Yes, I do. You mentioned you measure the  
4 glass coming out of the melter periodically. How does that--  
5 what is that period and how do the composition results agree  
6 with your melter batch tank composition?

7 BROSEE: One of the things that we'll cover a little bit  
8 more in detail--of the actual chemistry of the glass. We  
9 take on a requirement one glass sample per macro-batch. We  
10 are taking a little bit more frequent than that right now.  
11 We are taking a melter feed tank sample on every batch, every  
12 batch that we move forward from the Slurry Mix Evaporator.  
13 And, we have the comparison between the MFT and the SME which  
14 is matching very well. We have not completed all of the  
15 information for the actual radioactive glass pour stream  
16 samples yet. That is still ongoing.

17 DI BELLA: I don't understand the units that you're  
18 using. How many samples have you taken of the glass in--

19 BROSEE: For instance, right now, we are on the 60th  
20 batch from the Slurry Mix Evaporator. We have 60 samples  
21 from the Slurry Mix Evaporator.

22 DI BELLA: Okay.

23 BROSEE: We have also 60 samples from the melter feed  
24 tank. Right now, we have approximately seven from the pour  
25 streams themselves.

1 DI BELLA: Okay.

2 BROSEE: The requirement is one per macro-batch.

3 DI BELLA: Okay. My other question was you said you're  
4 using a surrogate for that PHA stream?

5 BROSEE: Precipitous hydrolysis aqueous, yes.

6 DI BELLA: I guess the real PHA stream, you're not yet  
7 feeding to the melter; is that it? And, is that a  
8 significant stream?

9 BROSEE: Right now, because of ITP, the in-tank  
10 precipitant chemical process, the actual radioactive  
11 precipitant side of this is not in service. We did test the  
12 precipitant side during the waste quarrel runs with again  
13 non-radioactive similar to its full degree. So, the salt  
14 cell has been tested to the full range of glass production  
15 that we have written into the waste form compliance plan.  
16 Right now, we're adding acid in order to make the simulant  
17 look like it's receiving the precipitant in the SRAT.

18 BULLEN: Alberto?

19 SAG<sub>γ</sub>IS: Yes. So, what is the fate of the water in the  
20 waste tank then? Like, where does most of the water go after  
21 this process?

22 BROSEE: Where does the waste go?

23 SAG<sub>γ</sub>IS: The water?

24 BROSEE: The water goes back to the tanks in order to be  
25 re-evaporated and then reprocessed again if it's high enough

1 level waste. It's through underground transfer lines that we  
2 send back to that tank farm that's on the other side of the  
3 canyon.

4 SAGYIS: I see. So, no water goes out the stack then?

5 BROSEE: No, sir, it does not. Any of the water is  
6 trained from these things and sent back through underground  
7 inter-area transfer lines to the tank farm.

8 SAGYIS: Uh-huh. So, they--after basically zero  
9 humidity?

10 BROSEE: That's correct. We do have a sand filter. It  
11 has roughly a DF around 200, seismically qualified and again  
12 under-grade.

13 SAGYIS: So, the water will end up basically in the  
14 Saltstone?

15 BROSEE: Well, the water--if it's high-level--in other  
16 words, if it's mixed back, evaporation will actually just  
17 have it removed. If it is part of the filtrate, it will end  
18 up going to Saltstone as the low-level waste, grout, or  
19 saltstone.

20 BULLEN: Any other questions from the Panel?

21 (No response.)

22 BULLEN: I actually have one more. I may have missed  
23 it, but you're on target, on schedule, and you've got 49  
24 tanks left. When is the job one and how does that coordinate  
25 with the opening and closing of the repository? I guess

1 that's kind of a loaded question, but when is the job done is  
2 the bottom line?

3       BROSEE: Right now based on the available existing waste  
4 in the tanks, we project around 6,000 canisters to handle the  
5 existing tanks. Right now, our accelerated cleanup plan is  
6 based on the repository starting to take shipments in the  
7 year 2015.

8       BULLEN: We have a little bit of time. Are there any  
9 questions from the audience?

10       (No response.)

11       BULLEN: Well, if not, we'll forge ahead. Thank you  
12 very much.

13               We'll get to more meat and potatoes with our next  
14 presentation which is going to be on the characteristics of  
15 the vitrified waste form. Our presenter is Sharon Marra.  
16 She is the manager of the chemical processing and analytical  
17 division within the DWPF engineering department, and she has  
18 been responsible for the development and implementation of  
19 plans and programs to insure the acceptance of the waste form  
20 as produced.

21       MARRA: Well, good afternoon and welcome to all of you  
22 to the Savannah River area.

23               Before I start, I wanted to give a little  
24 perspective. This Board, the Nuclear Waste Technical Review  
25 Board visited us in February of 1992, and at that time, DWPF

1 was finishing construction and getting ready for the startup  
2 test program. I realize many of the members have changed  
3 since then, but at the time we presented to the Board our  
4 strategy for how we were going to assure an acceptable glass  
5 product and how we were going to collect the records to be  
6 able to prove in the future that we made an acceptable  
7 product. And, just to give a little perspective, when I was  
8 preparing for this presentation, I went back and looked at  
9 what we presented to the Board at the time. And, while we  
10 don't plan on getting in as much technical detail today, the  
11 program has really stayed fairly stable over that time. I'll  
12 talk a little bit about our startup test program and how we  
13 proved the program was adequate and how we're using it today.  
14 I just wanted to point that out. There's always changes in  
15 implementation and operating procedures and things like that,  
16 but over that time period, we've really had a fairly stable  
17 approach and it has proved successful.

18           Neil talked about the waste acceptance process and  
19 the waste acceptance product specifications. I want to just  
20 give a quick overview from the glass perspective of what  
21 those require us to report and I'll get into a little bit of  
22 data later on. We're required to report the chemical  
23 composition of the glass on an oxide basis. We're given a  
24 limit of elements that are present at greater than .5 wt%.  
25 We're also required to report the radionuclide content.

1 Those radionuclides that contribute to greater than .05% on a  
2 curie basis to the total inventory at any time up to 1100  
3 years after production, and they also have to have a half-  
4 life of greater than 10 years. We also report the uranium  
5 and plutonium isotope content to meet International Atomic  
6 Energy Agency requirements. And, finally, and we'll talk a  
7 little bit more about this report, Product Consistency Test  
8 results and I'll talk about what that is. That's really the  
9 limit on insuring glass acceptability. We have to verify in  
10 our records, as Neil talked about, that we have made an  
11 acceptable product.

12           There's a couple other requirements I'll just  
13 mention that we're required to report at the time of  
14 shipment. Again, the previous ones were at the time of  
15 production. In other words, today when we finish the can  
16 today and these will be whenever we ship a particular  
17 canister. And, those are dose rate and heat generation. You  
18 can see the limits there. A limit of 1500 watts per canister  
19 and I'll talk in a little while about where we are compared  
20 to that limit. And then, the dose rates, the gamma and  
21 neutron dose rate limits.

22           Jumping back a little bit really where Neil  
23 finished off on glass product control, the requirements that  
24 we have require that we control our process so that our glass  
25 is better than a benchmark glass. And, that benchmark glass

1 referred to here is the environmental assessment glass. It  
2 was a glass that was used in the initial environmental  
3 assessment prior to the construction of DWPF. Our limit is  
4 based on a product consistency test which is an ASTM  
5 certified test. It's a crushed glass leach test. We use  
6 that to insure that we produce an acceptable product and a  
7 consistent product time after time. The way the test works  
8 is the glass is crushed, put into ASTM type water, heated at  
9 90 degrees C for seven days, and then we analyze the  
10 leachate. We look for the elements of lithium, sodium, and  
11 boron. And, previous work over the years has shown that  
12 these three elements leach out the fastest and are a good  
13 representation of what the quality of the glass is.

14           In order to meet that requirement, DWPF developed  
15 this glass product control program and that's the program I  
16 mentioned earlier that we spoke to you about five years ago  
17 or so and that we demonstrated through out startup test  
18 program. What it basically entails--and Neil talked a little  
19 bit about this--is that we control our feed composition at  
20 the last feed preparation vessel; in other words, the last  
21 vessel where we can make changes to it. That's the Slurry  
22 Mix Evaporator that you saw in Neil's slides. The program  
23 also requires that we provide the documented evidence that we  
24 have done this and again that gets into the wallets and the  
25 production records. Also, that the program be robust enough

1 to handle changes and I'll have a couple of slides on this of  
2 how we demonstrated that during our startup testing in our  
3 waste qualification runs.

4           Just very quickly not to get into a lot of detail,  
5 we talk about controlling the process and controlling the  
6 product. We have three extremes; the frit itself, the PHA  
7 which as Neil said we're simulating right now, and then the  
8 sludge. We have to make a feed material that will meet all  
9 of our constraints. One of our constraints is this product  
10 consistency test durability. We insure that by controlling  
11 the chemical composition and we have a correlation of that  
12 test to the chemical composition. We also have constraints  
13 for processing prospective viscosity. And, again, those are  
14 based on chemical composition and there's also some glass  
15 solubility limits. When you take all those limits together  
16 and you take these three feed streams, you get a region which  
17 represented here by that dark region of--if we're anywhere in  
18 that region, we're going to meet all these constraints and be  
19 able to process it and produce an acceptable glass.

20           Now, because we analyze the composition, we  
21 recognize that there are uncertainties associated with that  
22 composition; variability in our laboratory, things like that.  
23 So, we make the window a little bit smaller by taking into  
24 account those analytical uncertainties so we have a very high  
25 confidence that our product will be able to be processed

1 through the system, as well as meet the durability  
2 requirements.

3           Now, I mentioned these waste qualification runs.  
4 Just to give you a quick idea of what we did during those  
5 runs, this FA-13 designated our initial chemical operations.  
6 We prior to that used startup frit in the melter to get the  
7 melter heated up and we produced two batches of feed at that  
8 time and fed those to the melter and flushed it through the  
9 melter. The feed that we used was representative of really  
10 all the waste in the tank farm. We called it a composite  
11 feed. These were simulants. This was a simulated sludge and  
12 a simulated PHA that we used at the time.

13           Once we did that, we got all the systems ready. We  
14 had the startup flushed out of the melter and we began these  
15 waste qualification runs. And, during that campaign, we  
16 produced 56 canisters of simulated waste glass. The majority  
17 of these canisters were destructively examined and I'll show  
18 you some of the results of that. What we did during each of  
19 these campaigns, as we called them, is we buried the feed  
20 drastically. We started out with again a composite feed, but  
21 instead we doped it with a neodymium tracer. The purpose of  
22 this was to track that neodymium through our process and  
23 through the melter to determine whether we had a plug flow  
24 situation in the melter where you'd have step changes in  
25 composition or whether we really had a well-stirred tank

1 which is what the expectation was. As it turns out, we had a  
2 well-stirred tank, a slow increase in the neodymium content  
3 in our glass product.

4           The next campaign--and I show here this transition  
5 from a composite type, sort of run-of-the-mill feed material  
6 to a high iron feed which led to a low viscosity feed  
7 material. We were focusing on viscosity here because  
8 viscosity was the one parameter that we wanted to make sure  
9 that we wouldn't get any segregation of the feed. We wanted  
10 to make sure that we could make these transitions from a  
11 baseline to a low viscosity and vent it back to a high  
12 viscosity and we could still control our product, this  
13 program still worked, and we could still make an acceptable  
14 glass. After that high iron feed, we jumped to a high  
15 aluminum feed which was a high viscosity feed. Again, that  
16 was our most severe transition. During that time, we ran  
17 through the whole process, as Neil explained it, watched  
18 melter behavior, and then destructively examined the glass to  
19 make sure that we were still in compliance.

20           Then, the final campaign was back to a blend type  
21 feed that was similar to what we would be producing during  
22 the initial radioactive operations. So, we transitioned  
23 again from a high viscosity back to a baseline type feed. As  
24 I said, we produced 56 canisters during this campaign. These  
25 canisters were sectioned, had windows removed from the wall

1 of the can and glass samples were taken. Tomorrow on the  
2 tour, you'll see the bottom portion of the section of one of  
3 these cans in the lobby when we walk you through the  
4 building. All these glass samples that were taken were  
5 analyzed for this product consistency test I mentioned, as  
6 well as chemical composition.

7           Carl asked the question about how our prediction  
8 was comparing to the actual result. This line here shows the  
9 predicted PCT result, Product Consistency Test, on a gram per  
10 liter basis for boron. These are all the campaigns that we  
11 produced during that waste qualification runs. And, you can  
12 see it's fairly stable. It jumps up a little bit and jumps  
13 down, but really is fairly stable. And, we're down in the  
14 less than one region. Our limit is way up here. It would be  
15 off the chart. If I drew this to scale, it is 16.7. So, you  
16 can see how far away we are from our limit. The other  
17 important thing to point out is this line here is the  
18 predicted PCT value based on our batch analysis. And, you  
19 can see the straight lines indicating that this particular  
20 batch made roughly seven canisters; the next batch maybe made  
21 five canisters. So, you can see that jump up there. And,  
22 they really track fairly well and we did a lot of statistical  
23 analysis on it to show that we were well in control and that  
24 what we predicted was represented by what we actually tested  
25 in the glass.

1           Just to give you a feel for what this glass is made  
2 of and what it looks like on an oxide basis, this is a range  
3 for these major components of the glass of what we expect our  
4 glass to look like over time. This range is based on  
5 analyses that have been done of the sludge in our waste  
6 tanks. Neil mentioned our planning process. We do have a  
7 planning process where we designate what sludge batch will  
8 come next, but this range represents what we expect to see  
9 over the lifetime of the facility. As you can see, iron is a  
10 major component of the glass. There's some sodium in there,  
11 aluminum, things like that.

12           Now, jumping to radioactive operations, where are  
13 we today? Again, back to this Product Consistency Test, this  
14 particular column here--and I got the results for the first  
15 43 radioactive feed batches--this is the average of what we  
16 predicted for each of these feed batches, roughly .7 grams  
17 per liter. As Neil mentioned, we pulled about seven pour  
18 stream samples, tested four of them for Product Consistency  
19 Test, and the actual results are approximately .9 gram per  
20 liter; again very, very close within statistical  
21 significance. Again, you can see the limit there; far, far  
22 above where we're operating our facility.

23           Now, over on the radionuclide inventory side, there  
24 was some questions on where we are now. On a curie content  
25 per canister, this represents the major isotopes, the DWPF

1 design basis glass. This design basis glass is--you can  
2 think of it as a binding type case. It was used for some of  
3 the design of the DWPF facility from the perspective of  
4 shielding and environmental considerations. As you can see,  
5 the curie contribution there, fairly evenly split between  
6 strontium 90 and cesium 137. Then, if you look at where we  
7 are today, this Tank 51, that's the tank we're feeding out of  
8 now. So, this is information of the glass we're producing  
9 today. You can see the strontium 90 is the higher  
10 contributor for curie content. That's because we don't have  
11 this precipitant feed.

12 Yes?

13 DI BELLA: Could you clarify what Tank 51 means? Is  
14 this the batch melter feed tank?

15 MARRA: No, I'm sorry. This is the waste tank in the  
16 tank farm, that 500,000 gallons that Neil mentioned.

17 DI BELLA: Okay.

18 MARRA: That's the tank that we happen to be feeding  
19 from today. The melter feed tank batches are--we've made  
20 approximately 50 of those to date. Yeah, as I mentioned, the  
21 strontium 90 is the higher curie contributor because we don't  
22 have that precipitant feed on line. The precipitant feed  
23 will contribute most of the cesium 137 to our glass.

24 BULLEN: Before you leave that one, all the  
25 radionuclides that are listed on the bottom, are those that

1 are required because they are greater than the .05% of the  
2 curie concentration with half-lives greater than 10 years?

3 MARRA: That's correct.

4 BULLEN: Okay.

5 MARRA: That's correct. Yeah, I just listed the major  
6 ones here, but these are the other ones that we're required  
7 to report, as well.

8 I believe there was a question earlier about heat  
9 generation. Where are these canisters in comparison to  
10 design basis and limit? The DWPF design basis glass again  
11 led to 750 watts per can. Again, that's a bounding case  
12 extreme for waste at Savannah River Site. As you can see,  
13 the canisters we're making now, as we mentioned earlier, are  
14 on the cool side, only about 4 watts a can. This next batch  
15 of sludge we're receiving at Tank 42 indicates the large tank  
16 in the waste farms jumps up a little bit, but still very much  
17 on the low side. The limit of 1500 watts per canister, the  
18 SRS DWPF canisters, unless some unexpected new waste stream  
19 comes in, wouldn't approach that limit, at all. So,  
20 typically, once we get our precipitant on line and we get  
21 processing, we'll probably be in the 200 to 300 watts per  
22 canister range.

23 I thought I'd just finish up with a slide that's a  
24 graph that's representative of one of our waste qualification  
25 documents on decay rates on a curie per canister and a watts

1 per canister basis up to 1100 years after production. This  
2 is the design basis glass. So, this is the higher heat  
3 generating waste stream. But, the canisters we're making  
4 now, although they would start out much lower, would follow  
5 the same type of decay pattern.

6           So, that's all I have. If there's any other  
7 questions, I'd be happy to answer them.

8           BULLEN: Questions from the Board?

9           ARENDDT: Are you subject to IAEA inspections?

10          MARRA: I'm not sure I can answer that. Bill? No, not  
11 yet.

12          ARENDDT: Process control measurements versus quality  
13 control measurements. First off, are you in Neil's  
14 organization or what organization are you in?

15          MARRA: Neil is our program manager at DWPF. I'm with  
16 the engineering organization within DWPF, and we support  
17 Neil.

18          ARENDDT: What I'm trying to get at is differentiating  
19 between process control measurements and quality control  
20 measurements. Are they one in the same or--

21          BROSEE: There's a different program associated with  
22 both. Besides the engineering evaluation of the glass  
23 samples and the product consistency, we also have a quality  
24 assurance group who independently look at my production.  
25 They look at the wallets. They are actually an independent

1 arm under the quality assurance program for RW0333P.

2       ARENDR: Okay. On the canisters, I suppose the  
3 canisters are your design. Do you have any vendor inspection  
4 or do you inspect the canisters at the vendor? Do you do any  
5 in-house inspection of the canisters prior to use?

6       BROSEE: There's all of the above. First of all, we do  
7 have an inspection program that is done by the vendor at the  
8 vendor. And, we also have an independent quality assurance  
9 program that looks at his program while he's doing that. We  
10 then do a receipt inspection to show that there was no damage  
11 or change during the transportation from the fabrication  
12 location to our site. Before we actually move them into the  
13 canyon--and you'll see this tomorrow on the tour--we actually  
14 have a canister receipt area where we inspect for  
15 cleanliness, foreign material, and various inspections that  
16 we go through before we even put them into the canyon for  
17 processing.

18       ARENDR: And, I was going to ask earlier, but your final  
19 closure weld, do you do any inspection on the final closure  
20 weld?

21       BROSEE: We do visual only. We have done various burst  
22 testing on test canisters and also on test nozzles which have  
23 proven the parameters of our upset welding. Then,  
24 periodically, we do current tests in order to show that  
25 nothing has changed in that process.

1 ARENDT: Okay. Thank you.

2 BULLEN: Other questions from the Panel?

3 DI BELLA: From a repository point of view, I think it  
4 would be somewhat important to know on any given canister how  
5 well the radionuclide composition reported in the canister  
6 wallet matches with what's actually in the canister. I don't  
7 quite see from what's been presented how that is established.  
8 Could you elaborate?

9 MANNA: Yeah. What we're doing right now, these glass  
10 pour stream samples that we've pulled, obviously we couldn't  
11 test that when we were in startup testing like we could the  
12 composition. These glass pour stream samples that we've  
13 pulled now are being extensively analyzed for radionuclide  
14 content at the Savannah River Technology Center at SRS. We  
15 are comparing those results back to the melter feed--what we  
16 would report from the melter feed to make sure that we're in  
17 synch. And, we have a commitment to update our waste  
18 acceptance documentation to reflect any uncertainties or  
19 errors associated with that result. But, we couldn't do that  
20 until we got into initial radioactive operations.

21 BROSEE: One additional item. The production record  
22 that actually is shipped to the repository will have the  
23 information of the batch it came from, the radionuclides, as  
24 well as the chemical composition, the results of the Product  
25 Consistency Test, as well as if it had a pour stream sample,

1 the results of the pour stream sample, in that document.

2 BULLEN: As a followup to that one, you haven't noted in  
3 any of your analyses that there is selective segregation of  
4 any of the specific radionuclides. It's pretty much well-  
5 mixed and always coming through or do you see spiked changes  
6 within the same batch of, I don't know, neptunium 237 shows  
7 up in a higher concentration in one than the other?

8 MARRA: So far, we haven't seen that. We've seen a very  
9 homogenous product.

10 BROSEE: In fact, what we did during the WP-14 that  
11 Sharon mentioned, we actually tested for that using the dope  
12 feed to show not only their stay time in the melt pool for  
13 the melter life, but also for the consistency as we would  
14 make that transition from FA-13 to WP-14 test material.

15 SAGYIS: I guess this is a two part question. First,  
16 any idea as to what would be expected rate of generation of  
17 gasses like hydrogen isotopes or helium and the like? And,  
18 the other would be how much of a dead space is left at, say  
19 at the top of the canister when you load it?

20 MARRA: Let me answer your second question first because  
21 I may need a little help from the audience on the first one.  
22 We fill the canisters, our target is approximately 96  
23 inches. The container is 118 inches tall. That is  
24 equivalent on a volume basis of 90 percent of the volume of  
25 the canister.

1           BROSEE: The total volume of the canister is 26 cubic  
2 feet. We fill to 25 cubic feet and we need that extra volume  
3 for when we push the tapered plug and sleeve in for the final  
4 closure at the weld cell.

5           MARRA: And, as far as gas generation, we've done some  
6 studies on that and I'm not sure I could quote numbers. I  
7 don't know if Ned Bibler could help me if he's out there  
8 somewhere. As far as gas generation inside a canister, I  
9 assume you're referring to?

10          SAGγIS: Yeah.

11          MARRA: I'm going to let Ned help me out on that one.

12          BULLEN: Please, identify yourself?

13          BIBLER: Ned Bibler, Savannah River Technology Center.  
14 The only gas that's produced of any significant quantity in  
15 the canister is helium from the alpha particle  
16 neutralization. Tests have shown from helium implantation  
17 and from dope class with curium 244 that all of that helium  
18 that's produced will remain within the interstitials of the  
19 glass and not go up to the open space and no hydrogen. And,  
20 there's no water, very little water.

21          MARRA: Thanks, Ned.

22          DI BELLA: You showed that the canisters you've been  
23 making, so far, are very low wattage implying very low  
24 radionuclide loading. Will you be able to catch up and sort  
25 of recover for this or does this mean that many more

1 canisters are going to go to the repository than originally  
2 had been planned?

3       BROSEE: The original 6,000 canister projection was  
4 based on the sludge which is a controlling item in the mix  
5 feed of the coupled operations. We do not expect to see a  
6 loading problem, although that is one of the areas that we  
7 will have to look at as we then bring on the coupled  
8 operations. But, the original projection from the existing  
9 waste was based on the sludge volume, not the salt volume.

10       MARRA: And, we're still putting into our canister 700  
11 to 800 gallons of sludge per canister. It's just that this  
12 happens to be older sludge in addition to what Neil said.

13       BULLEN: This is a final question, I think, before the  
14 break. I guess it's a question of interfacing between the  
15 people at the Mountain who are doing performance assessment  
16 analysis and the types of inventories that you're producing.  
17 How do you provide communication? I know your wallets and  
18 your notebooks and all that are going to go in association  
19 with this. But, how do you provide communication to people  
20 now in the TSPA-VA and those kind of things with respect to  
21 inventories and potential release rates and mechanisms  
22 associated with that? Is there good communication or could  
23 there be room for improvement?

24       MARRA: Well, let me comment on one thing. Some of our  
25 requirements that I didn't talk about, in addition to all

1 those reporting requirements, we have a requirement for  
2 projecting what we expect to produce. We've done that and  
3 presented that in our waste qualification documentation.  
4 That was extensively reviewed by DOE and DOE is working  
5 together to provide that information to the repository folks.  
6 So, someone else might want to comment on that, but I think  
7 we've provided the information and the communications are  
8 fairly good in that area.

9 BULLEN: Any other questions from the Board or Panel?

10 (No response.)

11 BULLEN: Questions from the audience?

12 (No response.)

13 BULLEN: Okay. Now, I'm going to take a risk here.  
14 It's a very nice day and there's a nice river walk right  
15 outside the door there. I want everyone to promise to be  
16 back here at 3:00 o'clock which is about 22 minutes from now,  
17 and we will reconvene.

18 Thank you.

19 (Whereupon, a brief recess was taken.)

20 BULLEN: Could everybody take their seats, please, and  
21 I'll ask the Board members to come back up to the front so we  
22 can get started for the final session of today.

23 Our final two presentations deal with the issue of  
24 the immobilization of surplus weapons-grade plutonium. We're  
25 going to have a talk from Bill Danker, first. Bill is the

1 plutonium immobilization project lead for the DOE's Office of  
2 Fissile Materials Disposition, and he's going to talk about  
3 the immobilization of surplus weapons-grade plutonium.

4       DANKER: Good afternoon. I'm pleased to be here to talk  
5 about one track of this nation's dual-track strategy for  
6 securing surplus plutonium. With me today is Tom Gould who  
7 is at Lawrence Livermore National Laboratory and who heads up  
8 the immobilization research and development team. Tom will  
9 describe the immobilized form and how we're supporting  
10 repository analyses by the Office of Civilian Radioactive  
11 Waste Management and working closely with them. If you sense  
12 I'm not getting to an area where you have a question, please  
13 interrupt. I'm here to address topics which are of issue or  
14 concern to you.

15               This slide takes care of the toastmaster's  
16 requirement to tell how much you're going to tell them. I'll  
17 briefly review why we're trying to make big, heavy,  
18 radioactive objects. I'll describe progress since I last  
19 briefed members of the Board in January '96. Then, I'll give  
20 you a quick overview of where the immobilization project  
21 stands today. As I mentioned, Tom will close with a closer  
22 look at the form.

23               Four years ago, a number of reports, most notably  
24 one by the National Academy of Sciences, focused on the  
25 proliferation danger posed by plutonium being removed from

1 warheads at an unexpected rate. The good news was arms  
2 control was working; the bad news was that Russia and the  
3 United States quickly needed long-term plans to secure huge  
4 quantities of weapons and usable plutonium. The United  
5 States established the Office of Fissile Materials  
6 Disposition later that year to focus on reducing the global  
7 nuclear danger posed by this material. I'll describe in a  
8 minute earlier this year the President and the responsible  
9 Government agencies decided that this country should use both  
10 existing reactors and immobilization to secure this material.

11 I hope this is right because I carved up a slide  
12 that had both plutonium and uranium on it. It should be  
13 accurate. The United States has declared about 50 metric  
14 tons of plutonium surplus to national defense needs, the bulk  
15 of which is at Pantex, Rocky Flats, and Hanford. This slide  
16 is a reminder of the plutonium quantities and locations.  
17 Tom, I think, will show you one a little later that gives you  
18 a little bit of a different cut at it in different  
19 categories, and I've got a backup slide that will show how  
20 you rationalize the 52.5 shown here and the 50 that he has.  
21 But, the bottom line is it's about 50 metric tons. While I  
22 haven't seen a comparable slide released by Russian  
23 authorities, I understand the declaration of total surplus  
24 quantities is due shortly.

25 I have to say I appreciate the Panel not scheduling

1 in a blizzard in D.C. this week. The last time I briefed the  
2 Board, the snow complicated things. Since that time, the  
3 Administration has decided on the two-track approach for  
4 plutonium disposition. This decision was supported by three  
5 principal legs; technical, environmental, and  
6 nonproliferation analyses. I chickened out on lugging all of  
7 the reports with me, but the references here are available  
8 upon request from our office. Also, earlier this year, we  
9 issued a notice of intent, I think, in May to prepare the  
10 next set of environmental documents and therein said for  
11 immobilization we prefer to use the canister technology of  
12 the Savannah River Site. Principal attraction in doing it  
13 here is that the Defense Waste Processing Facility exists, as  
14 you just heard, and is producing high-level waste canisters.  
15 I don't think I'll dwell on this slide given the focus of  
16 this meeting. Note that it clearly shows that both plutonium  
17 disposition tracks result in forms which are intended to go  
18 to the repository which this Panel reviews.

19           The principal project driver is nonproliferation.  
20 We're trying to make this plutonium as unattractive and as  
21 inaccessible as that in commercial spent fuel. Making it  
22 big, heavy, and radioactive is one way to do that. I have to  
23 say this as an aside and as the son of a minister that some  
24 of the more theological discussions I've ever gotten into is  
25 on the concept of the spent fuel standard which is, in fact,

1 not a standard, but a perspective. But, the words I gave you  
2 right now are the best take on what we mean by that measure  
3 of proliferation resistance; make this plutonium as  
4 unattractive and inaccessible as that in commercial spent  
5 fuel.

6           A related driver is urgency. I consider the 2005  
7 date that you see here an aggressive schedule. While we have  
8 to develop a process that results in reliable, quality, cost-  
9 effective production, a key part of our development program  
10 is focused on providing characterization data to support  
11 acceptance for eventual disposal in the repository.

12           Pictures make it easier to understand what the can  
13 and canister looks like. If anyone hasn't seen a DWPF  
14 canister and you probably have given the previous briefings,  
15 that's the one in the middle between Leonard and Gene.  
16 Currently, we expect to put 28 cans, three inches in  
17 diameter, about 21 inches high inside each canister. These  
18 would be supported at--I'm sorry?

19           SPEAKER: 28 cans?

20           DANKER: 28 cans, yeah. 28 cans, three inches in  
21 diameter, 21 inches high inside each canister. These would  
22 be supported at four levels, seven cans in a circular ray  
23 given the current configuration. The picture at the right  
24 actually is one from one of the cold pours back January of  
25 1996 and actually shows a spiral array of about eight cans.

1           Since the January record of decision retained the  
2 option of all 50 metric tons of plutonium being immobilized,  
3 this project for now is using that as a planning assumption.  
4 There's a range of feed material from converted pits to  
5 stabilized impure oxides to fuel. So, material conversion is  
6 important. We're assuming that there are enough canisters  
7 with enough fission products to support the mission. And,  
8 again, as an aside, we don't mind that they're running on  
9 sludge at the moment. As an example, if each can contained  
10 roughly a kilogram of plutonium, then about 175 canisters a  
11 year would be needed for the full 50 metric ton case. In a  
12 hybrid case, you have 18 metric tons going to immobilization.  
13 It would require about 60 canisters a year for the plutonium  
14 mission. We're currently expecting that the spent fuel  
15 standard requires 100r/hr 30 years after the canisters are  
16 poured. As the next speakers will discuss, we'll need to  
17 qualify for repository disposal.

18       DI BELLA: How much plutonium per can did you say?

19       DANKER: Current guess is about a kilogram per can--

20       DI BELLA: Okay. A can is a little can?

21       DANKER: Right.

22       DI BELLA: Okay.

23       DANKER: So, 28--right, 28 kilograms per big canister.

24 We try and stick--we're not always successful, but we try and  
25 stick to the nomenclature of cans being the small ones.

1           It's a busy slide, but I'll stick to a few basic  
2 impressions I'd like to leave with you on this schedule.  
3 There are essentially two long poles in the tent. One is  
4 focused on activities needed to qualify for repository  
5 disposal which are shown in red at the bottom. I noted with  
6 interest Neil's reference earlier to the waste acceptance  
7 process being just as involved as the process itself, and  
8 that's probably a good perspective for us. Based on my time  
9 in that office, it is a complicated activity. The other is  
10 composed of activities needed to define the process and  
11 construct and start the facility. We're currently our way  
12 through the tasks identified in the latest immobilization  
13 plan describing development and characterization work needed  
14 to start up in 2005.

15           We might want to leave that one up, and if you  
16 could switch over--yeah. It's dangerous. It might stimulate  
17 questions. We're currently focusing our work on ceramic  
18 forms. In July and August of this year, data that had been  
19 gathered on both glass and ceramic forms were evaluated in an  
20 intensive process described by this busy slide. A technical  
21 evaluation panel identified discriminators between the forms  
22 using a set of technical criteria, but didn't try and pick  
23 the best form. Livermore, as head of the development team,  
24 then completed an integrated assessment and drafted a  
25 recommendation recommending ceramics. An independent peer

1 review panel examined those reports, met with the experts,  
2 and issued a letter report confirming that both forms could  
3 do the job and that ceramic had some advantages. Since  
4 September, our office has focused on the ceramic form.

5           More than a year of work into producing data to  
6 support this downselect process. As noted here, advantages  
7 for ceramics were identified in proliferation resistance in  
8 that basically less known about ceramics. Perhaps, Tom when  
9 he gets to this discussion of forms may touch on some of the  
10 other aspects of that. Worker dose was a discriminator and  
11 cost-effectiveness. A large part of that had to do with the  
12 higher density for the ceramics requiring fewer canisters and  
13 one compliment of that is saving on costs for repository  
14 disposal.

15           Since we're here in the neighborhood of the  
16 Savannah River Site, this is a reminder of why we've  
17 identified it as our preferred site for immobilization. As I  
18 noted earlier, DWPF is the key reason. In the future, the  
19 APSF also offers potential for synergy, storage, and so on.  
20 Westinghouse Savannah River Company is a key player on the  
21 immobilization team. Process experience here will help us  
22 expedite demonstration and startup. Right now, we're  
23 planning to produce prototypic cans. Current schedule 2000  
24 and, hopefully, a canister or two in 2001. Clearly, this is  
25 tied to our qualification process.

1           I'm going to allow enough time for Tom to get into  
2 form, process for producing the form, and discussion on our  
3 work in support of RW's analyses, but as a segue to Tom, this  
4 chart highlights the current project team structure. I'll  
5 resist the temptation to dive into detail here and turn it  
6 over to Tom to provide more detail. I might mention he is  
7 sort of a walking example of technology transfer in that he's  
8 at Livermore and heads the development team, but is on loan  
9 from Westinghouse Savannah River Company which is where we  
10 prefer to transfer this technology.

11       BULLEN: Any questions at this point or should we  
12 proceed?

13       CRAIG: Let me ask a couple. Maybe Tom is the one that  
14 wants to answer them, but I'd like to understand how--what is  
15 the source of radioactivity? How does it compare with waste  
16 after 1,000 years and what about retrieval since the  
17 plutonium is presumably still there as some form of plutonium  
18 that you'll tell us? Is it plutonium oxide?

19       DANKER: Okay. Let me read your question back to you.  
20 Your question is on the source of the radioactivity and also  
21 questions about retrieval of the plutonium form from the  
22 repository. Is that correct?

23       CRAIG: Yeah, and how the radioactivity compares with  
24 reactor waste after 1,000 years time frame rather than the  
25 shorter time frame that you mentioned?

1           DANKER: Okay. That brings to mind decay curves for--  
2 yeah. Let me try this and then hand it off. The bottom  
3 line, one driver to go to can and canister was to simplify  
4 the recipe, if you will. Isolating the external barrier  
5 simply implies you're doing the immobilization of plutonium  
6 in small cans and you're relying on the high-level waste that  
7 was discussed earlier by Neil and others within the canister  
8 to provide your radiation barrier. So, it's a pretty easy  
9 answer to the first question in that the radiation barrier  
10 comes from the high-level waste glass and we simply provide a  
11 first stage immobilization ram in a framework within the  
12 canister and then put it under the DWPF melter and do a pour.  
13 One of the differences is you have volume displacement. So,  
14 it ends up with additional canisters. But, the source of the  
15 radioactivity is, in effect, the high-level waste glass.

16                 In terms of retrieval, I may need help on the decay  
17 curves, but, Tom, are you going to be getting to that one?

18           GOULD: No, but I can cover it.

19           DANKER: Yeah.

20           GOULD: When I talk about the form.

21           DANKER: Yeah. The spent fuel standard again is--I  
22 think, Sharon had some decay curves before for the high-level  
23 waste glass that you can use to pick off the curies per  
24 canister and so on that would give you the decay curve. But,  
25 I can tell you that kicks you back to the spent fuel

1 standard. 100r/hr, 30 years after fabrication, is a fairly  
2 arbitrary point and is tied to a range of considerations  
3 including what is considered self-protecting in NRC and IAEA  
4 regimes and so on, but it's an art and not a science. I  
5 danced a little on that last one. So, you may want to hit  
6 Tom when he stands up here.

7 CRAIG: Yeah, it was a longer time frame than I was  
8 focusing on.

9 DANKER: Yeah, okay.

10 GOULD: I'll try to cover that.

11 BULLEN: Good. Any other questions at this point?

12 DI BELLA: Do you think the glass they're making right  
13 now would be radioactive enough to provide this protection  
14 that you're talking about?

15 DANKER: No, no.

16 DI BELLA: Okay.

17 DANKER: Sludge. What they're currently pouring is  
18 sludge.

19 GOULD: Yeah. No, it would need a little more cesium  
20 content in there to provide a level of protection comparable  
21 to spent nuclear fuel.

22 BULLEN: Right now, the cesium doesn't come in because  
23 they're just putting strontium is essentially--so, you need  
24 both cesium and strontium there to get you the high gamma  
25 field to self-protect?

1 GOULD: Yeah, the cesium is what's really going to  
2 protect you over the next several decades.

3 BULLEN: 300 years, yeah.

4 GOULD: Yeah, 300 years.

5 BULLEN: Okay. Paul, did you have any more questions or  
6 do you want to move right into--

7 CRAIG: No.

8 BULLEN: The segue got a little disrupted, but we'll let  
9 Tom step in and follow up on the waste form.

10 DANKER: By the way, he has with him a surrogate ceramic  
11 puck, 20 or 21 of which go into the cans. If he leaves any  
12 of these with you, don't do what I did at the airport. When  
13 it comes up, you know, and they ask you what it is, don't get  
14 complicated on the discussion. Don't say it's a surrogate  
15 ceramic for the plutonium disposition because the word they  
16 pick out of that sentence is plutonium.

17 GOULD: Actually, paperweight works well at the  
18 airports.

19 What I want to do right now is to give you sort of  
20 a brief overview of what the form is, very briefly on how  
21 we're going to make it, and talk a little bit about the  
22 development program we have in place primarily focusing on  
23 providing information to the repository analysis.

24 Let me start and at least summarize some of the  
25 materials that we're going to need to incorporate within the

1 ceramic form which will then go into the can and canister  
2 larger form. We will be focusing primarily on what are  
3 called the impure plutonium materials that are coming both  
4 from the weapons program, as well as from the fast reactor  
5 testing program in this country. In the pure dual-track  
6 approach, the clean material, primarily material returned  
7 directly from weapons, would be converted to mix oxide  
8 reactor fuel and that fuel then radiated in existing power  
9 reactors. That comprises about 32 to 33 metric tons of  
10 plutonium and it's very pure plutonium with just a little bit  
11 of gallium in it.

12           The other materials would require significant  
13 purification processing in order to convert that material to  
14 an acceptable oxide for fuel and this is one of the  
15 advantages of the immobilization process is, in effect, we're  
16 not doing any purification. We will be converting all of  
17 these materials, if they are not now oxides, into an oxide  
18 feed for the ceramic form. They include various alloys, as  
19 well as uranium, plutonium, alloys, and oxides, some impure  
20 plutonium oxides that are predominately plutonium but contain  
21 residues from the various processing steps that we'll use in  
22 the weapons program.

23           It also includes fuel that was used in the zipper  
24 reactor at Idaho, Argonne West in the fast reactor program,  
25 as well as un-irradiated fuel that had been prepared for

1 radiation in the Fast Flux Test reactor at Hanford. Coming  
2 with the plutonium in this fuel is going to be approximately  
3 17 metric tons of predominately uranium 238.

4           The plutonium form itself, the ceramic form, is  
5 based on titanate minerals. The early work in developing the  
6 form was performed in Australia by Ringwood in developing the  
7 so-called sin rock form for high-level waste. These minerals  
8 primarily that we're using are going to be pyrochlore and  
9 zirconolite and also a little bit of brannerite. The  
10 chemical formula for these is basically  $A^{(+2)B(+4)}Ti_2O_7$ .  
11 Basically, the plutonium, the uranium, and the hafnium will  
12 substitute indirectly to the B site. This is normally  
13 occupied by zirconium. Long-lived mineral phases, pyrochlore  
14 and zirconolite, have been around for a billion years or so.  
15 They have contained thorium, as well as uranium. So,  
16 there's some long-range data on these forms. The A site is  
17 occupied primarily by calcium. Gadolinium that we want to  
18 put in there again is another neutron absorber with hafnium.  
19 It's a +3 element and it will partition between the A and  
20 the B sites.

21           This is the ratio of the different oxides in the  
22 primary form. There are a lot of impurity materials, cations  
23 such as iron, chromium. There will be a little bit of moly,  
24 some aluminum. There's also going to be a little bit of  
25 silica coming in some of the plutonium materials. These will

1 be contained in the form at percentages that are less than  
2 two percent by weight. Some of these will substitute  
3 directly into the titanate base phases and in other cases,  
4 such as silica, there will be a silicate phase that forms.

5           We've got some preliminary data indicating that the  
6 thermodynamics at the formation of the ceramic pretty much  
7 force the plutonium and the uranium, as well as hafnium and  
8 gadolinium, into the primary titanate base phases. So, we  
9 really are doing a pretty good job at this point of  
10 associating with the fissile materials and appropriate  
11 quantity of neutron absorbers. We're putting in hafnium at  
12 about a one-to-one atomic ratio to plutonium. U-238 is going  
13 to be in there at about two-to-one to plutonium, and  
14 gadolinium, a little less than one-to-one with plutonium.

15           The can and canister form itself will be comprised  
16 of, as Bill indicated, 28 cans of--they're about 21 inch high  
17 cans; oh, about three inch OD cans containing 20 of these  
18 ceramic pellets. The pellets are about one inch thick and  
19 two and a half inches--2.6 inches in diameter. Each pellet  
20 contains about 50 grams of plutonium. So, there's going to  
21 be about 1kg of plutonium per can. The cans will be  
22 distributed in sort of a circular array, four layers of seven  
23 columns. The volume occupied by the cans and the support  
24 structure for the cans is estimated right now at about 12  
25 percent of the free volume within the canister. We have

1 gotten some preliminary results on two early pours done  
2 during the startup of the Defense Waste Processing Facility  
3 indicating that the glass fills effectively the entire volume  
4 with a slightly different configuration. We had a 20 can  
5 configuration that we poured. We are now doing analytical  
6 modeling for this configuration that indicates that we should  
7 get a complete occupancy of high-level waste glass within the  
8 canister. Later, we will be doing some pours with surrogate  
9 materials to confirm that.

10           In terms of the number of canisters that will be  
11 affected for the so-called 17 metric ton case which will  
12 probably involve about 18-1/2 metric tons of plutonium, for  
13 that case--and that will be all of the impure plutonium  
14 materials--we're going to occupy 635 canisters of glass and  
15 there will be 77 extra canisters that will have to be  
16 generated as a consequence of the volume displacement. In  
17 the case of all of the materials coming to immobilization,  
18 the 50 metric ton case, that would occupy a little over 1700  
19 canisters and there would have to be produced an additional  
20 210 canisters of high-level waste to accommodate the volume  
21 loss.

22           The process for making the canistered forms really  
23 can be considered in three parts or three stages. The first  
24 is feed materials characterization. Here, all of the various  
25 feed materials will be converted to an acceptable oxide feed

1 for the ceramic formation process. This head end will be  
2 accommodated within the same facility as what we're calling  
3 first stage immobilization.

4           In the first stage of immobilization, we basically  
5 are going to be performing a MOX like fabrication process  
6 with some small changes, but primarily a cold press and  
7 centering operation. The centering temperature is 1350  
8 degrees Centigrade for the ceramic form. At that  
9 temperature, you actually get the chemical reaction that  
10 takes place primarily in the solid state among all of the  
11 oxides occurring in the mixture. It requires basically a  
12 milling and granulation step, cold pressing, followed by  
13 centering, and we will be doing some individual  
14 nondestructive analysis on each of the pellets and then those  
15 will be basically stacked within the cans.

16           The cans will be brought out of the glove boxes in  
17 an operation that seals the cans. The cans then will be  
18 loaded in the second stage immobilization process, or at  
19 least leading to it, into basically an empty DWPF waste  
20 canister in a rack. These operations will be performed  
21 basically either in a new facility that will be joined to the  
22 actinide packaging and storage facility that is being  
23 constructed at the Savannah River Site in F-Area or the  
24 operations will be performed basically in areas of 221-F.  
25 Those are two facility options that the Department of Energy

1 is currently evaluating.

2           From the F-Area, the canisters loaded with  
3 plutonium will be transported to the DWPF facility. Those  
4 canisters will be taken into the processing canyon facility  
5 just like the other canisters except for a little added  
6 security and appropriate safeguards. The glass will then be  
7 poured into the canisters and the whole process is just like  
8 described earlier.

9           During the past year, we've gone through, as Bill  
10 had indicated, a laboratory scale development program in  
11 developing the ceramic form, as well as the glass form,  
12 looking at what the processing conditions would be doing some  
13 rough engineering analyses of the production process leading  
14 to a decision on which of the forms we wanted to develop  
15 further for the final production facility. At this point,  
16 we've got some preliminary information basically on the  
17 characteristics of the ceramic form. We've gotten some  
18 preliminary durability testing data, and over the next two to  
19 three years, we're going to be focusing basically on fine  
20 tuning the compositional aspects of the plutonium form  
21 itself, developing the processing condition envelope for the  
22 production system, and we will be doing some prototype  
23 equipment testing both with and without plutonium depending  
24 on the critical nature of the equipment and whether it's  
25 plutonium dependent. And, we are also going to finalize on

1 the design of the canister form.

2           What I want to do at this point is talk a little  
3 bit more in detail about what we specifically want to do to  
4 support RW so that they have adequate data to provide an  
5 analysis of how the form will perform in the repository over  
6 a long period of time. And, I just remembered your question,  
7 Paul, and I apologize for--let me pick it up at this point  
8 and then I'll talk about the repository aspect of our  
9 program.

10           One of the reasons that ceramic was chosen was that  
11 the plutonium is tied up in the titanate lattice in such a  
12 manner that normal processing that is used for plutonium  
13 materials in Russia, as well as the United States, doesn't  
14 work very well at extracting it from the crystalline lattice.  
15 As a matter of fact, you have to go to a different  
16 processing scheme entirely. So, it makes it a little more  
17 costly and more difficult to extract plutonium. Around the  
18 plutonium, of course, we're putting it in this high-level  
19 waste glass which provides a proliferation barrier at least  
20 for the period that we're going to--you know, between  
21 generating the canisters and emplacing them in the high-level  
22 waste repository.

23           Subsequent to emplacement, of course, the fission  
24 products are going to decay in a few hundred years. Just  
25 like with spent fuel, you're going to have a plutonium mine.

1 Okay? That's the result of decisions made in this country.  
2 In the case of spent fuel, the plutonium is going to be  
3 incorporated in a ceramic material with uranium. In the case  
4 of this form, it's going to be incorporated in a ceramic  
5 material that's going to be surrounded by glass that will be  
6 surrounded by a canister. The concentration of plutonium in  
7 the whole high-level waste canister is going to be, oh, a  
8 little bit over a percent; somewhere between 1 and 2 wt%.  
9 That's, more or less, comparable to what you're seeing in  
10 spent nuclear fuel. So, from the standpoint of meeting the  
11 spent nuclear fuel standard, it's roughly comparable. Trying  
12 to extract it from the high-level waste glass presents  
13 different kinds of processing problems than one would have  
14 for spent nuclear fuel. You can argue which would be more  
15 difficult. I guess, it depends on the types of facilities  
16 that you have or would have to build.

17           So, for the most part, I think we have satisfied  
18 the spent nuclear fuel standard. You know, the longer range  
19 question is, you know, we have created plutonium mine for the  
20 future generations, but that's the nature of our program.

21           One of the focus areas for our program is trying to  
22 provide necessary and sufficient data that would, number one,  
23 allow us to understand the mechanisms of degradation that  
24 this form will undergo in the repository environment, as well  
25 as providing some data for a variety of different repository

1 conditions, so that we can compare the behavior of this form  
2 with, say, glass and other forms that have been  
3 characterized.

4           We have a series of corrosion tests that have been  
5 set up at Argonne National Laboratory, at Pacific Northwest  
6 National Laboratory, as well as Livermore, and Savannah  
7 River. Both tests under static conditions, such as the PCT  
8 tests that Ms. Marra talked about, MCC-1 tests. We're also  
9 doing a variety of single path flow unsaturated tests at both  
10 Argonne and PNL and Livermore. Argonne is also doing vapor  
11 hydration tests and other tests to look at accelerated  
12 leaching conditions associated with these forms. A great  
13 deal of effort, especially at Argonne, is being spent to look  
14 at the nature of the secondary phases that are formed as the  
15 material degrades and the nature of these phases so we can  
16 understand how does the fissile material and the neutron  
17 absorbers, hafnium and gadolinium, partition into not only  
18 stay with the primary phases, pyrochlore and zirconolite, but  
19 also in the degradation phases how do they partition into  
20 those phases, hopefully, showing that indeed there will be  
21 enough neutron absorbers homogeneously distributed in any of  
22 the degradation phases that we have no concerns, whatsoever,  
23 about long-term criticality.

24           We are also developing basic thermodynamic data on  
25 hafnium and gadolinium so that we can input some of the

1 modeling work being done by the RW contractors. We're  
2 developing an analytical model to predict the degradation  
3 behavior long-range of the form itself. A lot of these tests  
4 that we're performing with the form are done with the high-  
5 level waste canister materials basically in the soup, so to  
6 speak, when we're testing that.

7           And then, finally, one of the issues that has been  
8 raised with the ceramic form has been the fact that the alpha  
9 damage that is caused over a period of approximately 1,000  
10 years will cause the ceramic to go from basically a  
11 crystalline form to a form that is more amorphous in  
12 structure; otherwise, becoming metamict. And so, we  
13 anticipate at this point with some of the existing data that  
14 we would probably see maybe an order of magnitude increase of  
15 the leach rate associated with the metamict form, but we're  
16 providing specific tests using PU-238 doping, as well as Ned  
17 Bibler is going to be doing some work with ion implantation  
18 to look at the effects of radiation. This is going to be  
19 like a four to five year program.

20           So, these are the things that we're going to be  
21 doing. In subsequent briefings to the TRB, what we'd like to  
22 do is some back and be more results specific. This is just  
23 to provide you with sort of an overview of what we're trying  
24 to accomplish in the program and I think, Bill, you took the  
25 schedule, but we really have from the repository analysis

1 standpoint some key milestones. Coming up in July of '99 is  
2 basically providing RW with input for the licensing  
3 application, and we will be updating that annually over the  
4 next several years as we learn more and more about the  
5 behavior of the form under simulated repository conditions  
6 and analyses of those conditions.

7 BULLEN: Thank you, Tom. Questions from the Panel?

8 DI BELLA: Just in the last thing that you said,  
9 "providing RW with information for the license application",  
10 I am almost positive their current program plan does not call  
11 for disposal of this material in the repository; not to say  
12 that it couldn't be, but I just don't think it's within their  
13 plan. Has their plan changed?

14 DANKER: Yes, it is in the process of being changed.

15 DI BELLA: Okay.

16 DANKER: Yeah, let me try. It's a timely question. As  
17 we speak--and I think Jim Brazee was alluding to it--there  
18 are changes to their technical baseline ongoing. So, it's  
19 active to get plutonium on their radar screen. I think, if  
20 I'm not mistaken, there's a meeting this Friday of their  
21 change control board to formalize the change to their  
22 technical baseline.

23 BULLEN: Any other questions from the Panel?

24 DI BELLA: Now, to a technical question. What size  
25 range do you have to mill the particles to to get the

1 centering at a reasonable time period?

2       GOULD: Basically, it's reactive centering and we've  
3 found that if we get the particle size down to 10 microns  
4 that we get pretty much a full dissolution of the plutonium  
5 oxide into the relevant mineral phases. We notice that if  
6 we're above 20 microns that we tend to have some small ions  
7 of plutonium oxide that exist in the centered form.

8       DI BELLA: Is there experience with milling plutonium  
9 that fine within glove boxes and what it does to your dust--

10       GOULD: Yes. As a matter of fact, that's typical of the  
11 mix oxide fuel fabrication business. As a matter of fact,  
12 our baseline flow sheet right now and choice of equipment  
13 mirrors quite a bit what BNFL is doing in their new  
14 Sellafield plant.

15       BULLEN: One last question. You mentioned the 10-fold  
16 degradation due to radiation damage and I guess that's in  
17 comparison when you made the selection of the ceramic waste  
18 form. Is similar radiation damage expected for a  
19 borosilicate glass waste form?

20       GOULD: I think actually borosilicate glass, we wouldn't  
21 expect to lose as--

22       BULLEN: Right. It's amorphous to begin with.

23       GOULD: Yeah, it's amorphous to begin with and I think  
24 Ned can--if Ned Bibler is still here, he can speak more  
25 authoritatively on this subject, but I don't think we would

1 anticipate that the leach rate to change as much for the  
2 glass waste form. The one we were looking at was a  
3 lanthanide borosilicate glass, a very high melting  
4 temperature glass at about 1500 degrees Centigrade, in order  
5 to get significant quantities of plutonium into the glass  
6 matrix. Our leach tests on actual samples containing  
7 plutonium showed both static and flow-through tests. If you  
8 looked at just sort of a range of the results, it showed the  
9 ceramic being a factor of 100 to  $10^4$  more durable than the  
10 glass.

11 BULLEN: So, even a 10-fold decrease in leachability--

12 GOULD: It still should be at least comparable with any  
13 glass form, but probably better.

14 BULLEN: Thank you.

15 Well, we come to the cleanup position again and I  
16 didn't realize that both times it was going to be Dave  
17 Haught, but Dave is going to close out the afternoon session  
18 as he did the morning session speaking about the disposal of  
19 vitrified high-level waste and immobilized weapons-grade  
20 plutonium from the DOE perspective or the Yucca Mountain Site  
21 Characterization Office perspective. Dave is still an  
22 engineer at the Yucca Mountain Site Characterization unless  
23 he got promoted since noon. He's responsible for the  
24 oversight and development of waste package design, materials  
25 testing, and modeling programs.

1           HAUGHT: I'd like to start--and this feels like a  
2 cardinal sin of presenting something--with an apology. There  
3 is a hard copy of this presentation. It is making its way  
4 here. It has not arrived. It was Fed-Ex'd yesterday and I  
5 had hopes of it actually being here by noon today, but that  
6 didn't happen.

7           I'm going to talk to you about the disposal of  
8 high-level waste and immobilized plutonium. Actually, as I  
9 get into this, I'm going to talk more about the immobilized  
10 plutonium than the high-level waste because it's a more  
11 interesting topic. When I show you some of the performance  
12 assessment curves that we have, you'll see why there is very  
13 little issue with vitrified high-level waste, but we do have  
14 to address criticality in the case of the immobilized  
15 plutonium because there is some fissile content.

16           The waste package design for both the vitrified  
17 high-level waste and immobilized plutonium is either a four  
18 or a five pack with the four high-level waste canisters per  
19 waste package. The containment barriers here are current  
20 design. Reference design is 10 centimeters of carbon steel  
21 and two centimeters of alloy 625. I note that we are  
22 evaluating C-22 as a replacement for the 625 and that is in  
23 review as we speak. The M&O may have actually worked that  
24 through by Christmas time, but as of today, it's still 625.

25           The only difference when you get to the immobilized

1 plutonium is you would swap out some of the vitrified high-  
2 level waste canisters with one or two plutonium containing  
3 canisters per package. Now, the basis of what I am telling  
4 you today is based on the old formulations for the ceramic.  
5 The new formulation is kind of moving in a safer direction  
6 and so these numbers here may go up somewhat, but we have not  
7 done that analysis yet.

8           BULLEN: Just a quick question, Dave. Those one or two  
9 are limited by criticality issues within the waste package  
10 itself?

11           HAUGHT: That's correct.

12                   And, you have seen the five pack. The five pack  
13 without co-disposal would just be a five pack without the  
14 canister down the center and the four pack looks like this.

15                   Performance assessment, I'm going to show you one  
16 curve and that is the sensitivity analysis for DOE and SNF  
17 which includes allowing for vitrified high-level waste and  
18 our current basis for vitrified high-level waste is DWPF  
19 canisters. The one thing that I would like to note and it  
20 has gotten a fair amount of attention recently is that the  
21 dose history curves I'm going to show you do not consider the  
22 colloidal transport of plutonium. It is planned for TSPA-VA,  
23 though.

24                   Our findings are that vitrified high-level waste  
25 does not significantly impact the dose at the accessible

1 environment. In comparison to commercial spent fuel, it's  
2 about two orders of magnitude less. The sensitivity analyses  
3 that we've done on immobilized plutonium show that its  
4 performance is similar to vitrified high-level waste. So, at  
5 least, our current understanding--again based on the old  
6 formulation--is that, you know, the vitrified high-level  
7 waste and immobilized plutonium curves are going to look just  
8 about the same.

9           This is a curve. The red line here is the--well,  
10 as you can see, it's 8,745 metric tons of heavy metal of  
11 high-level waste and the peaks are tracking about two orders  
12 of magnitude less than commercial spent fuel and is included  
13 in the black line. Actually, I misspoke; that's the base  
14 case, not just commercial spent fuel. But, given the very  
15 small contribution, it pretty much is just commercial spent  
16 fuel.

17           Now, what is different about the immobilized  
18 plutonium is that we do have to look at criticality and  
19 here's some of the bases that--I've gone through the analysis  
20 process, the Phased Analysis Process before. So, I'm not  
21 going to discuss that, but I will talk about here are the  
22 basic assumptions that we've brought into the thing. That is  
23 of the plutonium, it is 93 percent plutonium-239. The  
24 plutonium is immobilized in glass. We have done the  
25 immobilized plutonium in glass for intact degraded and

1 external configurations. We have done the old formulation of  
2 the ceramic for intact and internal degraded configurations.

3           Now, to try to relate what we've done to what  
4 you've heard from the folks from MD's program is I've got  
5 this comparison here of the old versus new. Some of this,  
6 you have already heard from both Tom and Bill. But, as you  
7 can see, the amount of plutonium per canister is going down,  
8 and more importantly, the amount of hafnium--well, as  
9 important--the amount of hafnium is going up. So, we have a  
10 better waste form.

11           Just like in the case of the aluminum clad fuels, I  
12 have an I chart for you on the scenario generation of how we  
13 arrive at the configurations. You'll see that's a Step 1.  
14 There is a Step 2 which gets into a lot more detail into some  
15 of the chemical processes that are going up in this round.  
16 But, just for simplicity sake of saying how we get to the  
17 configurations, I'm just going to show this one.

18           Now, here's how we kind of start as the degradation  
19 process begins. This is showing a four pack with two  
20 plutonium containing canisters in it and, as you can see,  
21 you've got some water, you've got some clays. The canisters  
22 are beginning to degrade; this particular one is breached in  
23 some of the cans. The plutonium containing cans are  
24 degrading.

25           As you go a bit further into it, we have a couple

1 different configurations. The one on the left here is that  
2 the glass and the plutonium are kind of degrading at  
3 approximately the same rate. On the right side is we have  
4 what would be a fairly severe configuration where we have  
5 the--the glass is degrading out of the plutonium containing  
6 canister and leaving the cans behind. In addition, you have  
7 in this case these canisters on the bottom are remaining  
8 intact enough such that the canisters on the top are  
9 supported above any of the clays or any of the neutron  
10 absorbers that might have been leached out. So, that is kind  
11 of a most severe configuration. We believe that it's fairly  
12 unlikely because it requires a period of a low pH in order to  
13 get the separation of the plutonium from the absorber. But,  
14 if you have that situation, you probably would not have these  
15 on the bottom still intact, but we have considered it.

16           And, just another look, this was inside a canister,  
17 the different configurations you could have in there. You  
18 could have a case where the glass is kind of turned into kind  
19 of a clay-like mixture and you still have some intact cans  
20 and then the other case of you have kind of a soup, if you  
21 will.

22           Here are some findings. Again, this is on the old  
23 formulation. An internal criticality can be prevented with a  
24 mass limit of 50 kilograms of plutonium-239 per waste  
25 package. We did make a recommendation that hafnium would

1 provide some additional defense and depth, and as you've  
2 seen, the amount of hafnium has gone up in the hockey pucks.  
3 And then, in the case of external configurations, there's a  
4 couple of ways that we can have the re-disposition of the  
5 fissile material. And, as we have run the codes, this is a  
6 configuration, a worst case configuration, that we believe  
7 can happen and that is that we have five kilograms of fissile  
8 material within a 15 cubic meter area under the footprint of  
9 the waste package. The K effective of that is below  
10 critical. In fact, I think, the highest we were able to get  
11 it is .95.

12           In the far field, it's well-known that zeolites are  
13 abundant in Yucca Mountain. We believe that the maximum  
14 uranium--and this is then after the plutonium is decayed into  
15 uranium--absorption is about 0.17 percent by weight in the  
16 zeolite and that's insufficient to accumulate a critical  
17 mass. Another mechanism would be a reducing environment and  
18 we just have not seen in the Yucca Mountain environment any  
19 more than trace quantities of this. So, we believe that  
20 that's a low probability of the re-precipitation of uranium  
21 by any reducing environment.

22           Now, we have done a consequence analysis for the  
23 external configurations. Now, in the case that I gave you of  
24 what I thought our worst case configuration was, it wasn't  
25 critical. So, the consequence is nothing. So, what we have

1 done is we have hypothetically put together a case where the  
2 plutonium in a single waste package would go critical and  
3 that's if we could get six kilograms of plutonium-239 to  
4 condense into a one cubic meter block and the result would be  
5 we'd have 500 watts of power generated for approximately  
6 4,000 years and we'd have about a 14 percent increase in the  
7 radioactivity of that package's plutonium.

8           Our current statuses, we are planning to update our  
9 analysis of the intact and internal degraded configurations  
10 using the new formulations of the ceramic during the course  
11 of FY-98. And, we will finish the analysis of the external  
12 configurations and the probabilities and consequences again  
13 in FY-99. That may sound like I've said it before and it has  
14 and we have deferred all of the external configuration  
15 analyses to later on because it is a cumulative effect, and  
16 it kind of doesn't make sense to really do it. It's not as  
17 efficient to do them on a case-by-case basis.

18           In summary, we believe that the impact of total  
19 system performance is small for most vitrified high-level  
20 waste and immobilized plutonium. Internal configurations of  
21 immobilized plutonium can be maintained at some critical  
22 levels. The disposal of immobilized plutonium appears  
23 workable from a technical point of view. I would like to  
24 reiterate what Bill said, although the plutonium is not  
25 currently in our baseline, it is in our--the BCP has made it

1 through?

2 SPEAKER: Right.

3 HAUGHT: Okay. It's a recent development. That's  
4 really all I have.

5 BULLEN: Thank you, Dave.

6 Questions from the Panel?

7 DI BELLA: You said that five kilograms of plutonium-239  
8 in a 15 meter volume under the waste package would be the  
9 worst case and that wouldn't be critical. Can you explain  
10 how you know that or how you think that is the worst case?

11 HAUGHT: Can I explain it? No, I'd like to call on  
12 Peter to come here and help me with that.

13 GOTTLIEB: Peter Gottlieb, M&O. The analysis was using  
14 the same geochemistry code that was mentioned for the  
15 internal criticality this morning. It was used in a  
16 different mode, but it was interpreted so that we could get  
17 the maximum distribution of deposits of plutonium and uranium  
18 in fractures immediately beneath the waste package which is  
19 the area of zone volume where they would be the most  
20 concentrated. Now, there are other possibilities for  
21 accumulating concentrations in reducing zones, organic  
22 materials, and so forth which were treated differently and  
23 which were not found to be critical either. But, in this  
24 instance that Dave is alluding to, if we look at what is  
25 predicted to precipitate or absorb in the fractures, it is

1 much too small to have anywhere near criticality. If we look  
2 at the maximum volume we could stuff into fractures that are  
3 there in the manner of some other analysis that's been done,  
4 then 15 cubic meters in that footprint would be close to  
5 critical. It would not be .95; it would still be well under  
6 that. But, in order to do a consequence analysis, we have to  
7 have a critical mass. And, so what we did, we artificially  
8 compressed that into one cubic meter which made the K  
9 effective up to one and so we could do a consequence analysis  
10 which led to the increased radionuclide that Dave quoted.  
11 But, we have a conservatism. We have an extreme conservatism  
12 on top of another extreme conservatism in order to get to a  
13 point where we can do a consequence analysis. So, it's only  
14 for illustrative purposes. It is not to be considered in any  
15 way a criteria for accepting the waste.

16 HAUGHT: I'd like to make sure that I can clarify what  
17 you've said, Peter, in terms that might answer Carl's  
18 question. I believe from what Peter just said that the  
19 answer of how we believe that the five kilograms deposited in  
20 a 15 cubic meter area or volume below the waste package is  
21 that we--in running the EQ 3/6 codes in the near field area,  
22 that those were the results that we got. Is that--did I  
23 characterize what you say correctly, Peter?

24 GOTTLIEB: No.

25 HAUGHT: No. I'm glad I asked then.

1           GOTTLIEB:  When we ran the EQ 3/6 codes--you see, what  
2 we find when we tried to--some people take the fractures and  
3 say you can stuff the fractures with plutonium and uranium  
4 and get criticality.  The whole purpose of running the EQ 3/6  
5 code is to say, all right, what else is going to happen?  You  
6 don't have pure water with uranium and plutonium in it.  You  
7 have lots of other stuff and that's going to be competing for  
8 space in those fractures.  And, when we do that analysis, it  
9 comes out far below anything approaching criticality.  So, in  
10 order to get close to criticality, we have to go a factor of  
11 10 or so on top of that in order to get anything close to  
12 criticality.  That still isn't critical over the 15 cubic  
13 meters which is sort of where it would all be coming out.  
14 So, if we compress into one cubic meter, then it is K  
15 effective equal to one and we can then conduct an analysis of  
16 the evolution or the consequences of criticality.

17                   Now, is that your question, Dave?

18           DI BELLA:  It was my question and this Carl.  Thank you  
19 very much.  I only wanted to know not about your criticality  
20 calculations, but how you know that five kilograms, not 5.1  
21 is the most--or 10 or whatever number it's going to be, is  
22 the most plutonium-239 that is going to come to rest in a 15  
23 cubic meter space under the waste packages.  Actually, I  
24 would think it would be much less than that and you could  
25 come forth with a plausible explanation.

1           GOTTLIEB: Well, the five kilograms is approximately  
2 what we get out of the EQ 3/6 calculation. That's straight  
3 from calculation. That's an approximate figure. It isn't  
4 that sharp between five and 5.1.

5           BULLEN: Another question, Carl?

6           DI BELLA: Yeah. Then, I have a question about the  
7 criticality calculations you did where you said you had six  
8 kilograms of Pu 239 and one cubic meter volume would be  
9 critical and chug along at 500 watts for 4,000 years. What  
10 are the basic parameters that go into that calculation? It  
11 seems to me it's going to be dependent on seepage flows.  
12 I'm looking for that number particularly.

13          GOTTLIEB: Can I do this again, huh? The inflow of  
14 dripping water into the waste package for that case would be  
15 approximately 5mm/yr. That's the flow rate and then that's  
16 over the area of the waste package. So, you can multiply  
17 that to get the cubic meters of about less than half a cubic  
18 meter.

19          DI BELLA: What if it were 10, as someone put on a slide  
20 today, maybe you in this morning's presentation? Or what if  
21 it were 20 or 60 even, as some of the people from the expert  
22 elicitation have offered as a possibility particularly with  
23 climate change? What do you think the consequences might be  
24 in that circumstance?

25          GOTTLIEB: Well, with a higher infiltration rate, it's

1 possible that you would have a higher--with higher drip rate,  
2 it's possible you would have a higher power level  
3 approximately linearly proportional because one of the  
4 limiting factors in the evolution of criticality is the heat  
5 dissipation and the heat will be removing water. And, if you  
6 dissipate too much heat, you remove too much water and you go  
7 subcritical. So, presumably, if you could replenish the  
8 water at a faster rate, you could sustain a higher power  
9 level.

10 BULLEN: I just have one quick question about the stack  
11 of 20 hockey pucks with the poisons in them. I'm assuming  
12 that that's a subcritical assembly if I immersed it in water.  
13 Is that not correct?

14 HAUGHT: Yes.

15 BULLEN: Okay. Now, you mentioned radiation effects and  
16 you're looking at fundamental thermodynamics of hafnium in  
17 the materials. What fraction of the hafnium do I have to  
18 take out before I have to worry or is there a very large  
19 margin and it sits in never mind?

20 HAUGHT: There is also gadolinium.

21 BULLEN: Gadolinium and hafnium. So, we have two BPs or  
22 two burnable poisons in there?

23 HAUGHT: Yeah.

24 BULLEN: And so, what fraction of those do I have to  
25 remove? What kind of margin do I have if thermodynamics

1 isn't necessarily my friend in the radiation damage  
2 environment?

3       HAUGHT: I don't know what the fraction of the  
4 gadolinium and hafnium have to stay in. I can answer the  
5 question a little differently. I believe we're considering  
6 that the hafnium is going to stay and that the gadolinium  
7 excepting a couple of low pH scenarios will migrate with the  
8 plutonium. In fact, the low pH scenario that would allow  
9 that to happen, we haven't quite convinced ourselves that it  
10 actually can. That would be the forming of chromic acid due  
11 to the corrosion of the canister and perhaps the can itself.  
12 So, Peter, do we have any real hard numbers on that or is  
13 that the best answer we can give?

14       GOTTLIEB: I'm not prepared to give a specific number,  
15 but it's, at least, 90 percent. We could lose, at least, 90  
16 percent and still be subcritical.

17       BULLEN: Thank you.

18               Any other questions from the Panel?

19       DI BELLA: A quick one, I hope. The weapons-grade  
20 plutonium, some of it anyway, has a small percentage of  
21 gallium in it. That, I believe, causes the MOX people some  
22 problems, an extra process step or two. Does it cause any  
23 problems in this ceramic process, particularly; the gallium?

24       GOULD: The answer at the levels of the gallium that  
25 would be coming in with the weapon-grade plutonium, the

1 answer is no.

2 BULLEN: Any other questions from the Panel?

3 (No response.)

4 BULLEN: Questions from the audience?

5 (No response.)

6 BULLEN: Thank you very much, Dave.

7 HAUGHT: Thanks.

8 BULLEN: I notice by the agenda that we have until 5:00  
9 o'clock. Let me make a couple of comments first. There was  
10 no one who signed up for the public comment period, and I  
11 would like to make one last call prior to closing remarks for  
12 anyone who would like to make public comment. If so, please,  
13 step to the microphone at this time and be recognized?

14 (No response.)

15 BULLEN: Seeing none, we move on to the last item on the  
16 agenda which is closing comments or closing remarks by Dan  
17 Bullen. I see that we're not supposed to be out of there  
18 until 5:00. So, that means I have 45 minutes to speak. Is  
19 that not correct?

20 (No response.)

21 BULLEN: No, I'm sure my classes would argue that I  
22 could take a two minute talk and make it 45 minutes, but  
23 today I would just like to express the appreciation of the  
24 Board and specifically the Repository Panel to all the  
25 speakers and to our DOE and Savannah River Site

1 representatives who have gone to great lengths to organize  
2 both today and tomorrow's tour.

3           With that, I would like to call these proceedings  
4 closed. Thank you very much.

5           (Whereupon, at 4:15 p.m., the meeting was  
6 adjourned.)

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